

U.S. Non-Provisional Patent Application entitled
NON-NUCLEOSIDE REVERSE TRANSCRIPTASE INHIBITORS

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NON-NUCLEOSIDE REVERSE TRANSCRIPTASE INHIBITORS

CROSS REFERENCE TO PRIOR APPLICATION

This application claims benefit under Title 35 U.S.C. 119(e) of U.S. Provisional Application No.
5 60/457,144, filed March 24, 2003, which is hereby incorporated by reference in its entirety

FIELD OF THE INVENTION

The invention relates to the field of antiviral therapy and, in particular, to non-nucleoside reverse
10 transcriptase inhibitors for treating Human Immunodeficiency Virus (HIV) mediated diseases. The
invention provides novel pyridazinone compounds, pharmaceutical compositions comprising these
compounds, methods for treatment or prophylaxis of HIV mediated diseases employing said compounds
in monotherapy or in combination therapy.

15 BACKGROUND OF THE INVENTION

The human immunodeficiency virus HIV is the causative agent of acquired immunodeficiency
syndrome (AIDS), a disease characterized by the destruction of the immune system, particularly of the
CD4⁺ T-cell, with attendant susceptibility to opportunistic infections. HIV infection is also associated
20 with a precursor AIDS-related complex (ARC), a syndrome characterized by symptoms such as
persistent generalized lymphadenopathy, fever and weight loss.

In common with other retroviruses, the HIV genome encodes protein precursors known as gag and gag-
pol which are processed by the viral protease to afford the protease, reverse transcriptase (RT),
25 endonuclease/integrase and mature structural proteins of the virus core. Interruption of this processing
prevents the production of normally infectious virus. Considerable efforts have been directed towards
the control of HIV by inhibition of virally encoded enzymes.

Currently available chemotherapy targets two crucial viral enzymes: HIV protease and HIV reverse
30 transcriptase (J. S. G. Montaner et al. Antiretroviral therapy: 'the state of the art', Biomed &
Pharmacother. 1999 53:63-72; R. W. Shafer and D. A. Vuitton, *Highly active retroviral therapy*
(HAART) for the treatment of infection with human immunodeficiency virus type 1, Biomed. &
Pharmacother. 1999 53:73-86; E. De Clercq, *New Developments in Anti-HIV Chemotherap. Curr. Med.*
Chem. 2001 8:1543-1572). Two general classes of RTI inhibitors have been identified: nucleoside
35 reverse transcriptase inhibitors (NRTI) and non-nucleoside reverse transcriptase inhibitors (NNRTI).

NRTIs typically are 2',3'-dideoxynucleoside (ddN) analogs which must be phosphorylated prior to interacting with viral RT. The corresponding triphosphates function as competitive inhibitors or alternative substrates for viral RT. After incorporation into nucleic acids the nucleoside analogs terminate the chain elongation process. HIV reverse transcriptase has DNA editing capabilities which enable resistant strains to overcome the blockade by cleaving the nucleoside analog and continuing the elongation. Currently clinically used NRTIs include zidovudine (AZT), didanosine (ddI), zalcitabine (ddC), stavudine (d4T), lamivudine (3TC) and tenofovir (PMPA).

NNRTIs were first discovered in 1989. NNRTI are allosteric inhibitors which bind reversibly at a nonsubstrate-binding site on the HIV reverse transcriptase thereby altering the shape of the active site or blocking polymerase activity. (R. W. Buckheit, Jr., *Non-nucleoside reverse transcriptase inhibitors: perspectives for novel therapeutic compounds and strategies for treatment of HIV infection*, *Expert Opin. Investig. Drugs* **2001** 10(8)1423-1442; E. De Clercq *The role of non0-nuceloside reverse transcriptase inhibitors (NNRTIs) in the therapy of HIV-1 infection*, *Antiviral Res.* **1998** 38:153-179; G. Moyle, *The Emerging Roles of Non-Nucleoside Reverse Transcriptase Inhibitors in Antiviral Therapy*, *Drugs* **2001** 61(1):19-26) Although over thirty structural classes of NNRTIs have been identified in the laboratory, only three compounds have been approved for HIV therapy: efavirenz, nevirapine and delavirdine. Although initially viewed as a promising class of compounds, *in vitro* and *in vivo* studies quickly revealed the NNRTIs presented a low barrier to the emergence of drug resistant HIV strains and class-specific toxicity. Drug resistance frequently develops with only a single point mutation in the RT.

While combination therapy with NRTIs, PIs and NNRTIs has, in many cases, dramatically lowered viral loads and slowed disease progression, significant therapeutic problems remain. The cocktails are not effective in all patients, potentially severe adverse reactions often occur and the rapidly reproducing HIV virus has proven adroit at creating mutant drug-resistant variants of wild type protease and reverse transcriptase.

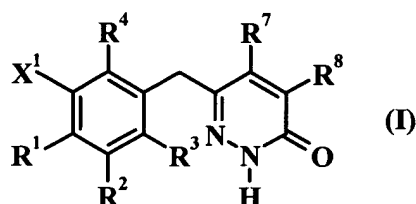
There remains a need for safer drugs with activity against wild type and commonly occurring resistant strains of HIV.

Benzyl-pyridazinone compounds have been extensively investigated as thyroxine analogs which can decrease plasma cholesterol without stimulating cardiac activity (A. H. Underwood *et al.* *A thyromimetic that decreases plasma cholesterol without increasing cardiovascular activity* *Nature* **1986** 324(6096):425-429; P. D. Leeson *et al.* *Selective thyromimetics. Cardiac-sparing thyroid hormone*

analogues containing 3'-arylmethyl substituents *J. Med Chem* **1989** 32(2):320-326; P. D. Leeson *et al.* EP 0188351). WO9624343 (D. J. Dunnington) discloses oxo-pyridazinylmethyl substituted tyrosines are selective antagonists for the haematopoietic phosphatase SH2 domain which may render them useful to increase erythropoiesis and haematopoiesis. WO 9702023 (D. J. Dunnington) and WO9702024 (D. J. Dunnington) further disclose these compounds are specific inhibitor of the human Stat 6 SH2 domain and may be useful for treating asthma, allergic rhinitis and anemia. WO2001085670 (H. Shiohara *et al.*) discloses related malonamide derivatives useful for treating circulatory diseases. EP 810218 (D. A. Allen *et al.*) discloses benzoyl substituted benzyl-pyridazinone compounds which are cyclooxygenase inhibitors and potential antiinflammatory or analgesic compounds. None of the references teaches therapy for HIV infections or inhibition of HIV reverse transcriptase.

SUMMARY OF THE INVENTION

The present invention relates to a compounds according to formula I, methods for treating diseases mediated by human immunodeficiency virus by administration of a compound according to formula I and pharmaceutical compositions for treating diseases mediated by human immunodeficiency virus containing a compound according to formula I,



X¹ is selected from the group consisting of R⁵O, R⁵S(O)_n, R⁵CH₂, R⁵CH₂O, R⁵CH₂S(O)_n, R⁵OCH₂, R⁵S(O)_nCH₂, NR⁵R⁶ and C(=O)R⁵;

R¹ and R² are

- (i) each independently selected from the group consisting of hydrogen, C₁₋₆ alkyl, C₁₋₆ haloalkyl, C₃₋₈ cycloalkyl, C₁₋₆ alkoxy, C₁₋₆ alkylthio, C₁₋₆ alkylsulfinyl, C₁₋₆ sulfonyl, C₁₋₆ haloalkoxy, C₁₋₆ haloalkylthio, halogen, amino, alkylamino, dialkylamino, aminoacyl, nitro and cyano; or,
- (ii) taken together are -CH=CH-CH=CH-, or
- (iii) taken together along with the carbons to which they are attached form a five- or six-membered heteroaromatic or heterocyclic ring with a one or two heteroatoms independently selected from the group consisting of O, S and NH;

R³ is selected from the group consisting of hydrogen, C₁₋₆ alkyl, C₁₋₆ haloalkyl, C₃₋₈ cycloalkyl, C₁₋₆ alkylthio, C₁₋₆ haloalkylthio, halogen, amino, alkylamino, dialkylamino, aminoacyl, nitro and cyano;

R⁴ is selected from the group consisting of hydrogen, C₁₋₆ alkyl, C₁₋₆ haloalkyl, C₃₋₈ cycloalkyl, C₁₋₆ alkoxy, C₁₋₆ alkylthio, C₁₋₆ haloalkoxy, C₁₋₆ haloalkylthio, halogen, amino, alkylamino, dialkylamino, aminoacyl, nitro and cyan

R⁵ is selected from the group consisting of alkyl, haloalkyl, cycloalkyl, phenyl, naphthyl, pyridinyl, pyridine N-oxide, indole, indole N-oxide, quinoline, quinoline N-oxide, pyrimidinyl, pyrazinyl and pyrrolyl; wherein,

said alkyl and said cycloalkyl are optionally substituted with one or two substituents independently selected from the group consisting of alkyl, hydroxy, alkoxy, thiol, alkylthio, halogen, amino, alkylamino, dialkylamino, aminoalkyl, alkylaminoalkyl, and dialkylamino; and,

said phenyl, said naphthyl, said pyridinyl, said pyridine N-oxide, said indole, said indole N-oxide, said quinoline, said quinoline N-oxide, said pyrimidinyl, said pyrazinyl and said pyrrolyl groups are optionally substituted with one to three substituents independently selected from the group consisting of C₁₋₆ alkyl, C₁₋₆ alkenyl, C₁₋₆ haloalkyl, C₃₋₈ cycloalkyl, C₁₋₆ alkoxy, C₁₋₆ alkylthio, C₁₋₆ alkylsulfinyl, C₁₋₆ sulfonyl, C₁₋₆ haloalkoxy, C₁₋₆ haloalkylthio, hydroxy, halogen, amino, C₁₋₆ alkylamino, C₁₋₆ dialkylamino, aminoacyl, acyl, C₁₋₆ alkoxycarbonyl, carbamoyl, C₁₋₆ N-alkylcarbamoyl, C₁₋₆ N,N-dialkylcarbamoyl, nitro and cyano;

R⁶ is hydrogen, C₁₋₆ alkyl, or acyl;

R⁷ and R⁸ (i) taken independently are selected from the group consisting of hydrogen amino, C₁₋₆ alkylamino, C₁₋₆ dialkylamino, amino-C₁₋₃ alkyl, alkylamino-C₁₋₃ alkyl, C₁₋₃ dialkylamino-C₁₋₃ alkyl or C₁₋₆ alkyl optionally substituted with one or two substituents independently selected from the group consisting of hydroxy, alkoxy, thiol, alkylthio, C₁₋₆ alkylsulfinyl, C₁₋₆ sulfonyl, halogen; or, (ii) R⁷ and R⁸ taken together are -(CH₂)₄-;

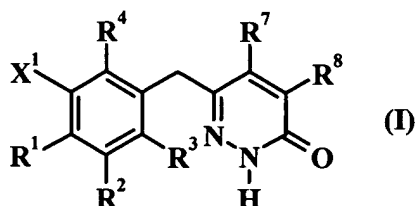
n is an integer from 0 to 2; and,

hydrates, solvates, clathrates and acid addition salts thereof.

The invention also relates to a process for preparing a compound according to formula I wherein X¹ is OR⁵ or SR⁵, R⁵ is an optionally substituted aryl, alkyl or aralkyl moiety and R¹-R⁴, R⁷ and R⁸ are as defined hereinabove

DETAILED DESCRIPTION OF THE INVENTION

In one embodiment of the invention there is provided a compound according to formula I,



wherein X¹ is R⁵O, R⁵S(O)_n, R⁵CH₂, R⁵CH₂O, R⁵CH₂S(O)_n, R⁵OCH₂, R⁵S(O)_nCH₂ or NR⁵R⁶; and R¹,
5 R², R³, R⁴, R⁵, R⁶, R⁷ and R⁸ are as defined hereinabove, and hydrates, solvates, clathrates and acid
addition salts thereof.

In another embodiment of the present invention there is provided a compound according to formula I
wherein X¹ is R⁵O, R⁵S(O)_n, R⁵CH₂, R⁵CH₂O, R⁵CH₂S(O)_n, R⁵OCH₂, R⁵S(O)_nCH₂ or NR⁵R⁶; R⁵ is
10 selected from the group consisting of alkyl, haloalkyl, cycloalkyl, phenyl, naphthyl, pyridinyl, pyridine
N-oxide, indole, indole N-oxide, quinoline, quinoline N-oxide pyrimidinyl, pyrazinyl and pyrrolyl;
wherein, said alkyl and said cycloalkyl are optionally substituted with one or two substituents
independently selected from the group consisting of alkyl, hydroxy, alkoxy, thiol, alkylthio, halogen,
amino, alkylamino, dialkylamino, aminoalkyl, alkylaminoalkyl, and dialkylamino; and, said phenyl,
15 said naphthyl, said pyridinyl, said pyrimidinyl, said pyrazinyl and said pyrrolyl groups are optionally
substituted with one to three substituents independently selected from the group consisting of C₁₋₆ alkyl,
C₁₋₆ alkenyl, C₁₋₆ haloalkyl, C₃₋₈ cycloalkyl, C₁₋₆ alkoxy, C₁₋₆ alkylthio, C₁₋₆ alkylsulfinyl, C₁₋₆ sulfonyl,
C₁₋₆ haloalkoxy, C₁₋₆ haloalkylthio, halogen, alkylamino, dialkylamino, acylamino, acyl and cyano; and,
R¹, R², R³, R⁴, R⁶, R⁷ and R⁸ are as defined hereinabove, and hydrates, solvates, clathrates and acid
20 addition salts thereof.

In another embodiment of the present invention there is provided a compound according to formula I
wherein X¹ is OR⁵ or SR⁵; R³ is hydrogen or fluoro; R⁴ is hydrogen, chloro, fluoro or methyl; R⁵ is
phenyl optionally substituted with one to three substituents independently selected from the group
25 consisting of C₁₋₆ alkyl, C₁₋₆ alkenyl, C₁₋₆ haloalkyl, C₃₋₈ cycloalkyl, C₁₋₆ alkoxy, C₁₋₆ alkylthio, C₁₋₆
alkylsulfinyl, C₁₋₆ sulfonyl, C₁₋₆ haloalkoxy, C₁₋₆ haloalkylthio, halogen, alkylamino, dialkylamino,
acylamino, cyano, and acyl; and, R⁷ and R⁸ are independently selected from the group consisting of
hydrogen, amino, alkylamino, dialkylamino, aminoalkyl, alkylaminoalkyl and dialkylaminoalkyl and
C₁₋₆ alkyl optionally substituted with hydroxy, alkoxy, thiol, alkylthio or halogen; and R¹ and R² are as
30 defined hereinabove; and hydrates, solvates, clathrates and acid addition salts thereof.

In another embodiment of the present invention there is provided a compound according to formula I wherein X^1 is OR^5 or SR^5 ; R^1 is methyl, ethyl, trifluoromethyl or halogen; R^3 is hydrogen or fluoro; R^4 is hydrogen, chloro, fluoro or methyl; R^5 is phenyl optionally substituted with one to three substituents independently selected from the group consisting of C_{1-6} alkyl, C_{1-6} alkenyl, C_{1-6} haloalkyl, C_{3-8} cycloalkyl, C_{1-6} alkoxy, C_{1-6} alkylthio, C_{1-6} alkylsulfinyl, C_{1-6} sulfonyl, C_{1-6} haloalkoxy, C_{1-6} haloalkylthio, halogen, alkylamino, dialkylamino, acylamino, cyano, and acyl; and, R^7 and R^8 are independently selected from the group consisting of hydrogen, amino, alkylamino, dialkylamino, aminoalkyl, alkylaminoalkyl and dialkylaminoalkyl and C_{1-6} alkyl optionally substituted with hydroxy, alkoxy, thiol, alkylthio or halogen; and R^2 is as defined hereinabove; and hydrates, solvates, clathrates and acid addition salts thereof.

In another embodiment of the present invention there is provided a compound according to formula I wherein X^1 is OR^5 or SR^5 ; R^1 is methyl, ethyl, trifluoromethyl or halogen; R^3 is hydrogen or fluoro; R^4 is hydrogen, chloro, fluoro or methyl; R^5 is phenyl substituted with one substituent selected from the group consisting of C_{1-6} alkyl, C_{1-6} alkenyl, C_{1-6} haloalkyl, C_{3-8} cycloalkyl, C_{1-6} alkoxy, C_{1-6} alkylthio, C_{1-6} alkylsulfinyl, C_{1-6} sulfonyl, C_{1-6} haloalkoxy, C_{1-6} haloalkylthio, halogen, alkylamino, dialkylamino, acylamino, cyano, and acyl; and, R^7 and R^8 are independently selected from the group consisting of hydrogen, amino, alkylamino, dialkylamino, aminoalkyl, alkylaminoalkyl and dialkylaminoalkyl and C_{1-6} alkyl optionally substituted with hydroxy, alkoxy, thiol, alkylthio or halogen; and R^2 is as defined hereinabove; and hydrates, solvates, clathrates and acid addition salts thereof.

In another embodiment of the present invention there is provided a compound according to formula I wherein X^1 is OR^5 or SR^5 ; R^1 is methyl, ethyl, trifluoromethyl or halogen; R^3 is hydrogen or fluoro; R^4 is hydrogen, chloro, fluoro or methyl; R^5 is 2,5-disubstituted phenyl which substituents are independently selected from the group consisting of C_{1-6} alkyl, C_{1-6} alkenyl, C_{1-6} haloalkyl, C_{3-8} cycloalkyl, C_{1-6} alkoxy, C_{1-6} alkylthio, C_{1-6} alkylsulfinyl, C_{1-6} sulfonyl, C_{1-6} haloalkoxy, C_{1-6} haloalkylthio, halogen, alkylamino, dialkylamino, acylamino, cyano, and acyl; and, R^7 and R^8 are independently selected from the group consisting of hydrogen, amino, alkylamino, dialkylamino, aminoalkyl, alkylaminoalkyl and dialkylaminoalkyl and C_{1-6} alkyl optionally substituted with hydroxy, alkoxy, thiol, alkylthio or halogen; and R^2 is as defined hereinabove; and hydrates, solvates, clathrates and acid addition salts thereof.

In another embodiment of the present invention there is provided a compound according to formula I wherein X^1 is OR^5 or SR^5 ; R^1 is methyl, ethyl, trifluoromethyl or halogen; R^3 is hydrogen or fluoro; R^4 is hydrogen, chloro, fluoro or methyl; R^5 is 3,5-disubstituted phenyl which substituents are independently selected from the group consisting of C_{1-6} alkyl, C_{1-6} alkenyl, C_{1-6} haloalkyl, C_{3-8} cycloalkyl, C_{1-6} alkoxy, C_{1-6} alkylthio, C_{1-6} alkylsulfinyl, C_{1-6} sulfonyl, C_{1-6} haloalkoxy, C_{1-6} haloalkylthio, halogen, alkylamino, dialkylamino, acylamino, cyano, and acyl; and, R^7 and R^8 are independently selected from the group consisting of hydrogen, amino, alkylamino, dialkylamino, aminoalkyl, alkylaminoalkyl and dialkylaminoalkyl and C_{1-6} alkyl optionally substituted with hydroxy, alkoxy, thiol, alkylthio or halogen; and R^2 is as defined hereinabove; and hydrates, solvates, clathrates and acid addition salts thereof.

In another embodiment of the present invention there is provided a compound according to formula I wherein X^1 is OR^5 or SR^5 ; R^1 is methyl, ethyl, trifluoromethyl or halogen; R^3 is hydrogen or fluoro; R^4 is hydrogen, chloro, fluoro or methyl; R^5 is 2,4-disubstituted phenyl which substituents are independently selected from the group consisting of C_{1-6} alkyl, C_{1-6} alkenyl, C_{1-6} haloalkyl, C_{3-8} cycloalkyl, C_{1-6} alkoxy, C_{1-6} alkylthio, C_{1-6} alkylsulfinyl, C_{1-6} sulfonyl, C_{1-6} haloalkoxy, C_{1-6} haloalkylthio, halogen, alkylamino, dialkylamino, acylamino, cyano, and acyl; and, R^7 and R^8 are independently selected from the group consisting of hydrogen, amino, alkylamino, dialkylamino, aminoalkyl, alkylaminoalkyl and dialkylaminoalkyl and C_{1-6} alkyl optionally substituted with hydroxy, alkoxy, thiol, alkylthio or halogen; and R^2 is as defined hereinabove; and hydrates, solvates, clathrates and acid addition salts thereof.

In another embodiment of the present invention there is provided a compound according to formula I wherein X^1 is OR^5 or SR^5 ; R^1 is methyl, ethyl, trifluoromethyl or halogen; R^3 is hydrogen or fluoro; R^4 is hydrogen, chloro, fluoro or methyl; R^5 is 2,6-disubstituted phenyl which substituents are independently selected from the group consisting of C_{1-6} alkyl, C_{1-6} alkenyl, C_{1-6} haloalkyl, C_{3-8} cycloalkyl, C_{1-6} alkoxy, C_{1-6} alkylthio, C_{1-6} alkylsulfinyl, C_{1-6} sulfonyl, C_{1-6} haloalkoxy, C_{1-6} haloalkylthio, halogen, alkylamino, dialkylamino, acylamino, cyano, and acyl; R^7 and R^8 are independently selected from the group consisting of hydrogen, amino, alkylamino, dialkylamino, aminoalkyl, alkylaminoalkyl and dialkylaminoalkyl and C_{1-6} alkyl optionally substituted with hydroxy, alkoxy, thiol, alkylthio or halogen; and R^2 is as defined hereinabove; and hydrates, solvates, clathrates and acid addition salts thereof.

In another embodiment of the present invention there is provided a compound according to formula I wherein X^1 is OR^5 or SR^5 ; R^1 and R^2 are independently selected from the group consisting of hydrogen, C_{1-6} alkyl, C_{1-6} alkenyl, C_{1-6} haloalkyl, C_{3-8} cycloalkyl, C_{1-6} alkoxy, C_{1-6} alkylthio, C_{1-6} alkylsulfinyl, C_{1-6} sulfonyl, C_{1-6} haloalkoxy, C_{1-6} haloalkylthio, halogen, amino, alkylamino, dialkylamino, aminoacyl, nitro and cyano; R^3 is hydrogen or fluoro; R^4 , R^5 , R^7 and R^8 are as defined hereinabove; and, hydrates, solvates, clathrates and acid addition salts thereof.

In another embodiment of the present invention there is provided a compound according to formula I wherein X^1 is OR^5 ; R^1 is methyl, ethyl, trifluoromethyl or halogen; R^2 and R^4 are independently hydrogen, chloro, fluoro, methyl or ethyl; R^3 is hydrogen or fluoro; R^5 is selected from the group consisting of alkyl, haloalkyl, cycloalkyl, phenyl, naphthyl, pyridinyl, pyridine N-oxide, indole, indole N-oxide, quinoline, quinoline N-oxide, pyrimidinyl, pyrazinyl and pyrrolyl; wherein, said alkyl and said cycloalkyl are optionally substituted with one or two substituents independently selected from the group consisting of alkyl, hydroxy, alkoxy, thiol, alkylthio, halogen, amino, alkylamino, dialkylamino, aminoalkyl, alkylaminoalkyl, and dialkylamino; and, said phenyl, said naphthyl, said pyridinyl, said pyrimidinyl, said pyrazinyl and said pyrrolyl groups are optionally substituted with one to three substituents independently selected from the group consisting of C_{1-6} alkyl, C_{1-6} alkenyl, C_{1-6} haloalkyl, C_{3-8} cycloalkyl, C_{1-6} alkoxy, C_{1-6} alkylthio, C_{1-6} alkylsulfinyl, C_{1-6} sulfonyl, C_{1-6} haloalkoxy, C_{1-6} haloalkylthio, halogen, alkylamino, dialkylamino, acylamino, acyl and cyano; R^7 is hydrogen, methyl or ethyl; R^8 is selected from the group consisting of hydrogen, amino, alkylamino, dialkylamino, aminoalkyl, alkylaminoalkyl, dialkylaminoalkyl and C_{1-6} alkyl optionally substituted with hydroxy, alkoxy, thiol, alkylthio or halogen; and, hydrates, solvates, clathrates and acid addition salts thereof.

In another embodiment of the present invention there is provided a compound according to formula I wherein X^1 is OR^5 ; R^1 is methyl, ethyl, trifluoromethyl or halogen; R^2 and R^4 are independently hydrogen, chloro, fluoro, methyl or ethyl; R^3 is hydrogen or fluoro; R^5 is phenyl substituted with one substituent selected from the group consisting of C_{1-6} alkyl, C_{1-6} alkenyl, C_{1-6} haloalkyl, C_{3-8} cycloalkyl, C_{1-6} alkoxy, C_{1-6} alkylthio, C_{1-6} alkylsulfinyl, C_{1-6} sulfonyl, C_{1-6} haloalkoxy, C_{1-6} haloalkylthio, halogen, alkylamino, dialkylamino, aminoacyl, cyano and acyl substituent; R^7 is hydrogen, methyl or ethyl; R^8 is selected from the group consisting of hydrogen, amino, alkylamino, dialkylamino, aminoalkyl, alkylaminoalkyl and dialkylaminoalkyl and C_{1-6} alkyl optionally substituted with hydroxy, alkoxy, thiol, alkylthio or halogen; and, hydrates, solvates, clathrates and acid addition salts thereof.

In another embodiment of the present invention there is provided a compound according to formula I wherein X^1 is OR^5 ; R^1 is methyl, ethyl, trifluoromethyl or halogen; R^2 and R^4 are independently hydrogen, chloro, fluoro, methyl or ethyl; R^3 is hydrogen or fluoro; R^5 is phenyl substituted with one substituent selected from the group consisting of halogen, cyano, C_{1-6} alkyl, C_{1-6} alkenyl, C_{3-8} cycloalkyl, C_{1-6} haloalkyl, C_{1-6} alkoxy, C_{1-6} alkylthio and C_{1-6} haloalkoxy; R^7 is hydrogen, methyl or ethyl; R^8 is selected from the group consisting of hydrogen, amino, alkylamino, dialkylamino, aminoalkyl, alkylaminoalkyl and dialkylaminoalkyl and C_{1-6} alkyl optionally substituted with hydroxy, alkoxy, thiol, alkylthio or halogen; and, hydrates, solvates, clathrates and acid addition salts thereof.

10 In another embodiment of the present invention there is provided a compound according to formula I wherein X^1 is OR^5 ; R^1 is halogen, methyl or ethyl; R^2 and R^4 are independently halogen, chloro, fluoro, methyl or ethyl; R^3 and R^7 are hydrogen; R^5 is a phenyl substituted with one substituent selected from the group consisting of halogen, cyano, C_{1-6} alkyl and C_{1-6} haloalkyl; and, R^8 is hydrogen, methyl or ethyl and hydrates, solvates, clathrates and acid addition salts thereof.

15 In another embodiment of the present invention there is provided a compound according to formula I wherein X^1 is OR^5 ; R^1 is methyl, ethyl, trifluoromethyl or halogen; R^2 and R^4 are independently hydrogen, chloro, fluoro, methyl or ethyl; R^3 is hydrogen or fluoro; R^5 is 2,5-disubstituted phenyl which substituents are independently selected from the group consisting of C_{1-6} alkyl, C_{1-6} alkenyl, C_{1-6} haloalkyl, C_{3-8} cycloalkyl, C_{1-6} alkoxy, C_{1-6} alkylthio, C_{1-6} alkylsulfinyl, C_{1-6} sulfonyl, C_{1-6} haloalkoxy, C_{1-6} haloalkylthio, halogen, alkylamino, dialkylamino, aminoacyl, cyano and acyl substituent; R^7 is hydrogen, methyl or ethyl; R^8 is selected from the group consisting of hydrogen, amino, alkylamino, dialkylamino, aminoalkyl, alkylaminoalkyl and dialkylaminoalkyl and C_{1-6} alkyl optionally substituted with hydroxy, alkoxy, thiol, alkylthio or halogen; and, hydrates, solvates, clathrates and acid addition salts thereof.

25 In another embodiment of the present invention there is provided a compound according to formula I wherein X^1 is OR^5 ; R^1 is methyl, ethyl, trifluoromethyl or halogen; R^2 and R^4 are independently hydrogen, chloro, fluoro, methyl or ethyl; R^3 is hydrogen or fluoro; R^5 is 2,5-disubstituted phenyl which substituents are independently selected from the group consisting of halogen, cyano, C_{1-6} alkyl, C_{1-6} alkenyl, C_{3-8} cycloalkyl, C_{1-6} haloalkyl, C_{1-6} alkoxy, C_{1-6} alkylthio and C_{1-6} haloalkoxy; R^7 is hydrogen, methyl or ethyl; R^8 is selected from the group consisting of hydrogen, amino, alkylamino, dialkylamino, aminoalkyl, alkylaminoalkyl and dialkylaminoalkyl and C_{1-6} alkyl optionally substituted with hydroxy, alkoxy, thiol, alkylthio or halogen; and, hydrates, solvates, clathrates and acid addition salts thereof.

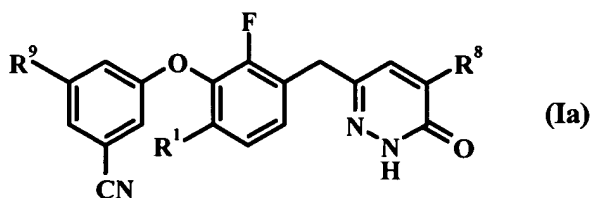
In another embodiment of the present invention there is provided a compound according to formula I wherein X^1 is OR^5 ; R^1 is halogen, methyl or ethyl; R^2 and R^4 are independently halogen, chloro, fluoro, methyl or ethyl; R^3 and R^7 are hydrogen; R^5 is a 2,5-disubstituted phenyl which substituents are
5 independently selected from the group consisting of halogen, cyano, C_{1-6} alkyl and C_{1-6} haloalkyl; and, R^8 is hydrogen, methyl or ethyl and hydrates, solvates, clathrates and acid addition salts thereof.

In another embodiment of the present invention there is provided a compound according to formula I wherein X^1 is OR^5 ; R^1 is methyl, ethyl, trifluoromethyl or halogen; R^2 and R^4 are independently
10 hydrogen, chloro, fluoro, methyl or ethyl; R^3 is hydrogen or fluoro; R^5 is 3,5-disubstituted phenyl which substituents are independently selected from the group consisting of C_{1-6} alkyl, C_{1-6} alkenyl, C_{1-6} haloalkyl, C_{3-8} cycloalkyl, C_{1-6} alkoxy, C_{1-6} alkylthio, C_{1-6} alkylsulfinyl, C_{1-6} sulfonyl, C_{1-6} haloalkoxy, C_{1-6} haloalkylthio, halogen, alkylamino, dialkylamino, aminoacyl, cyano and acyl substituent; R^7 is hydrogen, methyl or ethyl; R^8 is selected from the group consisting of hydrogen, amino, alkylamino,
15 dialkylamino, aminoalkyl, alkylaminoalkyl and dialkylaminoalkyl and C_{1-6} alkyl optionally substituted with hydroxy, alkoxy, thiol, alkylthio or halogen; and, hydrates, solvates, clathrates and acid addition salts thereof.

In another embodiment of the present invention there is provided a compound according to formula I wherein X^1 is OR^5 ; R^1 is methyl, ethyl, trifluoromethyl or halogen; R^2 and R^4 are independently
20 hydrogen, chloro, fluoro, methyl or ethyl; R^3 is hydrogen or fluoro; R^5 is 3,5-disubstituted phenyl which substituents are independently selected from the group consisting of halogen, cyano, C_{1-6} alkyl, C_{1-6} alkenyl, C_{3-8} cycloalkyl, C_{1-6} haloalkyl, C_{1-6} alkoxy, C_{1-6} alkylthio and C_{1-6} haloalkoxy; R^7 is hydrogen, methyl or ethyl; R^8 is selected from the group consisting of hydrogen, amino, alkylamino, dialkylamino,
25 aminoalkyl, alkylaminoalkyl and dialkylaminoalkyl and C_{1-6} alkyl optionally substituted with hydroxy, alkoxy, thiol, alkylthio or halogen; and, hydrates, solvates, clathrates and acid addition salts thereof.

In another embodiment of the present invention there is provided a compound according to formula I wherein X^1 is OR^5 ; R^1 is halogen, methyl or ethyl; R^2 and R^4 are independently halogen, chloro, fluoro,
30 methyl or ethyl; R^3 and R^7 are hydrogen; R^5 is a 3,5-disubstituted phenyl which substituents are independently selected from the group consisting of halogen, cyano, C_{1-6} alkyl and C_{1-6} haloalkyl; and, R^8 is hydrogen, methyl or ethyl and hydrates, solvates, clathrates and acid addition salts thereof.

In another embodiment of the present invention there is provided a compound according to formula **Ia** wherein R¹ is fluoro, chloro, bromo or methyl; R⁸ is hydrogen, methyl or ethyl and R⁹ is C₁₋₆ alkyl, C₃₋₈ cycloalkyl, C₁₋₆ haloalkyl, halogen or cyano.



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In another embodiment of the present invention there is provided a compound according to formula I wherein X¹ is OR⁵; R¹ is methyl, ethyl, trifluoromethyl or halogen; R² and R⁴ are independently hydrogen, chloro, fluoro, methyl or ethyl; R³ is hydrogen or fluoro; R⁵ is 2,4-disubstituted phenyl which substituents are independently selected from the group consisting of C₁₋₆ alkyl, C₁₋₆ alkenyl, C₁₋₆ haloalkyl, C₃₋₈ cycloalkyl, C₁₋₆ alkoxy, C₁₋₆ alkylthio, C₁₋₆ alkylsulfinyl, C₁₋₆ sulfonyl, C₁₋₆ haloalkoxy, C₁₋₆ haloalkylthio, halogen, alkylamino, dialkylamino, aminoacyl, cyano and acyl substituent; R⁷ is hydrogen, methyl or ethyl; R⁸ is selected from the group consisting of hydrogen, amino, alkylamino, dialkylamino, aminoalkyl, alkylaminoalkyl and dialkylaminoalkyl and C₁₋₆ alkyl optionally substituted with hydroxy, alkoxy, thiol, alkylthio or halogen; and, hydrates, solvates, clathrates and acid addition salts thereof.

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In another embodiment of the present invention there is provided a compound according to formula I wherein X¹ is OR⁵; R¹ is methyl, ethyl, trifluoromethyl or halogen; R² and R⁴ are independently hydrogen, chloro, fluoro, methyl or ethyl; R³ is hydrogen or fluoro; R⁵ is 2,4-disubstituted phenyl which substituents are independently selected from the group consisting of halogen, cyano, C₁₋₆ alkyl, C₁₋₆ alkenyl, C₃₋₈ cycloalkyl, C₁₋₆ haloalkyl, C₁₋₆ alkoxy, C₁₋₆ alkylthio and C₁₋₆ haloalkoxy; R⁷ is hydrogen, methyl or ethyl; R⁸ is selected from the group consisting of hydrogen, amino, alkylamino, dialkylamino, aminoalkyl, alkylaminoalkyl and dialkylaminoalkyl and C₁₋₆ alkyl optionally substituted with hydroxy, alkoxy, thiol, alkylthio or halogen; and, hydrates, solvates, clathrates and acid addition salts thereof.

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In another embodiment of the present invention there is provided a compound according to formula I wherein X¹ is OR⁵; R¹ is halogen, methyl or ethyl; R² and R⁴ are independently halogen, chloro, fluoro, methyl or ethyl; R³ and R⁷ are hydrogen; R⁵ is a 2,4-disubstituted phenyl which substituents are independently selected from the group consisting of halogen, cyano, C₁₋₆ alkyl and C₁₋₆ haloalkyl; and, R⁸ is hydrogen, methyl or ethyl and hydrates, solvates, clathrates and acid addition salts thereof.

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In another embodiment of the present invention there is provided a compound according to formula I wherein X^1 is OR^5 ; R^1 is methyl, ethyl, trifluoromethyl or halogen; R^2 and R^4 are independently hydrogen, chloro, fluoro, methyl or ethyl; R^3 is hydrogen or fluoro; R^5 is 2,6-disubstituted phenyl which substituents are independently selected from the group consisting of C_{1-6} alkyl, C_{1-6} alkenyl, C_{1-6} haloalkyl, C_{3-8} cycloalkyl, C_{1-6} alkoxy, C_{1-6} alkylthio, C_{1-6} alkylsulfinyl, C_{1-6} sulfonyl, C_{1-6} haloalkoxy, C_{1-6} haloalkylthio, halogen, alkylamino, dialkylamino, aminoacyl, cyano and acyl substituent; R^7 is hydrogen, methyl or ethyl; R^8 is selected from the group consisting of hydrogen, amino, alkylamino, dialkylamino, aminoalkyl, alkylaminoalkyl and dialkylaminoalkyl and C_{1-6} alkyl optionally substituted with hydroxy, alkoxy, thiol, alkylthio or halogen; and, hydrates, solvates, clathrates and acid addition salts thereof.

In another embodiment of the present invention there is provided a compound according to formula I wherein X^1 is OR^5 ; R^1 is methyl, ethyl, trifluoromethyl or halogen; R^2 and R^4 are independently hydrogen, chloro, fluoro, methyl or ethyl; R^3 is hydrogen or fluoro; R^5 is 2,6-disubstituted phenyl which substituents are independently selected from the group consisting of halogen, cyano, C_{1-6} alkyl, C_{1-6} alkenyl, C_{3-8} cycloalkyl, C_{1-6} haloalkyl, C_{1-6} alkoxy, C_{1-6} alkylthio and C_{1-6} haloalkoxy; R^7 is hydrogen, methyl or ethyl; R^8 is selected from the group consisting of hydrogen, amino, alkylamino, dialkylamino, aminoalkyl, alkylaminoalkyl and dialkylaminoalkyl and C_{1-6} alkyl optionally substituted with hydroxy, alkoxy, thiol, alkylthio or halogen; and, hydrates, solvates, clathrates and acid addition salts thereof.

In another embodiment of the present invention there is provided a compound according to formula I wherein X^1 is OR^5 ; R^1 is halogen, methyl or ethyl; R^2 and R^4 are independently halogen, chloro, fluoro, methyl or ethyl; R^3 and R^7 are hydrogen; R^5 is a 2,6-disubstituted phenyl which substituents are independently selected from the group consisting of is halogen, cyano, C_{1-6} alkyl and C_{1-6} haloalkyl; and, R^8 is hydrogen, methyl or ethyl and hydrates, solvates, clathrates and acid addition salts thereof.

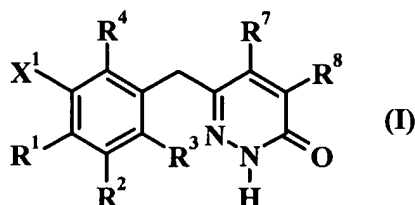
In another embodiment of the present invention there is provided a compound according to formula I wherein X^1 is OR^5 ; R^1 is halogen, methyl or ethyl; R^2 and R^4 are independently halogen, chloro, fluoro, methyl or ethyl; R^3 and R^7 are hydrogen; R^5 is a 2,3,5-trisubstituted phenyl which substituents are independently selected from the group consisting of is halogen, cyano, C_{1-6} alkyl and C_{1-6} haloalkyl, C_{3-8} cycloalkyl, C_{1-6} alkoxy, C_{1-6} alkylthio, C_{1-6} alkylsulfinyl, C_{1-6} sulfonyl, C_{1-6} haloalkoxy, C_{1-6} haloalkylthio, halogen, alkylamino, dialkylamino, aminoacyl, cyano, and acyl ; and, R^8 is hydrogen, methyl or ethyl and hydrates, solvates, clathrates and acid addition salts thereof.

In another embodiment of the present invention there is provided a compound according to formula I wherein X^1 is OR^5 or SR^5 ; R^3 and R^4 are hydrogen, chloro, fluoro or methyl; and R^5 is an optionally substituted heteroaryl selected from the group consisting of pyridinyl, pyridine N-oxide, indole, indole N-oxide, quinoline, quinoline N-oxide, pyrimidinyl, pyrazinyl and pyrrolyl; and R^1 , R^2 , R^7 and R^8 are as defined hereinabove; and hydrates, solvates, clathrates and acid addition salts thereof.

In another embodiment of the present invention there is provided a compound according to formula I wherein X^1 is R^5O , $R^5S(O)_n$, R^5CH_2 , R^5CH_2O , $R^5CH_2S(O)_n$, R^5OCH_2 , $R^5S(O)_nCH_2$ or NR^5R^6 ; R^1 and R^2 along with the atoms to which they are attached form a fused phenyl, dihydropyran, dihydrofuran or furan ring; and R^3 , R^4 , R^5 , R^6 , R^7 and R^8 are as defined hereinabove; and hydrates, solvates, clathrates and acid addition salts thereof.

In another embodiment there is provided a compound according to formula I wherein X^1 is R^5O or R^5S ; R^1 and R^2 along with the atoms to which they are attached form a fused phenyl, dihydropyran, dihydrofuran or furan ring; R^3 and R^7 are hydrogen; R^4 is hydrogen or fluoro; R^8 is hydrogen or methyl; R^5 is optionally substituted phenyl; R^7 is as defined hereinabove; and hydrates, solvates, clathrates and acid addition salts thereof.

In another embodiment of the present invention there is provided a method for treating an HIV infection, or preventing an HIV infection, or treating AIDS or ARC, comprising administering to a host in need thereof a therapeutically effective amount of a compound of formula I



wherein, X^1 , R^1 , R^2 , R^3 , R^4 , R^5 , R^6 , R^7 and R^8 are as defined hereinabove, and hydrates, solvates, clathrates and acid addition salts thereof.

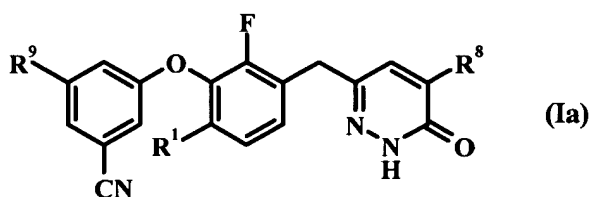
In another embodiment of the present invention there is provided a method for treating an HIV infection, or preventing an HIV infection, or treating AIDS or ARC, comprising administering to a host in need thereof a therapeutically effective amount of a compound of formula I wherein, X^1 is OR^5 , R^1 is methyl, ethyl, trifluoromethyl or halogen; R^2 and R^4 are hydrogen, fluoro, chloro, methyl or ethyl; R^3 is

hydrogen or fluoro; R⁵ is optionally substituted phenyl, R⁷ is hydrogen, methyl or ethyl; and R⁸ is as defined hereinabove, and hydrates, solvates, clathrates and acid addition salts thereof.

In another embodiment of the present invention there is provided a method for treating an HIV

- 5 infection, or preventing an HIV infection, or treating AIDS or ARC, comprising administering to a host in need thereof a therapeutically effective amount of a compound of formula **Ia** wherein R¹ is selected from the group consisting of fluoro, chloro, bromo and methyl; R⁸ is selected from the group consisting of hydrogen, methyl and ethyl; and, R⁹ is selected from the group consisting of alkyl, cycloalkyl, haloalkyl, halogen and cyano.

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In another embodiment of the present invention there is provided a method for treating an HIV

- infection, or preventing an HIV infection, or treating AIDS or ARC, comprising co-administering to a
15 host in need thereof a therapeutically effective amount of a compound of formula **I** wherein, X¹ is OR⁵, R¹ is methyl, ethyl, trifluoromethyl or halogen; R² and R⁴ are hydrogen, fluoro, chloro, methyl or ethyl; R³ is hydrogen or fluoro; R⁵ is optionally substituted phenyl, R⁷ is hydrogen, methyl or ethyl; and R⁸ is as defined hereinabove, and hydrates, solvates, clathrates and acid addition salts thereof; and, at least one compound selected from the group consisting of HIV protease inhibitors, nucleoside reverse
20 transcriptase inhibitors, non-nucleoside reverse transcriptase inhibitors, CCR5 inhibitors and viral fusion inhibitors.

In another embodiment of the present invention there is provided a method for treating an HIV

- infection, or preventing an HIV infection, or treating AIDS or ARC, comprising co-administering to a
25 host in need thereof a therapeutically effective amount of a compound of formula **I** wherein, X¹ is OR⁵, R¹ is methyl, ethyl, trifluoromethyl or halogen; R² and R⁴ are hydrogen, fluoro, chloro, methyl or ethyl; R³ is hydrogen or fluoro; R⁵ is optionally substituted phenyl, R⁷ is hydrogen, methyl or ethyl; and R⁸ is as defined hereinabove, and hydrates, solvates, clathrates and acid addition salts thereof; and at least one compound selected from the group consisting of saquinavir, ritonavir, nelfinavir, indinavir, amprenavir,
30 lopinavir, and/or reverse transcriptase inhibitors selected from the group consisting of zidovudine,

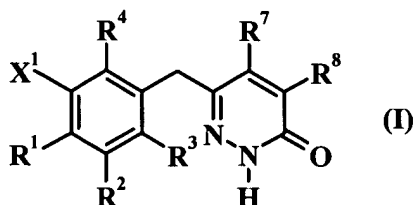
lamivudine, didanosine, zalcitabine, stavudine, rescriptor, sustiva, virumune, efavirenz, nevirapine and delavirdine.

5 In another embodiment of the present invention there is provided a method for inhibiting a retroviral reverse transcriptase comprising administering to a host in need thereof a therapeutically effective amount of a compound of formula **I** wherein, X^1 , R^1 , R^2 , R^3 , R^4 , R^5 , R^6 , R^7 and R^8 are as defined hereinabove, and hydrates, solvates, clathrates and acid addition salts thereof.

10 In another embodiment of the present invention there is provided a method for inhibiting a retroviral reverse transcriptase having at least one mutation with respect to the wild type virus comprising administering to a host in need thereof a therapeutically effective amount of a compound of formula **I** wherein, X^1 , R^1 , R^2 , R^3 , R^4 , R^5 , R^6 , R^7 and R^8 are as defined hereinabove, and hydrates, solvates, clathrates and acid addition salts thereof.

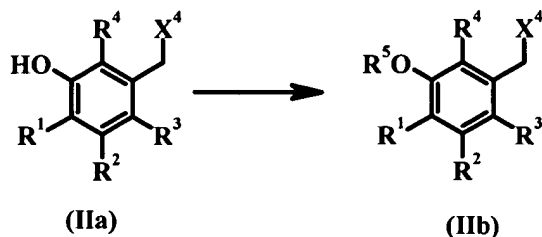
15 In another embodiment of the present invention there is provided a method for treating an HIV infection, or preventing an HIV infection, or treating AIDS or ARC wherein the host is infected with a strain of HIV with reduced susceptibility to efavirenz, nevirapine or delavirdine comprising administering to a host in need thereof a therapeutically effective amount of a compound of formula **I** wherein, X^1 , R^1 , R^2 , R^3 , R^4 , R^5 , R^6 , R^7 and R^8 are as defined hereinabove, and hydrates, solvates, 20 clathrates and acid addition salts thereof.

In another embodiment of the present invention there is provided a pharmaceutical composition comprising a therapeutically effective quantity of a compound of formula **I**,

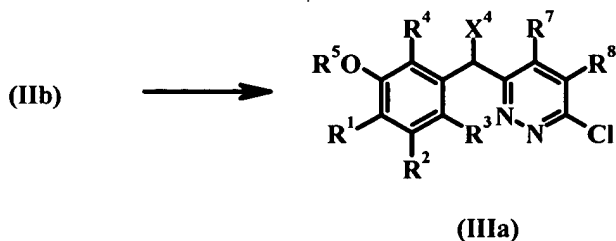


25 wherein, X^1 is R^5O , $R^5S(O)_n$, R^5CH_2 , R^5CH_2O , $R^5CH_2S(O)_n$, R^5OCH_2 , $R^5S(O)_nCH_2$ or NR^5R^6 ; is hydrogen, C_{1-6} alkyl, C_{1-6} haloalkyl, C_{3-8} cycloalkyl, C_{1-6} alkylthio, C_{1-6} haloalkylthio, halogen, amino, alkylamino, dialkylamino, aminoacyl, nitro and cyano; and R^1 , R^2 , R^3 , R^4 , R^5 , R^6 , R^7 and R^8 are as defined hereinabove, hydrates, solvates, clathrates and acid addition salts thereof in admixture with at least one pharmaceutically acceptable carrier, excipient or diluent.

In another embodiment of the present invention there is provided a process for preparing a compound of formula **I** wherein X^1 is OR^5 or SR^5 , R^5 is an optionally substituted aryl, alkyl or aralkyl moiety and R^1 , R^4 , R^7 and R^8 are as defined hereinabove which process comprises the steps of: (i) coupling an aryl
 5 compound of formula **IIa** wherein



X^4 is hydrogen, alkoxy carbonyl or nitrile with (A) an aryl boronic acid or an aryl halide, or (B) an alcohol, alkyl halide or aralkyl halide to produce an ether of formula **IIb**; (ii) if X^4 is hydrogen, (a) brominating the methyl group with NBS and displacing the bromide ($X^4 = Br$) with sodium cyanide to
 10 produce the corresponding nitrile ($X^4 = CN$); (iii) treating a compound of formula **IIb** with base and condensing the



conjugate base with a pyrazine compound to produce a compound of formula **IIIa**; and (iv) subjecting the alkoxy carbonyl or nitrile to acidic or basic hydrolysis, decarboxylating the resulting carboxylic acid
 15 and hydrolysing the chloropyrazine to a pyridazinone of formula **I** with acetic acid and aqueous hydrochloric acid.

In another embodiment of the present invention there is provided a process as described above for preparing a compound of formula **I** wherein X^4 hydrogen, alkoxy carbonyl or nitrile, R^5 is optionally
 20 substituted aryl and the ether is prepared by coupling an arylboronic acid and a phenol **IIa** in the presence of a $Cu(II)$ salt.

In another embodiment of the present invention there is provided a process as described above for preparing a compound of formula **I** wherein X^4 is hydrogen, alkoxy carbonyl or nitrile, R^5 is optionally
 25 substituted aryl and the ether is prepared by coupling an aryl halide and a phenol **IIa** in the presence of a $Cu(I)$ salt.

5 In another embodiment of the present invention there is provided a process as described above for preparing a compound of formula **I** wherein X^4 is hydrogen, alkoxycarbonyl or nitrile, R^5 is optionally substituted aryl, alkyl or aralkyl moiety and the ether is prepared by coupling an aryl halide further substituted by electron withdrawing groups, an optionally substituted alkyl halide or an optionally substituted aralkyl halide and a phenol **IIa** in the presence of a base.

10 In another embodiment of the present invention there is provided a process as described above for preparing a compound of formula **IIIa** wherein the base is sodium hydride and the pyrazine compound is a 3,6-dihalopyrazine or a 3-halo-6-alkoxypyrazine.

15 In another embodiment of the present invention there is provided a process as described above for preparing a compound of formula **I** wherein the acidic hydrolysis conditions comprise a carboxylic acid and an aqueous hydrohalic acid.

In another embodiment of the present invention there is provided a process as described above for preparing a compound of formula **I** wherein the acidic hydrolysis conditions comprise acetic acid and aqueous hydrochloric acid.

20 In another embodiment of the present invention there is provided a process as described above for preparing a compound of formula **I** wherein the acidic hydrolysis conditions comprise acetic acid, sodium acetate and aqueous hydrochloric acid.

25 In another embodiment of the present invention there is provided a process as described above for preparing a compound of formula **I** wherein the alkoxycarbonyl compound is hydrolyzed with base and said chloropyrazine is hydrolyzed with acetic acid and aqueous hydrochloric acid.

DEFINITIONS

30 The phrase "a" or "an" entity as used herein refers to one or more of that entity; for example, a compound refers to one or more compounds or at least one compound. As such, the terms "a" (or "an"), "one or more", and "at least one" can be used interchangeably herein.

The phrase "as defined hereinabove" refers to the first definition provided in the Summary of the Invention.

5 The term "optional" or "optionally" as used herein means that a subsequently described event or circumstance may, but need not, occur, and that the description includes instances where the event or circumstance occurs and instances in which it does not. For example, "optionally substituted" means that the moiety may be hydrogen or a substituent.

10 The term "C₁₋₆ alkyl" as used herein denotes an unbranched or branched chain, saturated, monovalent hydrocarbon residue containing 1 to 6 carbon atoms. Examples of alkyl groups include, but are not limited to, lower alkyl groups include methyl, ethyl, propyl, *i*-propyl, *n*-butyl, *i*-butyl, *t*-butyl or pentyl, isopentyl, neopentyl, hexyl.

15 The term "haloalkyl" as used herein denotes an unbranched or branched chain alkyl group as defined above wherein 1, 2, 3 or more hydrogen atoms are substituted by a halogen. Examples are 1-fluoromethyl, 1-chloromethyl, 1-bromomethyl, 1-iodomethyl, trifluoromethyl, trichloromethyl, tribromomethyl, triiodomethyl, 1-fluoroethyl, 1-chloroethyl, 1-bromoethyl, 1-iodoethyl, 2-fluoroethyl, 2-chloroethyl, 2-bromoethyl, 2-iodoethyl, 2,2-dichloroethyl, 3-bromopropyl or 2,2,2-trifluoroethyl.

20 The term "C₃₋₈ cycloalkyl" as used herein denotes a saturated carbocyclic ring containing 3 to 8 carbon atoms, i.e. cyclopropyl, cyclobutyl, cyclopentyl, cyclohexyl, cycloheptyl or cyclooctyl.

25 The term "aryl" as used herein means a monocyclic or polycyclic-aromatic group comprising carbon and hydrogen atoms. Examples of suitable aryl groups include, but are not limited to, phenyl, tolyl, indenyl, and 1- or 2-naphthyl, as well as benzo-fused carbocyclic moieties such as 5,6,7,8-tetrahydronaphthyl. An aryl group can be unsubstituted or substituted with one or more suitable substituents which substituents include C₁₋₆ alkyl, C₁₋₆ haloalkyl, C₃₋₈ cycloalkyl, C₁₋₆ alkoxy, C₁₋₆ alkylthio, C₁₋₆ alkylsulfinyl, C₁₋₆ sulfonyl, C₁₋₆ haloalkoxy, C₁₋₆ haloalkylthio, halogen, amino, alkylamino, dialkylamino, aminoacyl, acyl, alkoxycarbonyl, carbamoyl, N-alkylcarbamoyl, N,N-dialkylcarbamoyl, nitro and cyano.

30 A "heteroaryl group" or "heteroaromatic" as used herein means a monocyclic- or polycyclic aromatic ring comprising up to 15 carbon atoms, hydrogen atoms, and one or more heteroatoms, preferably, 1 to 3 heteroatoms, independently selected from nitrogen, oxygen, and sulfur. As well known to those skilled in the art, heteroaryl rings have less aromatic character than their all-carbon counter parts. Thus, for the purposes of the invention, a heteroaryl group need only have some degree of aromatic character.

Illustrative examples of heteroaryl groups include, but are not limited to, pyridinyl, pyridine N-oxide, pyridazinyl, pyrimidyl, pyrazyl, triazinyl, pyrrolyl, pyrazolyl, imidazolyl, (1,2,3,-) and (1,2,4)-triazolyl, pyrazinyl, pyrimidinyl, tetrazolyl, furyl, thienyl, isoxazolyl, thiazolyl, thienyl, isoxazolyl, indole, indole N-oxide, quinoline, quinoline N-oxide and oxazolyl. A heteroaryl group can be unsubstituted or substituted with one or more suitable substituents selected from hydroxy, oxo, cyano, alkyl, alkoxy, haloalkoxy, alkylthio, halo, haloalkyl, nitro, alkoxycarbonyl, amino, alkylamino, dialkylamino, aminoacyl, alkylsulfonyl, arylsulfinyl, alkoxycarbonyl, carbamoyl, N-alkylcarbamoyl, N,N-dialkylcarbamoyl, acyl unless otherwise indicated.

The term "heterocyclyl" means the monovalent saturated cyclic radical, consisting of one or more rings, preferably one to two rings, of three to eight atoms per ring, incorporating one or more ring heteroatoms (chosen from N,O or S(O)₀₋₂), and which can optionally be substituted with one or more, preferably one to three substituents selected from hydroxy, oxo, cyano, alkyl, alkoxy, haloalkoxy, alkylthio, halo, haloalkyl, nitro, alkoxycarbonyl, amino, alkylamino, dialkylamino, aminoacyl, alkylsulfonyl, arylsulfinyl, alkoxycarbonyl, carbamoyl, N-alkylcarbamoyl, N,N-dialkylcarbamoyl, acyl unless otherwise indicated. Examples of heterocyclic radicals include, but are not limited to, furanyl, tetrahydropyranyl, tetrahydrothiophenyl and the like. A nitrogen atom in the heteroaryl ring can optionally be an N-oxide.

The term "alkoxy group" as used herein means an -O-alkyl group, wherein alkyl is as defined above such as methoxy, ethoxy, *n*-propyloxy, *i*-propyloxy, *n*-butyloxy, *i*-butyloxy, *t*-butyloxy, pentyloxy, hexyloxy, heptyloxy including their isomers.

The term "alkylthio group" as used herein means an -S-alkyl group, wherein alkyl is as defined above such as meththio, eththio, *n*-propylthio, *i*-propylthio, *n*-butylthio, *i*-butylthio, *t*-butylthio, pentylthio including their isomers.

The term "haloalkoxy group" as used herein means an -O-haloalkyl group, wherein haloalkyl is as defined above. Examples of haloalkoxy groups include, but are not limited to, 2,2,2-trifluoroethoxy, difluoromethoxy and 1,1,1,3,3,3-hexafluoro-*iso*-propoxy.

The term "haloalkthio group" as used herein means an -S-haloalkyl group, wherein haloalkyl is as defined above. An example of haloalkthio group includes, but are not limited to, 2,2,2-trifluoroeththanthiol.

5 The term "aryloxy group" as used herein means an O-aryl group wherein aryl is as defined above. An aryloxy group can be unsubstituted or substituted with one or more suitable substituents. Preferably, the aryl ring of an aryloxy group is a monocyclic ring, wherein the ring comprises 6 carbon atoms, referred to herein as "(C₆) aryloxy". The term "optionally substituted aryloxy" means the aryl or group may be substituted with one to three groups selected from the group consisting of C₁₋₆ alkyl, C₁₋₆ haloalkyl, C₃₋₈ cycloalkyl, C₁₋₆ alkoxy, C₁₋₆ alkylthio, C₁₋₆ alkylsulfinyl, C₁₋₆ sulfonyl, C₁₋₆ haloalkoxy, C₁₋₆ haloalkylthio, halogen, amino, alkylamino, dialkylamino, aminoacyl, acyl, alkoxycarbonyl, carbamoyl, N-alkylcarbamoyl, N,N-dialkylcarbamoyl, nitro and cyano.

10

The term "heteroaryloxy group" as used herein means an O-heteroaryl group, wherein heteroaryl is as defined above. The heteroaryl ring of a heteroaryloxy group can be unsubstituted or substituted with one or more suitable substituents. Examples of heteroaryl groups include, but are not limited to, 2-pyridyloxy, 3-pyrrolyloxy, 3-pyrazolyloxy, 2-imidazolyloxy, 3-pyrazinyloxy, and 4-pyrimidyloxy.

15

The term "acyl" or "alkylcarbonyl" as used herein denotes a radical of formula C(=O)R wherein R is hydrogen, unbranched or branched alkyl containing 1 to 6 carbon atoms or a phenyl group.

20 The term "alkoxycarbonyl" as used herein denotes a radical of formula C(=O)OR wherein R is, unbranched or branched alkyl as described above.

The term "acylamino" as used herein denotes a radical of formula -NH-(acyl) where acyl is as defined herein.

25 The term "arylboronic acid" as used herein denotes a radical of formula ArB(OH)₂ wherein Ar is an optionally substituted aryl group as described above.

30 The term "alkylene" as used herein denotes a divalent linear or branched saturated hydrocarbon radical, having from one to six carbons inclusive, unless otherwise indicated. Examples of alkylene radicals include, but are not limited to, methylene, ethylene, propylene, 2-methyl-propylene, butylene, 2-ethylbutylene.

The term "arylalkyl" or "aralkyl" as used herein denotes the radical R'R"-, wherein R' is an aryl radical as defined herein, and R" is an alkylene radical as defined herein and the arylalkyl group is attached

through the alkylene radical. Examples of arylalkyl radicals include, but are not limited to, benzyl, phenylethyl, 3-phenylpropyl.

5 The term "halogen" as used herein means fluorine, chlorine, bromine, or iodine. Correspondingly, the meaning of the term "halo" encompasses fluoro, chloro, bromo, and iodo. The term "hydrohalic acid" refers to an acid comprised of hydrogen and a halogen.

The term "alkylsulfinyl" as used herein means the radical $-S(O)R'$, wherein R' is alkyl as defined herein. Examples of alkylaminosulfonyl include, but are not limited to methylsulfinyl and *iso*-propylsulfinyl.

10

The term "alkylsulfonyl" as used herein means the radical $-S(O)_2R'$, wherein R' is alkyl as defined herein. Examples of alkylaminosulfonyl include, but are not limited to methylsulfonyl and *iso*-propylsulfonyl.

15 The terms "amino", "alkylamino" and "dialkylamino" as used herein refer to $-NH_2$, $-NHR$ and $-NR_2$ respectively and R is alkyl as defined above. The two alkyl groups attached to a nitrogen in a dialkyl moiety can be the same or different. The terms "aminoalkyl", "alkylaminoalkyl" and "dialkylaminoalkyl" as used herein refer to $NH_2(CH_2)_n$ -, $RHN(CH_2)_n$ -, and $R_2N(CH_2)_n$ - respectively wherein n is 1 to 6 and R is alkyl as defined above

20 The prefix "carbamoyl" as used herein means the radical $-CONH_2$. The prefix "N-alkylcarbamoyl" and "N,N-dialkylcarbamoyl" means the radical $CONHR'$ or $CONR'R''$ respectively wherein the R' and R'' groups are independently alkyl as defined herein.

25 The term "conjugate base" as used herein means the chemical species produced when an acid (including here a carbon acid) gives up its proton.

Compounds of formula **I** exhibit tautomerism. Tautomeric compounds can exist as two or more interconvertible species. Prototropic tautomers result from the migration of a covalently bonded hydrogen atom between two atoms. Tautomers generally exist in equilibrium and attempts to isolate an individual tautomers usually produce a mixture whose chemical and physical properties are consistent with a mixture of compounds. The position of the equilibrium is dependent on chemical features within the molecule. For example, in many aliphatic aldehydes and ketones, such as acetaldehyde, the keto form predominates while; in phenols, the enol form predominates. Common prototropic tautomers

30

include keto/enol ($\text{-C(=O)-CH-} \rightleftharpoons \text{-C(-OH)=CH-}$), amide/imidic acid ($\text{-C(=O)-NH-} \rightleftharpoons \text{-C(-OH)=N-}$) and amidine ($\text{-C(=NR)-NH-} \rightleftharpoons \text{-C(-NHR)=N-}$) tautomers. The latter two are particularly common in heteroaryl and heterocyclic rings and the present invention encompasses all tautomeric forms of the compounds.

5

Compounds of formula I which are basic can form pharmaceutically acceptable acid addition salts with inorganic acids such as hydrohalic acids (e.g. hydrochloric acid and hydrobromic acid), sulphuric acid, nitric acid and phosphoric acid, and the like, and with organic acids (e.g. with acetic acid, tartaric acid, succinic acid, fumaric acid, maleic acid, malic acid, salicylic acid, citric acid, methanesulphonic acid and *p*-toluenesulfonic acid, and the like).

10

The term "solvate" as used herein means a compound of the invention or a salt, thereof, that further includes a stoichiometric or non-stoichiometric amount of a solvent bound by non-covalent intermolecular forces. Preferred solvents are volatile, non-toxic, and/or acceptable for administration to humans in trace amounts.

15

The term "hydrate" as used herein means a compound of the invention or a salt thereof, that further includes a stoichiometric or non-stoichiometric amount of water bound by non-covalent intermolecular forces.

20

The term "clathrate" as used herein means a compound of the invention or a salt thereof in the form of a crystal lattice that contains spaces (e. g., channels) that have a guest molecule (e. g., a solvent or water) trapped within.

The term "wild type" as used herein refers to the HIV virus strain which possesses the dominant genotype which naturally occurs in the normal population which has not been exposed to reverse transcriptase inhibitors. The term "wild type reverse transcriptase" used herein has refers to the reverse transcriptase expressed by the wild type strain which has been sequenced and deposited in the SwissProt database with an accession number P03366.

30

The term "reduced susceptibility" as used herein refers to about a 10 fold, or greater, change in sensitivity of a particular viral isolate compared to the sensitivity exhibited by the wild type virus in the same experimental system

The term "nucleoside and nucleotide reverse transcriptase inhibitors" ("NRTI"s) as used herein means nucleosides and nucleotides and analogues thereof that inhibit the activity of HIV-1 reverse transcriptase, the enzyme which catalyzes the conversion of viral genomic HIV-1 RNA into proviral HIV-1 DNA.

5

Typical suitable NRTIs include zidovudine (AZT) available under the RETROVIR tradename; didanosine (ddI) available under the VIDEX tradename.; zalcitabine (ddC) available under the HIVID tradename; stavudine (d4T) available under the ZERIT trademark.; lamivudine (3TC) available under the EPIVIR tradename; abacavir (1592U89) disclosed in WO96/30025 and available under the ZIAGEN trademark; adefovir dipivoxil [bis(POM)-PMEA] available under the PREVON tradename; lobucavir (BMS-180194), a nucleoside reverse transcriptase inhibitor disclosed in EP-0358154 and EP-0736533 and under development by Bristol-Myers Squibb; BCH-10652, a reverse transcriptase inhibitor (in the form of a racemic mixture of BCH-10618 and BCH-10619) under development by Biochem Pharma; emtricitabine [(-)-FTC] licensed from Emory University under U.S. Pat. No. 5,814,639 and under development by Triangle Pharmaceuticals; beta-L-FD4 (also called beta-L-D4C and named beta-L-2', 3'-dideoxy-5-fluoro-cytidine) licensed by Yale University to Vion Pharmaceuticals; DAPD, the purine nucleoside, (-)-beta-D-2,6,-diamino-purine dioxolane disclosed in EP-0656778 and licensed to Triangle Pharmaceuticals; and lodenosine (FddA), 9-(2,3-dideoxy-2-fluoro-b-D-threo-pentofuranosyl)adenine, an acid stable purine-based reverse transcriptase inhibitor discovered by the NIH and under development by U.S. Bioscience Inc..

20

The term "non-nucleoside reverse transcriptase inhibitors" ("NNRTI"s) as used herein means non-nucleosides that inhibit the activity of HIV-1 reverse transcriptase.

Typical suitable NNRTIs include nevirapine (BI-RG-587) available under the VIRAMUNE tradename; delaviradine (BHAP, U-90152) available under the RESCRIPTOR tradename; efavirenz (DMP-266) a benzoxazin-2-one disclosed in WO94/03440 and available under the SUSTIVA tradename; PNU-142721, a furopyridine-thio-pyrimide; AG-1549 (formerly Shionogi # S-1153); 5-(3,5-dichlorophenyl)-thio-4-isopropyl-1-(4-pyridyl)methyl-1H-imidazol-2-ylmethyl carbonate disclosed in WO 96/10019; MKC-442 (1-(ethoxy-methyl)-5-(1-methylethyl)-6-(phenylmethyl)-(2,4(1H,3H)-pyrimidinedione); and (+)-calanolide A (NSC-675451) and B, coumarin derivatives disclosed in U.S. Pat. No. 5,489,697

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The term "protease inhibitor" ("PI") as used herein means inhibitors of the HIV-1 protease, an enzyme required for the proteolytic cleavage of viral polyprotein precursors (e.g., viral GAG and GAG Pol

polyproteins), into the individual functional proteins found in infectious HIV-1. HIV protease inhibitors include compounds having a peptidomimetic structure, high molecular weight (7600 daltons) and substantial peptide character, e.g. CRIXIVAN as well as nonpeptide protease inhibitors e.g., VIRACEPT.

5

Typical suitable PIs include saquinavir available in hard gel capsules under the INVIRASE tradename and as soft gel capsules under the FORTOVASE tradename; ritonavir (ABT-538) available under the NORVIR tradename; indinavir (MK-639) available under the CRIXIVAN tradename; nelfinavir (AG-1343) available under the VIRACEPT; amprenavir (141 W94), tradename AGENERASE, a non-peptide
10 protease inhibitor; lasinavir (BMS-234475; originally discovered by Novartis, Basel, Switzerland (CGP-61755); DMP-450, a cyclic urea discovered by Dupont; BMS-2322623, an azapeptide under development by Bristol-Myers Squibb, as a 2nd-generation HIV-1 PI; ABT-378; AG-1549 an orally active imidazole carbamate.

15 Other antiviral agents include hydroxyurea, ribavirin, IL-2, IL-12, pentafuside and Yisum Project No. 11607. Hydroxyurea (Droxia), a ribonucleoside triphosphate reductase inhibitor, the enzyme involved in the activation of T-cells. Hydroxyurea was shown to have a synergistic effect on the activity of didanosine and has been studied with stavudine. IL-2 is disclosed in Ajinomoto EP-0142268, Takeda EP-0176299, and Chiron U.S. Pat. Nos. RE 33,653, 4,530,787, 4,569,790, 4,604,377, 4,748,234,
20 4,752,585, and 4,949,314, and is available under the PROLEUKIN (aldesleukin) tradename as a lyophilized powder for IV infusion or sc administration upon reconstitution and dilution with water; a dose of about 1 to about 20 million IU/day, sc is preferred; a dose of about 15 million IU/day, sc is more preferred. IL-12 is disclosed in WO96/25171 and is available as a dose of about 0.5 microgram/kg/day to about 10 microgram/kg/day, sc is preferred. Pentafuside (DP-178, T-20) a 36-
25 amino acid synthetic peptide, disclosed in U.S. Pat. No. 5,464,933 and available under the FUZEON tradename; pentafuside acts by inhibiting fusion of HIV-1 to target membranes. Pentafuside (3-100 mg/day) is given as a continuous sc infusion or injection together with efavirenz and 2 PIs to HIV-1 positive patients refractory to a triple combination therapy; use of 100 mg/day is preferred. Yisum Project No. 11607, a synthetic protein based on the HIV-1 Vif protein. Ribavirin, 1-.beta.-D-
30 ribofuranosyl-1H-1,2,4-triazole-3-carboxamide, is described in U.S. Pat. No. 4,211,771.

The term "anti-HIV-1 therapy" as used herein means any anti-HIV-1 drug found useful for treating HIV-1 infections in man alone, or as part of multidrug combination therapies, especially the HAART triple and quadruple combination therapies. Typical suitable known anti-HIV-1 therapies include, but

are not limited to multidrug combination therapies such as (i) at least three anti-HIV-1 drugs selected from two NRTIs, one PI, a second PI, and one NNRTI; and (ii) at least two anti-HIV-1 drugs selected from NNRTIs and PIs. Typical suitable HAART--multidrug combination therapies include:

- 5 (a) triple combination therapies such as two NRTIs and one PI; or (b) two NRTIs and one NNRTI; and (c) quadruple combination therapies such as two NRTIs, one PI and a second PI or one NNRTI. In treatment of naive patients, it is preferred to start anti-HIV-1 treatment with the triple combination therapy; the use of two NRTIs and one PI is preferred unless there is intolerance to PIs. Drug compliance is essential. The CD4^{sup.} and HIV-1-RNA plasma levels should be monitored every 3-6
- 10 months. Should viral load plateau, a fourth drug, e.g., one PI or one NNRTI could be added.

ABBREVIATIONS

The following abbreviations are used throughout this application and they have the meaning listed below:

15	AIBN	azo- <i>bis</i> -isobutyrylnitrile
	atm	atmospheres
	BBN or 9-BBN	9-borabicyclo[3.3.1]nonane
	Boc	<i>tert</i> -butoxycarbonyl
	BOC ₂ O	Di- <i>tert</i> -butyl pyrocarbonate or boc anhydride
20	Bn	benzyl
	cbz or Z	benzyloxycarbonyl
	DABCO	diazabicyclooctane
	DAST	diethylaminosulfur trifluoride
	DCE	1,2-dichloroethane
25	DCM	dichloromethane
	DEAD	diethyl azodicarboxylate
	DIAD	di- <i>iso</i> -propylazodicarboxylate
	DIBAL-H	di- <i>iso</i> -butylaluminumhydride
	DMA	N,N-dimethyl acetamide
30	DMAP	4-N,N-dimethylaminopyridine
	DMF	N,N-dimethylformamide
	dppf	1,1'-Bis(diphenylphosphino)ferrocene
	EDCI	1-(3-dimethylaminopropyl)-3-ethylcarbodiimide hydrochloride
	EtOAc	ethyl acetate

	Et ₂ O	diethyl ether
	Et	ethyl
	EtOH	ethanol
	LAH	lithium aluminum hydride
5	LiHMDS	lithium hexamethyl disilazane
	h	hour(s)
	HOAc	acetic acid
	<i>i</i> -Pr	iso-propyl
	m	minute(s)
10	Me	methyl
	MeCN	acetonitrile
	MeOH	methanol
	MTBE	methyl <i>t</i> -butyl ether
	NBS	N-bromosuccinimide
15	NMP	N-methylpyrrolidone
	PCC	pyridinium chlorochromate
	PDC	pyridinium dichromate
	psi	pounds per square inch
	pyr	pyridine
20	rt or RT	room temperature
	TEA or Et ₃ N	triethylamine
	Tf	triflate CF ₃ SO ₂ -
	TFA	trifluoroacetic acid
	THF	tetrahydrofuran
25	TLC	thin layer chromatography
	TMHD	2,2,6,6-tetramethylheptane-2,6-dione
	TsOH	<i>p</i> -toluenesulfonic acid monohydrate

EXAMPLES OF COMPOUNDS

Examples of representative compounds within the scope of the invention are provided in the Tables I, II and III.. Table I depicts new compounds wherein X¹ is OR⁵ and R⁵ is optionally substituted aryl or heteroaryl encompassed by the present invention. Table II depicts new compounds wherein X¹ is other than OR⁵ and R⁵ is optionally substituted aryl or heteroaryl encompassed by the present invention. Table III depicts COX-2 inhibitors wherein X¹ is -C(=O)-which have now, unexpectedly, been found to inhibit RT (EP 810218; D. A. Allen *et al.*). These examples and preparations are provided to enable those skilled in the art to more clearly understand and to practice the present invention. They should not be considered as limiting the scope of the invention, but merely as being illustrative and representative thereof.

In general, the nomenclature used in this Application is based on AUTONOMTM v.4.0, a Beilstein Institute computerized system for the generation of IUPAC systematic nomenclature.

Table 1 contains representative examples of [3-phenoxybenzyl]pyridazinones. References in the application to compounds in the tables are designated by the table number followed by a dash and the compound number (*e.g.*, I-100)

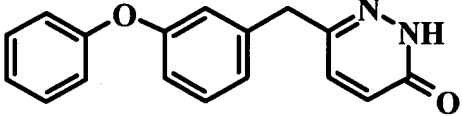
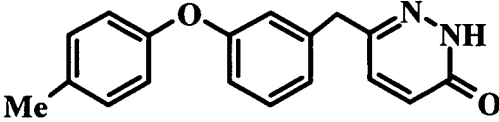
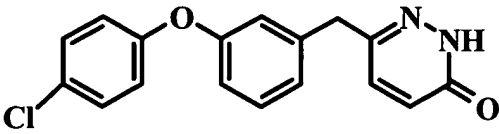
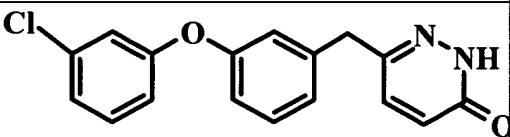
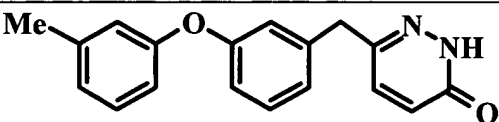
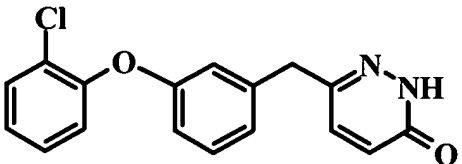
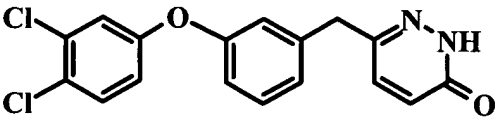
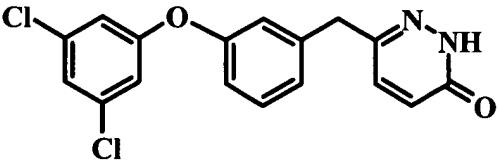
Table I			
cpd #	Structure	Name	[M+H] ⁺ (mw) mp
I-1		6-(3-Phenoxy-benzyl)-2H-pyridazin-3-one	278 ¹ (278.31)
I-2		6-(3-p-Tolyloxy-benzyl)-2H-pyridazin-3-one	292 ¹ (292.33)
I-3		6-[3-(4-Chloro-phenoxy)-benzyl]-2H-pyridazin-3-one	312 ¹ (312.75)
I-4		6-[3-(3-Chloro-phenoxy)-benzyl]-2H-pyridazin-3-one	312 ¹ (312.75)
I-5		6-(3-m-Tolyloxy-benzyl)-2H-pyridazin-3-one	292 ¹ (292.33)
I-6		6-[3-(2-Chloro-phenoxy)-benzyl]-2H-pyridazin-3-one	312 ¹ (312.75)
I-7		6-[3-(3,4-Dichloro-phenoxy)-benzyl]-2H-pyridazin-3-one	347 ¹ (347.2)
I-8		6-[3-(3,5-Dichloro-phenoxy)-benzyl]-2H-pyridazin-3-one	347 ¹ (347.2)

Table I			
cpd #	Structure	Name	[M+H] ⁺ (mw) mp
I-9		6-[3-(3,5-Dimethyl-phenoxy)-5-methoxy-benzyl]-2H-pyridazin-3-one	336 ¹ (336.38)
I-10		6-[3-(3,5-Dichloro-phenoxy)-5-methoxy-benzyl]-2H-pyridazin-3-one	377 ¹ (377.22)
I-11		6-[3-(3,5-Dichloro-phenoxy)-5-ethoxy-benzyl]-2H-pyridazin-3-one	391 ¹ (391.25)
I-12		6-[3-(3,5-Dichloro-phenoxy)-4-methoxy-benzyl]-2H-pyridazin-3-one	377 (377.22)
I-13		6-(4-Methoxy-3-phenoxy-benzyl)-2H-pyridazin-3-one	308 ¹ (308.33)
I-14		6-[3-(4-Chloro-phenoxy)-4-methoxy-benzyl]-2H-pyridazin-3-one	342 ¹ (342.78)
I-15		6-(5-Methoxy-3-phenoxy-benzyl)-2H-pyridazin-3-one	308 ¹ (308.33)
I-16		6-[3-(4-Chloro-phenoxy)-5-methoxy-benzyl]-2H-pyridazin-3-one	342 ¹ (342.78)

Table I			
cpd #	Structure	Name	[M+H] ⁺ (mw) mp
I-17		6-[3-(3,5-Dichloro-phenoxy)-5-ethyl-benzyl]-2H-pyridazin-3-one	375 ^I (375.25)
I-18		6-[3-(4-Chloro-phenoxy)-4-methyl-benzyl]-2H-pyridazin-3-one	327 (326.78)
I-19		6-(4-Methyl-3-phenoxy-benzyl)-2H-pyridazin-3-one	293 (292.33) 83-85
I-20		3-Cyano-5-[3-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzoic acid methyl ester	361 ^I (361.35)
I-21		5-[3-(6-Oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-isophthalonitrile	328 ^I (328.33)
I-22		4-[2-Methoxy-5-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	333 ^I (333.35)
I-23		3-[2-Methoxy-5-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	333 ^I (333.35)
I-24		6-[3-(4-Chloro-phenoxy)-4-ethyl-benzyl]-2H-pyridazin-3-one	341 (340.80)

Table I			
cpd #	Structure	Name	[M+H] ⁺ (mw) mp
I-25		6-[3-(3-Chloro-phenoxy)-5-methyl-benzyl]-2H-pyridazin-3-one	326 ¹ (326.78)
I-26		6-[3-(4-Chloro-phenoxy)-5-methyl-benzyl]-2H-pyridazin-3-one	326 ¹ (326.78)
I-27		6-[3-(3,5-Dichloro-phenoxy)-5-methyl-benzyl]-2H-pyridazin-3-one	361 ¹ (361.22)
I-28		6-(4-Ethyl-3-phenoxy-benzyl)-2H-pyridazin-3-one	307 (306.36) oil
I-29		6-[3-(3-Chloro-phenoxy)-4-ethyl-benzyl]-2H-pyridazin-3-one	341 (340.80) oil
I-30		6-[3-(3-Chloro-phenoxy)-4-methyl-benzyl]-2H-pyridazin-3-one	327 (326.78) oil
I-31		6-[4-Chloro-3-(3-chloro-phenoxy)-benzyl]-2H-pyridazin-3-one	347 (347.2) oil
I-32		6-[3-(3-Chloro-phenoxy)-5-ethyl-benzyl]-2H-pyridazin-3-one	340 ¹ 340.80

Table I			
cpd #	Structure	Name	[M+H] ⁺ (mw) mp
I-33		6-[3-(4-Chloro-phenoxy)-5-ethyl-benzyl]-2H-pyridazin-3-one	340 ¹ (340.80)
I-34		6-[3-(2-Chloro-phenoxy)-4-methyl-benzyl]-2H-pyridazin-3-one	327 (326.78) 116-118
I-35		6-[3-(2-Chloro-phenoxy)-4-ethyl-benzyl]-2H-pyridazin-3-one	341 (340.80) oil
I-36		6-[4-Chloro-3-(2-chloro-phenoxy)-benzyl]-2H-pyridazin-3-one	347 (347.2) 132.1-133.7
I-37		6-[3-(3-Methoxy-phenoxy)-4-methyl-benzyl]-2H-pyridazin-3-one	323 (322.36) 66.5-70.7
I-38		6-(4-Chloro-3-phenoxy-benzyl)-2H-pyridazin-3-one	313 (312.75)
I-39		6-[4-Methyl-3-(3-trifluoromethyl-phenoxy)-benzyl]-2H-pyridazin-3-one	361 (360.33) oil
I-40		6-[3-(3-Chloro-phenoxy)-4,5-dimethyl-benzyl]-2H-pyridazin-3-one	341 340.80 121-123

Table I			
cpd #	Structure	Name	[M+H] ⁺ (mw) mp
I-41		6-(4-Methyl-3-m-tolyloxy-benzyl)-2H-pyridazin-3-one	307 (306.36) oil
I-42		6-[3-(3-Bromo-phenoxy)-4-methyl-benzyl]-2H-pyridazin-3-one	371 (371.23) 96.6-101.3
I-43		6-[3-(3-Chloro-phenoxy)-4-isopropyl-benzyl]-2H-pyridazin-3-one	355 (354.83) 104-106
I-44		6-[3-(3-Fluoro-phenoxy)-4-methyl-benzyl]-2H-pyridazin-3-one	311 (310.32) 119.2-121
I-45		6-[3-(3-Bromo-phenoxy)-4-chloro-benzyl]-2H-pyridazin-3-one	391 (391.65) 112.3-112.5
I-46		6-[4-Chloro-3-(4-chloro-phenoxy)-benzyl]-2H-pyridazin-3-one	347 (347.2) 104.4-107.1
I-47		6-[4-Chloro-3-(3-methoxy-phenoxy)-benzyl]-2H-pyridazin-3-one	343 (342.78) foam
I-48		6-[4-Chloro-3-(3,5-difluoro-phenoxy)-benzyl]-2H-pyridazin-3-one	349 (348.73) 91-93

Table I			
cpd #	Structure	Name	[M+H] ⁺ (mw) mp
I-49		6-(4-Chloro-3-phenoxy-benzyl)-4-methyl-2H-pyridazin-3-one	327 (326.78)
I-50		6-(4-Chloro-3-phenoxy-benzyl)-5-methyl-2H-pyridazin-3-one	327 (326.78)
I-51		6-[4-Chloro-3-(3,5-dichloro-phenoxy)-benzyl]-2H-pyridazin-3-one	381 (381.64) 160.9-163.0
I-52		6-[3-(3-Bromo-5-fluoro-phenoxy)-4-chloro-benzyl]-2H-pyridazin-3-one	409 (409.64) 98-101
I-53		3-[2-Chloro-5-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	338 (337.76) 114.5-119.8
I-54		6-[4-Chloro-3-(3-hydroxy-phenoxy)-benzyl]-2H-pyridazin-3-one	329 (328.75) 115-117
I-55		3-[2-Methyl-5-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	318 (317.34) n
I-56		6-[3-(4-Bromo-phenoxy)-4-chloro-benzyl]-2H-pyridazin-3-one	391 (391.65) 139.2-142.8
I-57		6-[4-Chloro-3-(3-fluoro-phenoxy)-benzyl]-2H-pyridazin-3-one	331 (330.74)

Table I			
cpd #	Structure	Name	[M+H] ⁺ (mw) mp
I-58		6-[4-Chloro-3-(3-trifluoromethyl-phenoxy)-benzyl]-2H-pyridazin-3-one	381 (380.75) 105-107
I-59		6-[3-(3-Bromo-phenoxy)-4-chloro-benzyl]-4,5-dimethyl-2H-pyridazin-3-one	419 (419.70)
I-60		6-[3-(3-Chloro-phenoxy)-4-fluoro-benzyl]-2H-pyridazin-3-one	331 (330.74) 141-144
I-61		6-[4-fluoro-3-(3-trifluoromethyl-phenoxy)-benzyl]-2H-pyridazin-3-one	365 (364.29) 120-123
I-62		6-[3-(3-Bromo-phenoxy)-4-fluoro-benzyl]-2H-pyridazin-3-one	375 (375.19) 119-123
I-63		6-[4-Chloro-3-(2-chloro-phenoxy)-benzyl]-4-isopropyl-2H-pyridazin-3-one	389 (389.28)
I-64		6-[4-Chloro-3-(2-chloro-phenoxy)-benzyl]-4-ethyl-2H-pyridazin-3-one	375 (375.25)
I-65		6-[4-Chloro-3-(2-chloro-phenoxy)-benzyl]-4-methyl-2H-pyridazin-3-one	361 (361.22)

Table I			
cpd #	Structure	Name	[M+H] ⁺ (mw) mp
I-66		6-[3-(3-Bromo-phenoxy)-4-chloro-benzyl]-4-methyl-2H-pyridazin-3-one	405 (405.67)
I-67		6-[3-(3-Bromo-phenoxy)-4-chloro-benzyl]-5-methyl-2H-pyridazin-3-one	405 (405.67)
I-68		6-(2,4-Dichloro-3-phenoxy-benzyl)-2H-pyridazin-3-one	(347.2)
I-69		6-(4-Fluoro-3-phenoxy-benzyl)-2H-pyridazin-3-one	297 (296.3) 130-132
I-70		6-[4-Chloro-3-(2,4-dichloro-phenoxy)-benzyl]-2H-pyridazin-3-one	381 (381.64) 88-92
I-71		3-[2-fluoro-5-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	322 (321.31) 169.2-171.3
I-72		6-[3-(2-Chloro-phenoxy)-4-fluoro-benzyl]-2H-pyridazin-3-one	331 (330.74) 137.5-139
I-73		6-[4-Chloro-3-(3,5-dichloro-phenoxy)-benzyl]-4-methyl-2H-pyridazin-3-one	(395.67)
I-74		6-[4-Chloro-3-(3,5-dichloro-phenoxy)-benzyl]-5-methyl-2H-pyridazin-3-one	(395.67)

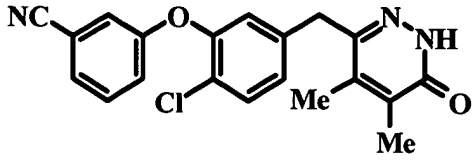
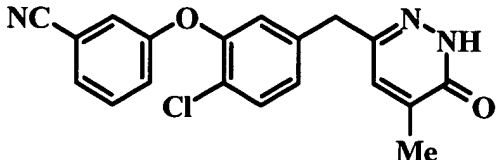
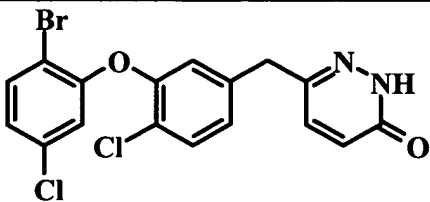
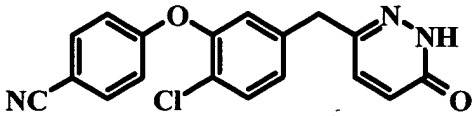
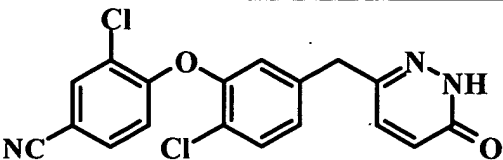
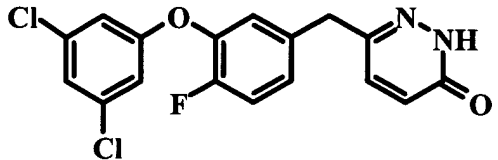
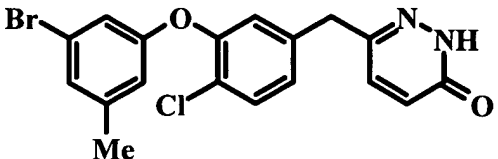
Table I			
cpd #	Structure	Name	[M+H] ⁺ (mw) mp
I-75		3-[2-Chloro-5-(4,5-dimethyl-6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	366 (365.81)
I-76		3-[2-Chloro-5-(4-methyl-6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	352 (351.79)
I-77		6-[3-(2-Bromo-5-chloro-phenoxy)-4-chloro-benzyl]-2H-pyridazin-3-one	425 (426.09) 152.7-155.1
I-78		4-[2-Chloro-5-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	338 (337.76) 193.4-201.4
I-79		3-Chloro-4-[2-chloro-5-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	372 (372.21) 180.7-182.1
I-80		6-[3-(3,5-Dichloro-phenoxy)-4-fluoro-benzyl]-2H-pyridazin-3-one	365 (365.19) 129.1-132.1
I-81		6-[3-(3-Bromo-5-methyl-phenoxy)-4-chloro-benzyl]-2H-pyridazin-3-one	405 (405.67) 155.1-157.4

Table I			
cpd #	Structure	Name	[M+H] ⁺ (mw) mp
I-82		6-[3-(3-Bromo-phenoxy)-4-chloro-benzyl]-4-chloromethyl-2H-pyridazin-3-one	441 (440.12)
I-83		6-[3-(3-Bromo-phenoxy)-4-chloro-benzyl]-4-hydroxymethyl-2H-pyridazin-3-one	423 (421.67)
I-84		6-(4-Chloro-2-methyl-3-phenoxy-benzyl)-2H-pyridazin-3-one	326 (326.78)
I-85		6-[3-(3-Bromo-phenoxy)-4-chloro-2-methyl-benzyl]-2H-pyridazin-3-one	405 (405.67)
I-86		4-[2-Chloro-5-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-3-methyl-benzonitrile	352 (351.79) 146.1-150.2
I-87		3-[2-Chloro-5-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-5-methyl-benzonitrile	352 (351.79)
I-88		6-[3-(3-Bromo-phenoxy)-4-chloro-benzyl]-5-dimethylamino-2H-pyridazin-3-one	436 (434.71)
I-89		6-[3-(3-Bromo-phenoxy)-4-chloro-benzyl]-4-dimethylamino-2H-pyridazin-3-one	436 (434.71)

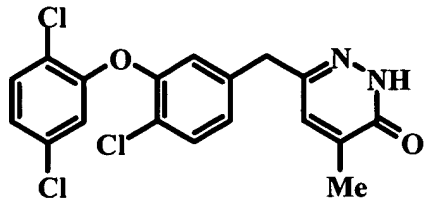
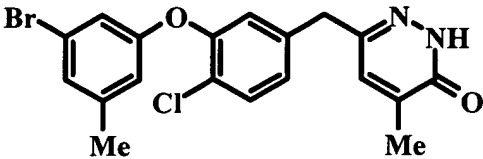
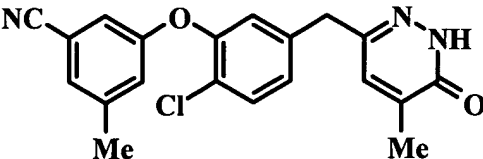
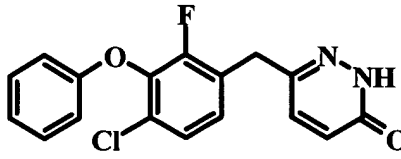
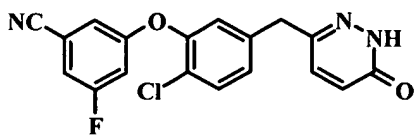
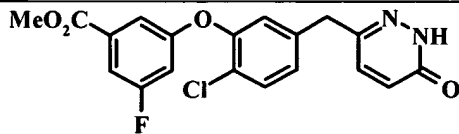
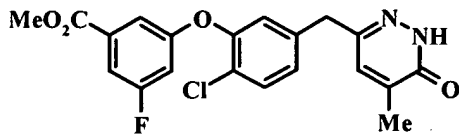
Table I			
cpd #	Structure	Name	[M+H] ⁺ (mw) mp
I-90		6-[4-Chloro-3-(2,5-dichloro-phenoxy)-benzyl]-4-methyl-2H-pyridazin-3-one	395 (395.67) 195-198
I-91		6-[3-(3-Bromo-5-methyl-phenoxy)-4-chloro-benzyl]-4-methyl-2H-pyridazin-3-one	419 (419.70) 121.7-133
I-92			366 (365.81) 177.5-180
I-93		6-(4-Chloro-2-fluoro-3-phenoxy-benzyl)-2H-pyridazin-3-one	330 (330.74)
I-94		3-[2-Chloro-5-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-5-fluoro-benzonitrile	(355.75)
I-95		3-[2-Chloro-5-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-5-fluoro-benzoic acid methyl ester	(388.78)
I-96		3-[2-Chloro-5-(5-methyl-6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-5-fluoro-benzoic acid methyl ester	(402.81)

Table I			
cpd #	Structure	Name	[M+H] ⁺ (mw) mp
I-97		6-(4-Bromo-3-phenoxy-benzyl)-2H-pyridazin-3-one	357 (357.20)
I-98		3-Chloro-5-[2-ethyl-5-(5-methyl-6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	(379.84)
I-99		3-Chloro-5-[2-ethyl-5-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	(365.82)
I-100		3-[2-Ethyl-5-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-5-methyl-benzonitrile	(345.40)
I-101		3-[2-Ethyl-5-(5-methyl-6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-5-fluoro-benzonitrile	(363.39)
I-102		6-[4-Chloro-3-(2,5-dichloro-phenoxy)-benzyl]-2H-pyridazin-3-one	(381.64)
I-103		6-[3-(2-Bromo-5-chloro-phenoxy)-4-chloro-benzyl]-4-methyl-2H-pyridazin-3-one	(440.12)

Table I			
cpd #	Structure	Name	[M+H] ⁺ (mw) mp
I-104		6-[4-Bromo-3-(3-chloro-phenoxy)-benzyl]-2H-pyridazin-3-one	391 (391.65)
I-105		6-(4-Bromo-3-phenoxy-benzyl)-4-methyl-2H-pyridazin-3-one	371 (371.23)
I-106		3-[2-Ethyl-5-(5-methyl-6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-5-fluoro-benzonitrile	(349.36)
I-107		3-[2-Chloro-5-(5-methyl-6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-5-fluoro-benzonitrile	(369.78)
I-108		3-Chloro-5-[2-chloro-5-(5-methyl-6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	(386.24)
I-109		3-[2-Ethyl-5-(5-methyl-6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-5-methyl-benzonitrile	(359.43)
I-110		6-[4-Chloro-3-(3-chloro-phenoxy)-benzyl]-5-methylamino-2H-pyridazin-3-one	376 (376.24)

Table I			
cpd #	Structure	Name	[M+H] ⁺ (mw) mp
I-111		6-[4-Chloro-3-(3-chloro-phenoxy)-benzyl]-4-methylamino-2H-pyridazin-3-one	376 (376.24)
I-112		6-[4-Chloro-3-(3-chloro-phenoxy)-benzyl]-4-morpholin-4-yl-2H-pyridazin-3-one	432 (432.31)
I-113		2-[2-Chloro-5-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-3-fluoro-benzonitrile	355 (355.76)
I-114		3-Fluoro-5-[2-methyl-5-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	
I-115		3-Bromo-4-[2-chloro-5-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	416 (466.66)
I-116		3-Chloro-2-[2-chloro-5-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	372 (372.21)
I-117		6-[4-Chloro-3-(2,6-dichloro-phenoxy)-benzyl]-2H-pyridazin-3-one	381 (381.65)

Table I			
cpd #	Structure	Name	[M+H] ⁺ (mw) mp
I-118		6-[4-Chloro-3-(2-chloro-6-fluorophenoxy)-benzyl]-2H-pyridazin-3-one	365 (365.19)
I-119		4-[2-Chloro-5-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-isophthalonitrile	363 (362.78)
I-120		4-Chloro-3-[2-chloro-5-(5-methyl-6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	
I-121		3-Chloro-5-[2-methyl-5-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	
I-122		4-Chloro-3-[2-chloro-5-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	
I-123		6-[4-Chloro-3-(5-chloro-2-methylphenoxy)-benzyl]-4-methyl-2H-pyridazin-3-one	
I-124		3-Methyl-5-[2-methyl-5-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	

Table I			
cpd #	Structure	Name	[M+H] ⁺ (mw) mp
I-125		6-[4-Chloro-3-(5-chloro-2-methyl-phenoxy)-benzyl]-2H-pyridazin-3-one	
I-126		3-Chloro-5-[2-chloro-5-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	
I-127		3-Fluoro-5-[2-methyl-5-(4-methyl-6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	
I-128		3-Fluoro-5-[2-methyl-5-(5-methyl-6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	
I-129		3-Fluoro-5-[2-methyl-5-(5-methyl-6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzoic acid methyl ester	
I-130		6-[4-Chloro-3-(3-chloro-phenoxy)-benzyl]-4-[(2-hydroxy-ethyl)-methyl-amino]-2H-pyridazin-3-one	420 (420.30)
I-131		Carbamic acid 2-({6-[4-chloro-3-(3-chloro-phenoxy)-benzyl]-3-oxo-2,3-dihydro-pyridazin-4-yl}-methyl-amino)-ethyl ester	463 (463.32)

Table I			
cpd #	Structure	Name	[M+H] ⁺ (mw) mp
I-132		3-Bromo-4-[2-chloro-5-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzamide	436 (434.68)
I-133		6-[4-Chloro-3-(3,5-dibromo-phenoxy) -benzyl]-2H-pyridazin-3-one	471 (470.55)
I-134		6-[4-Chloro-3-(3,5-dibromo-phenoxy) -benzyl]-4-methyl-2H-pyridazin-3-one	485 (484.58)
I-135		5-[2-Chloro-5-(5-methyl-6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-isophthalonitrile	377 (376.81)
I-136		6-[3-(2-Bromo-5-chloro-phenoxy)-4-chloro-2-fluoro-benzyl]-2H-pyridazin-3-one	445 (444.09)
I-137		5-[2-Chloro-5-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-isophthalonitrile	363 (362.78)
I-138		6-[7-(4-Chloro-phenoxy)-benzofuran-5-ylmethyl]-2H-pyridazin-3-one	352 (352.7) 205-210

Table I			
cpd #	Structure	Name	[M+H] ⁺ (mw) mp
I-139		6-[7-(3,5-Dichloro-phenoxy)-benzofuran-5-ylmethyl]-2H-pyridazin-3-one	386 (387.2)
I-140		6-[7-(3-Chloro-phenoxy)-benzofuran-5-ylmethyl]-2H-pyridazin-3-one	352 ¹ (352.8)
I-141		6-[7-(3,5-Difluoro-phenoxy)-benzofuran-5-ylmethyl]-2H-pyridazin-3-one	355 (354.3) 213.4-215.5
I-142		6-[7-(3,5-Dibromo-phenoxy)-benzofuran-5-ylmethyl]-2H-pyridazin-3-one	477 (476.1) 183.9-185.3
I-143		6-[7-(3,5-Dichloro-phenoxy)-2,3-dihydro-benzofuran-5-ylmethyl]-2H-pyridazin-3-one	389 (389.2) 216.5-218.8
I-144		6-(7-Phenoxy-2,3-dihydro-benzofuran-5-ylmethyl)-2H-pyridazin-3-one	321 (320.4) 133.7-135.8
I-145		4-[5-(6-Oxo-1,6-dihydro-pyridazin-3-ylmethyl)-benzofuran-7-yloxy]-benzonitrile	342 ² (343.3) 174-175
I-146		6-[7-(3,5-Dimethyl-phenoxy)-benzofuran-5-ylmethyl]-2H-pyridazin-3-one	347 (346.4) 56-60

Table I			
cpd #	Structure	Name	[M+H] ⁺ (mw) mp
I-147		3-[5-(6-Oxo-1,6-dihydro-pyridazin-3-ylmethyl)-benzofuran-7-yloxy]-benzonitrile	344 (343.3) 171.0-173.1
I-148		6-[2-Chloro-5-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-pyridine-2-carbonitrile	(338.76)
I-149		4-Chloro-2-[2-chloro-5-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	372 (372.2) 228.0-229.6
I-150		6-[3-(3-Bromo-phenoxy)-4-chloro-2-fluoro-benzyl]-2H-pyridazin-3-one	411 (409.6)
I-151		4-Chloro-2-[6-chloro-2-fluoro-3-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	390 (390.2)
I-152		2-[6-Chloro-2-fluoro-3-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-terephthalonitrile	381 (380.77)
I-153		3-[6-Chloro-2-fluoro-3-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-5-fluorobenzonitrile	374 (373.75)
I-154		4-Chloro-2-[2-chloro-5-(5-methyl-6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	386 (386.2) 215.1-215.9

Table I			
cpd #	Structure	Name	[M+H] ⁺ (mw) mp
I-155		6-(4-Chloro-2-fluoro-3-phenoxy-benzyl)-4-methyl-2H-pyridazin-3-one	345.16 (344.78)
I-156		6-[4-Chloro-3-(3,5-dichloro-phenoxy)-2-fluoro-benzyl]-4-methyl-2H-pyridazin-3-one	414.97 (413.67)
I-157		3-[6-Chloro-2-fluoro-3-(5-methyl-6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-5-fluoro-benzonitrile	388 (387.78)
I-158		3-[2-Chloro-5-(5-dimethylamino-6-oxo-1,6-dihydro-pyridazin-3-ylmethyl) -phenoxy]-benzonitrile	381 (380.84)
I-159		6-[3-(5-Bromo-2-chloro-phenoxy)-4-chloro-2-fluoro-benzyl]-2H-pyridazin-3-one	445 (444.09)
I-160		6-(4-Bromo-2-fluoro-3-phenoxy-benzyl)-2H-pyridazin-3-one	375(375.2)
I-161		6-[4-Bromo-3-(3-bromo-phenoxy)-2-fluoro-benzyl]-2H-pyridazin-3-one	455 (454.10)
I-162		4-Chloro-3-[6-chloro-2-fluoro-3-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	390 390.20
I-163		4-Bromo-2-[2-chloro-5-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	418 416.66

Table I			
cpd #	Structure	Name	[M+H] ⁺ (mw) mp
I-164		3-[6-Chloro-2-fluoro-3-(5-methyl-6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	370 (369.79)
I-165		4-Bromo-2-[2-chloro-5-(5-methyl-6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	432 (430.69)
I-166		2-[2-Chloro-5-(5-methyl-6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-terephthalonitrile	377 (376.81)
I-167		3-[2-Bromo-5-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	382 (382.22)
I-168		2-[2-Chloro-5-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-terephthalonitrile	363 (362.78)
I-169		6-[4-Bromo-3-(2-chloro-phenoxy)-benzyl]-4-methyl-2H-pyridazin-3-one	405 (405.68) foam
I-170		3-[6-Bromo-2-fluoro-3-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-5-fluorobenzonitrile	418 (418.20) 185.2-186.7
I-171		2-Chloro-4-[2-methyl-5-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	352 (351.80) 184.0-184.8
I-172		5-[6-Chloro-2-fluoro-3-(5-methyl-6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-isophthalonitrile	395 (394.80) 230.1-231.9

Table I			
cpd #	Structure	Name	[M+H] ⁺ (mw) mp
I-173		5-[6-Chloro-2-fluoro-3-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-isophthalonitrile	381 (380.77)
I-174		2-Chloro-6-[2-chloro-5-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	372 (372.21) 213.9-215.5
I-175		2-Chloro-6-[2-chloro-5-(5-methyl-6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	386 (386.24) 220.2-221.9
I-176		4-Chloro-2-[2-chloro-5-(5-methyl-6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	366 (365.82) 201.2-202.3
I-177		6-[4-Chloro-3-(2-methanesulfonyl-phenoxy)-benzyl]-2H-pyridazin-3-one	391 (390.85)
I-178		N-{2-[2-Chloro-5-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-phenyl}-acetamide	370 (369.81) 190.0-190.4
I-179		N-{2-[2-Chloro-5-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-phenyl}-methanesulfonamide	406 (405.86)
I-180		6-[4-Chloro-3-(2-nitro-phenoxy)-benzyl]-2H-pyridazin-3-one	358 (357.76)
I-181		2-[2-Chloro-5-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	338 (337.77) 173.0-174.5

Table I			
cpd #	Structure	Name	[M+H] ⁺ (mw) mp
I-182		2-[2-Chloro-5-(5-methyl-6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-terephthalonitrile	(394.80)
I-183		6-[3-(3-Chloro-phenoxy)-4-trifluoromethyl-benzyl]-2H-pyridazin-3-one	(380.76)
I-184		4-Chloro-3-[6-chloro-2-fluoro-3-(5-methyl-6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	403 (404.23) 225.9-228.9
I-185		6-[3-(3-Chloro-phenoxy)-4-trifluoromethyl-benzyl]-4-methyl-2H-pyridazin-3-one	(394.78)
I-186		6-[3-(2-Amino-phenoxy)-4-chloro-benzyl]-2H-pyridazin-3-one; compound with hydrochloric acid	328 (364.23) 240.1-242.7
I-187		4-Chloro-2-[2-chloro-5-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	352 (351.80) 205.9-208.9
I-188		3-Fluoro-5-[2-fluoro-6-methyl-3-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	354 (353.33) 170.1-172.0
I-189		4-[6-Chloro-2-fluoro-3-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	356 355.76
I-190		5-[6-Chloro-2-fluoro-3-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-isophthalonitrile	343 342.36

Table I			
cpd #	Structure	Name	[M+H] ⁺ (mw) mp
I-191		3-[2-Methyl-5-(5-methyl-6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	332 331.38
I-192		3-[6-Bromo-2-fluoro-3-(5-methyl-6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	414 (414.24)
I-193		3-[2,3-Dibromo-6-fluoro-5-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	480 (479.11)
I-194		3-[2-Chloro-5-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-4-trifluoromethyl-benzonitrile	406 (405.77 195-196.8
I-195		3-[2-Fluoro-6-methyl-3-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	335 (335.34) 139.2-144.9
I-196		3-[2-Fluoro-6-methyl-3-(5-methyl-6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	349 (349.37) 175.0-175.7
I-197		6-[3-(2-Chloro-phenoxy)-4-trifluoromethyl-benzyl]-2H-pyridazin-3-one	(380.76)
I-198		6-[3-(4-Chloro-phenoxy)-4-trifluoromethyl-benzyl]-2H-pyridazin-3-one	(380.76)
I-199		6-(3-Phenoxy-4-trifluoromethyl-benzyl)-2H-pyridazin-3-one	(346.31)

Table I			
cpd #	Structure	Name	[M+H] ⁺ (mw) mp
I-200		3-[6-Bromo-2-fluoro-3-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	402 (400.21)
I-201		4-Methyl-6-(3-phenoxy-4-trifluoromethyl-benzyl)-2H-pyridazin-3-one	(360.34)
I-202		6-[3-(4-Chloro-phenoxy)-4-trifluoromethyl-benzyl]-4-methyl-2H-pyridazin-3-one	(394.78)
I-203		3-[2-Bromo-5-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-4-chloro-benzonitrile	(416.66) 196.8-199.9
I-204		3-[2-Bromo-5-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-4-chloro-benzonitrile	352 (351.80) 189.9-191.0
I-205		3-[2-Bromo-5-(5-methyl-6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-4-chloro-benzonitrile	432 (430.69) 179.5-197.7
I-206		3-[2-Bromo-5-(5-methyl-6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-4-chloro-benzonitrile	366 (365.82) 188.7-189.7
I-207		3-[2-Chloro-5-(5-methyl-6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-phthalonitrile	377 (376.81) 201.2-204.1

Table I			
cpd #	Structure	Name	[M+H] ⁺ (mw) mp
I-208		2-[5-(6-Oxo-1,6-dihydro-pyridazin-3-ylmethyl)-2-trifluoromethyl-phenoxy]-benzonitrile	372 (371.32) 183.3-184.9
I-209		3-[6-Bromo-2-fluoro-3-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-4-chloro-benzonitrile	433 (464.65) 224.2-224.8
I-210		3-[6-Bromo-2-fluoro-3-(5-methyl-6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-4-chloro-benzonitrile	447 (448.68) 237.2-238.0
I-211		4-Chloro-3-[6-ethyl-2-fluoro-3-(5-methyl-6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxyl]-benzonitrile	397 (397.84)
I-212		4-Chloro-3-[6-ethyl-2-fluoro-3-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxyl]-benzonitrile	384 (383.81)
I-213		3-Chloro-5-[6-chloro-2-fluoro-3-(5-methyl-6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxyl]-benzonitrile	404.1 (404.23) 186.1-185.4
I-214		3-[6-Chloro-2-fluoro-3-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxyl]-5-methyl-benzonitrile	370 (369.79) 190-192.5
I-215		3-[6-Chloro-2-fluoro-3-(5-methyl-6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxyl]-5-methyl-benzonitrile	384 (383.81) 202.4-203.7

Table I			
cpd #	Structure	Name	[M+H] ⁺ (mw) mp
I-216		4-Chloro-3-[2-fluoro-6-methyl-3-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	370 (369.79) 195.8-197.7
I-217		3-[2-Fluoro-6-methyl-3-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-4-methyl-benzonitrile	350 (349.37) 202.8-204.1
I-218		3-[2-Fluoro-6-methyl-3-(5-methyl-6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-4-methyl-benzonitrile	364 (363.39) 204-205.5
I-219		4-Chloro-3-[2-fluoro-6-methyl-3-(5-methyl-6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	384 (383.81) 241.9-247.4
I-220		3-[5-(6-Oxo-1,6-dihydro-pyridazin-3-ylmethyl)-2-trifluoromethyl-phenoxy]-benzonitrile	372 (371.32) foam
I-221		3-[6-Bromo-3-(5-dimethylamino-6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-2-fluoro-phenoxy]-4-chloro-benzonitrile	477 (477.72) 249.2-251.3
I-222		6-[4-Chloro-2-fluoro-3-(1H-indol-4-yloxy)-benzyl]-2H-pyridazin-3-one	(369.79)
I-223		6-[4-Chloro-2-fluoro-3-(quinolin-5-yloxy)-benzyl]-4-methyl-2H-pyridazin-3-one	370.1 (395.82) 194.5-196.3

Table I			
cpd #	Structure	Name	[M+H] ⁺ (mw) mp
I-224		3-Chloro-5-[6-ethyl-2-fluoro-3-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	384 (383.81)
I-225		3-[2-Fluoro-6-methyl-3-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-5-methyl-benzonitrile	350 (349.37) 180.1-181.9
I-226		6-[4-Chloro-2-fluoro-3-(quinolin-8-yloxy)-benzyl]-4-methyl-2H-pyridazin-3-one	396 (395.82) 239.7-242
I-227		6-[4-Chloro-2-fluoro-3-(1-oxy-quinolin-8-yloxy)-benzyl]-4-methyl-2H-pyridazin-3-one	412 (411.82) 118-120.9
I-228		6-[3-(5-Bromo-2-methanesulfonyl-phenoxy)-4-chloro-benzyl]-2H-pyridazin-3-one	469 (469.74) 242.0-243.8
I-229		3-[2-Chloro-5-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-4-methanesulfonyl-benzonitrile	416 (415.86) 258.3-259.5
I-230		3-[6-Bromo-2-fluoro-3-(5-methyl-6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-5-chloro-benzonitrile	448 (448.68) 155-158.5
I-231		3-[6-Bromo-2-fluoro-3-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-5-chloro-benzonitrile	434 (434.65) 197.8-199.4

Table I			
cpd #	Structure	Name	[M+H] ⁺ (mw) mp
I-232		3-Chloro-5-[2-fluoro-6-methyl-3-(5-methyl-6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	384 (383.81) 202.5-206.2
I-233		6-(7-Phenoxy-benzofuran-5-ylmethyl)-2H-pyridazin-3-one	319 (318.3) 116.4-117.7
I-234		6-[4-Chloro-2-fluoro-3-(1H-indol-4-yloxy)-benzyl]-4-methyl-2H-pyridazin-3-one	383 (383.81) 255.6-257.6
I-235		6-[4-Chloro-2-fluoro-3-(1H-indol-7-yloxy)-benzyl]-4-methyl-2H-pyridazin-3-one	383 (383.81) 201.7-203.7
I-236		3-Chloro-5-[6-chloro-2-fluoro-3-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile	391 (390.20) 206-208
I-237		3-[6-Chloro-2-fluoro-3-(5-methyl-6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-5-difluoromethyl-benzonitrile	420 (419.79) 179.2-181.2
I-238		3-[6-Chloro-2-fluoro-3-(4-methyl-6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-5-difluoromethyl-benzonitrile	(419.79)
I-239		3-[6-Chloro-2-fluoro-3-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-5-ethyl-benzonitrile	383 (383.81) 140-151

Table I			
cpd #	Structure	Name	[M+H] ⁺ (mw) mp
I-240		3-[6-Chloro-2-fluoro-3-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-5-ethenyl-benzonitrile	381 (381.79) 170.2-173.4
I-241		2-(3-Cyano-phenoxy)-3-fluoro-4-(5-methyl-6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-benzonitrile	361 (360.35) 191.8-192.4
I-242		6-[4-Chloro-3-(3-chloro-phenoxy)-2-fluoro-benzyl]-4-ethyl-2H-pyridazin-3-one	(375.23)
I-243		6-[4-Chloro-5-(3-chloro-phenoxy)-2-fluoro-benzyl]-4-methyl-2H-pyridazin-3-one	(379.22)
I-244		6-[4-Chloro-5-(3-chloro-phenoxy)-2-fluoro-benzyl]-2H-pyridazin-3-one	(365.19)
I-245		5-[6-Ethyl-2-fluoro-3-(5-methyl-6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-isophthalonitrile	389 (388.40)
I-246		2-[2-Chloro-5-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-isonicotinonitrile	(338.76)
I-247		6-[3-(6-Chloro-pyridin-2-yloxy)-4-methyl-benzyl]-4-methyl-2H-pyridazin-3-one	342 (341.80)
I-248		6-[3-(6-Chloro-pyridin-2-yloxy)-4-methyl-benzyl]-5-methyl-2H-pyridazin-3-one	342 (341.80) 173.0-176.4

Table I			
cpd #	Structure	Name	[M+H] ⁺ (mw) mp
I-249		6-[4-Chloro-3-(4-methyl-pyridin-2-yloxy)-benzyl]-4-methyl-2H-pyridazin-3-one	342 (341.80) 188.0-188.9
I-250		6-[4-Chloro-3-(4-methyl-1-oxy-pyridin-2-yloxy)-benzyl]-4-methyl-2H-pyridazin-3-one	358 (357.80) 195.6-197.3
I-251		6-[2-Chloro-5-(4-methyl-6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-4-methyl-pyridine-2-carbonitrile	367 (366.81) 225-228.1
I-252		6-[2-Chloro-5-(5-methyl-6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-4-methyl-pyridine-2-carbonitrile	(366.81) 249.9-253.6
I-253		2-[2-Chloro-5-(5-methyl-6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-6-methyl-isonicotinonitrile	366 (366.81) 185-187.3
I-254		6-[2-Fluoro-4-methyl-3-(pyridin-3-yloxy)-benzyl]-4-methyl-2H-pyridazin-3-one	326 (325.35) 164.39-167.7
I-255		6-[2-Fluoro-4-methyl-3-(1-oxy-pyridin-3-yloxy)-benzyl]-4-methyl-2H-pyridazin-3-one	342 (341.34) 207.8-212.7
I-256		6-[3-(5-Bromo-pyridin-3-yloxy)-4-chloro-2-fluoro-benzyl]-4-methyl-2H-pyridazin-3-one	424 (424.66) 184.9-188
I-257		6-[3-(5-Bromo-1-oxy-pyridin-3-yloxy)-4-chloro-2-fluoro-benzyl]-4-methyl-2H-pyridazin-3-one	(440.66) 197.9-198.3

Table I			
cpd #	Structure	Name	[M+H] ⁺ (mw) mp
I-258		6-[3-(3-Bromo-5-cyclopropyl-phenoxy)-4-chloro-2-fluoro-benzyl]-2H-pyridazin-3-one	449 (449.7) 16536-167.7
I-259		3-[6-Chloro-2-fluoro-3-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-5-cyclopropyl-benzonitrile	395 (395.8) 177.4-180.5
I-260		6-[4-Chloro-2-fluoro-3-(2,3,5-trichloro-phenoxy)-benzyl]-2H-pyridazin-3-one	(434) 187.5-189.5
			¹ . [M] ⁺ ² . [M-H] ⁺

Table II			
cpd #	Structure	Name	[M+H] ⁺ (mw) mp
II-1		6-[3-(3-chloro-phenylsulfanyl)-4-nitro-benzyl]-2H-pyridazin-3-one	373.82
II-2		2-(2-Chloro-phenylsulfanyl)-4-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl) - phenyl-ammonium; chloride	380.30
II-3		2-(3-Chloro-phenylsulfanyl)-4-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl) - phenyl-ammonium; chloride	380.30

Table II			
cpd #	Structure	Name	[M+H] ⁺ (mw) mp
II-4		6-(7-Benzyl-benzofuran-5-ylmethyl)-2H-pyridazin-3-one	317 (316.36)
II-5		6-[7-(2-Fluoro-benzyl)-benzofuran-5-ylmethyl]-2H-pyridazin-3-one	335 334.35
II-6		6-[7-(2,6-Difluoro-benzyl)-benzofuran-5-ylmethyl]-2H-pyridazin-3-one	[M] ⁺ =352 352.34
II-7		6-[7-(2-Methoxy-benzyl)-benzofuran-5-ylmethyl]-2H-pyridazin-3-one	347 346.39
II-8		6-(3-Benzyl-4-methyl-benzyl)-2H-pyridazin-3-one	291 290.37
II-9		6-[3-(4-Chloro-benzyl)-4-methyl-benzyl]-2H-pyridazin-3-one	325 324.81
II-10		6-[7-(3,5-Dichloro-benzyl)-benzofuran-5-ylmethyl]-2H-pyridazin-3-one	[M] ⁺ =385 385.25
II-11		6-[3-(2-Chloro-benzyl)-4-methyl-benzyl]-2H-pyridazin-3-one	325 324.81
II-12		6-(3-Benzyl-4-chloro-benzyl)-2H-pyridazin-3-one	311 310.79
II-13		6-[3-(3-Chloro-benzyl)-4-methyl-benzyl]-2H-pyridazin-3-one	325 324.81

Table II			
cpd #	Structure	Name	[M+H] ⁺ (mw) mp
II-14		6-[3-(3,5-Dichloro-benzyl)-4-methyl-benzyl]-2H-pyridazin-3-one	[M] ⁺ =359 359.26
II-15		6-[7-(4-Bromo-benzyl)-benzofuran-5-ylmethyl]-2H-pyridazin-3-one	[M] ⁺ =395 395.26
II-16		6-[4-(4-Chloro-benzyl)-naphthalen-2-ylmethyl]-2H-pyridazin-3-one	361 360.85
II-17		6-[3-(3-Chloro-benzyloxy)-benzyl]-2H-pyridazin-3-one	326 ¹ (326.78)
II-18		6-[3-(2-Chloro-benzyloxy)-benzyl]-2H-pyridazin-3-one	326 ¹ (326.78)
II-19		6-[3-(4-Chloro-benzyloxy)-benzyl]-2H-pyridazin-3-one	326 ¹ (326.78)
II-20		6-(7-Isobutoxy-benzofuran-5-ylmethyl)-2H-pyridazin-3-one	299 (298.34) 99.7-100.9
II-21		6-(7-Butoxy-benzofuran-5-ylmethyl)-2H-pyridazin-3-one	299 (298.34) 78.5-82.5
II-22		6-[7-(1-Ethyl-propoxy)-benzofuran-5-ylmethyl]-2H-pyridazin-3-one	313 (312.36)

Table II			
cpd #	Structure	Name	[M+H] ⁺ (mw) mp
II-23		6-(3-Butoxy-4-chloro-benzyl)-2H-pyridazin-3-one	293 (292.76) oil
II-24		6-[4-Chloro-3-(1-ethyl-propoxy)-benzyl]-2H-pyridazin-3-one	307 (306.79) 103.9-105.2
II-25		6-(3-Cyclopropylmethoxy-4-ethyl-2-fluoro-benzyl)-2H-pyridazin-3-one	302 (302.34)
II-26		6-[3-(3-Bromo-phenylsulfanyl)-4-chloro-benzyl]-2H-pyridazin-3-one	409 (407.72)
II-27		6-[3-(3-Bromo-benzenesulfonyl)-4-chloro-benzyl]-2H-pyridazin-3-one	439 (439.72)
II-28		6-[3-(3-Bromo-benzenesulfinyl)-4-chloro-benzyl]-2H-pyridazin-3-one	425 (423.72)

Table III			
cpd#	Structure	Name	[M+H] ⁺ (mw) mp
III-1		6-[7-(4-Methylsulfanyl-benzoyl)-benzofuran-5-ylmethyl]-2H-pyridazin-3-one	376(M) ⁺ (376.44) 93-96°
III-2		6-[7-(4-Chloro-benzoyl)-2,3-dihydro-benzofuran-5-ylmethyl]-2H-pyridazin-3-one	366 (366.81) 196-200°

Table III			
cpd#	Structure	Name	[M+H] ⁺ (mw) mp
III-3		6-[7-(4-Chloro-benzoyl)-benzofuran-5-ylmethyl]-2H-pyridazin-3-one	364(M) ⁺ (364.79) 197.5-148.7°
III-4		6-(4-Benzoyl-naphthalen-2-ylmethyl)-2H-pyridazin-3-one	(340.38) 175.6-175.9°
III-5		6-[2,4-Difluoro-3-(4-methyl-benzoyl)-benzyl]-2H-pyridazin-3-one	(340.33)
III-6		6-[4-Methoxy-3-(4-methyl-benzoyl)-benzyl]-2H-pyridazin-3-one	(334.38)
III-7		6-[7-(2-Chloro-benzoyl)-benzofuran-5-ylmethyl]-2H-pyridazin-3-one	364 (M) ⁺ (364.79) 196-198°
III-8		6-(7-Benzoyl-benzofuran-5-ylmethyl)-2H-pyridazin-3-one	330 (M) ⁺ (330.35) 177-181°
III-9		6-[7-(2,4-Dichloro-benzoyl)-benzofuran-5-ylmethyl]-2H-pyridazin-3-one	398 (M) ⁺ (399.24) 208-214°
III-10		6-[5-(4-Chloro-benzoyl)-benzofuran-7-ylmethyl]-2H-pyridazin-3-one	364.79
III-11		6-[7-(3-Chloro-benzoyl)-benzofuran-5-ylmethyl]-2H-pyridazin-3-one	364.79

Table III			
cpd#	Structure	Name	[M+H] ⁺ (mw) mp
III-12		6-[7-(2-Fluoro-benzoyl)-benzofuran-5-ylmethyl]-2H-pyridazin-3-one	348.34
III-13		6-[7-(2,6-Difluoro-benzoyl)-benzofuran-5-ylmethyl]-2H-pyridazin-3-one	366.33
III-14		6-[7-(3-Methoxy-benzoyl)-benzofuran-5-ylmethyl]-2H-pyridazin-3-one	360.373
III-15		6-[7-(4-Chloro-benzoyl)-benzo[b]thiophen-5-ylmethyl]-2H-pyridazin-3-one	380.86
III-16		6-[4-(4-Chloro-benzoyl)-naphthalen-2-ylmethyl]-2H-pyridazin-3-one	374.83
III-17		6-(3-Benzoyl-4-methyl-benzyl)-2H-pyridazin-3-one	304.35
III-18		6-[3-(4-Chloro-benzoyl)-4-methyl-benzyl]-2H-pyridazin-3-one	338.80
III-19		6-[3-(3,5-Dichloro-benzoyl)-4-methyl-benzyl]-2H-pyridazin-3-one	373.24

Table III			
cpd#	Structure	Name	[M+H] ⁺ (mw) mp
III-20		6-[7-(3,5-Dichloro-benzoyl)-benzofuran-5-ylmethyl]-2H-pyridazin-3-one	399.24
III-21		6-(3-Benzoyl-4-chloro-benzyl)-2H-pyridazin-3-one	324.77
III-22		6-[3-(3-Chloro-benzoyl)-4-methyl-benzyl]-2H-pyridazin-3-one	338.80
III-23		6-[7-(4-Bromo-benzoyl)-benzofuran-5-ylmethyl]-2H-pyridazin-3-one	409.24
III-24		4-[5-(6-Oxo-1,6-dihydro-pyridazin-3-ylmethyl)-benzofuran-7-carbonyl]-benzonitrile	355.36

PREPARATION OF COMPOUNDS OF THE INVENTION

Compounds of the present invention can be made by a variety of methods depicted in the illustrative synthetic reaction schemes shown and described below. The starting materials and reagents used in preparing these compounds generally are either available from commercial suppliers, such as Aldrich Chemical Co., or are prepared by methods known to those skilled in the art following procedures set forth in references such as *Fieser and Fieser's Reagents for Organic Synthesis*; Wiley & Sons: New York, Volumes 1-21; R. C. LaRock, *Comprehensive Organic Transformations*, 2nd edition Wiley-VCH, New York 1999; *Comprehensive Organic Synthesis*, B. Trost and I. Fleming (Eds.) vol. 1-9 Pergamon, Oxford, 1991; *Comprehensive Heterocyclic Chemistry*, A. R. Katritzky and C. W. Rees (Eds) Pergamon, Oxford 1984, vol. 1-9; *Comprehensive Heterocyclic Chemistry II*, A. R. Katritzky and C. W. Rees (Eds) Pergamon, Oxford 1996, vol. 1-11; and *Organic Reactions*, Wiley & Sons: New York, 1991, Volumes 1-40. The following synthetic reaction schemes and examples are merely illustrative of some methods by which the compounds of the present invention can be synthesized, and various

modifications to these synthetic reaction schemes can be made and will be suggested to one skilled in the art having referred to the disclosure contained in this Application.

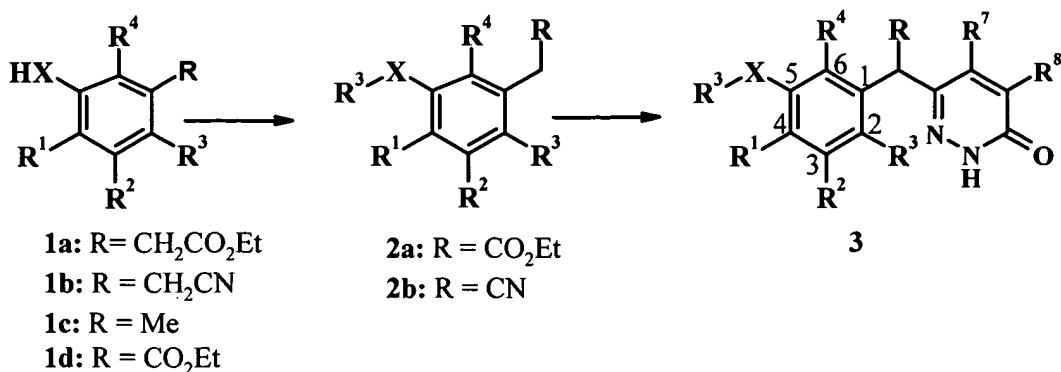
5 The starting materials and the intermediates of the synthetic reaction schemes can be isolated and purified if desired using conventional techniques, including but not limited to, filtration, distillation, crystallization, chromatography, and the like. Such materials can be characterized using conventional means, including physical constants and spectral data.

10 Unless specified to the contrary, the reactions described herein preferably are conducted under an inert atmosphere at atmospheric pressure at a reaction temperature range of from about -78 °C to about 150 °C, more preferably from about 0°C to about 125 °C, and most preferably and conveniently at about room (or ambient) temperature, e.g., about 20 °C. Moreover, the reaction conditions are exemplary and alternative conditions are well known. The reaction sequences in the following examples are not meant to limit the scope of the invention as set forth in the claims.

15 The heterocyclic compounds of the present invention are prepared by a two-stage process (Scheme 1) comprising construction of an appropriately substituted aryl ring **2** and subsequently introducing the heterocyclic ring to produce **3**. Although stages can be accomplished in any order, the heterocyclic ring is generally introduced after the modifications of the aryl ring are completed. Substituted alkyl *m*-hydroxyphenylacetate **1a** or *m*-hydroxyphenylacetonitrile **1b** derivatives are convenient starting materials. They are often commercially available or readily prepared from commercially available precursors. Alternatively the aryl ring may be substituted with a methyl **1c** or carboxylic acid ester **1d** substituent which is subsequently converted to **1b** (for example, see schemes 4 and 5). One skilled in the art will also appreciate the substituents can altered after introduction of the heterocyclic ring.

25

SCHEME 1

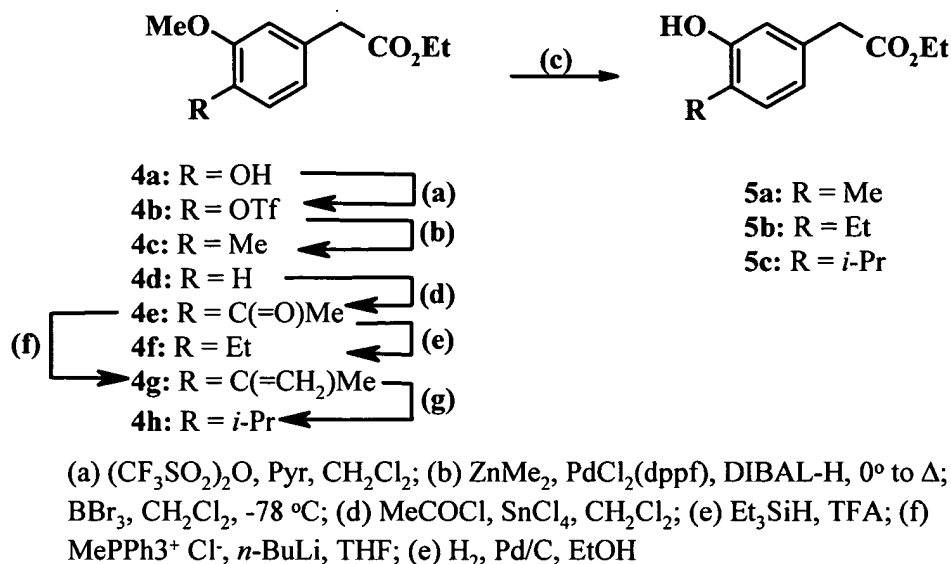


Preparation of Phenylacetic Acid and Phenylacetone Precursors (Scheme 1; 1)

Ethyl 3-hydroxy-4-methylphenylacetate (**5a**) was prepared from ethyl 3-methoxy-4-hydroxy-phenylacetate as shown in Scheme 2. The phenol was converted to the triflate ester **4b** which was subjected to displacement with Me₂Zn, DIBAL-H and PdCl₂(dppf) (E.-i. Negishi in *Metal-catalyzed Cross-Coupling Reactions*, F. Diederich and P. J. Stang (eds.), Wiley-VCH, Mannheim **1998**, chap.1; E. Erdik, *Tetrahedron* **1992** 48:9577-9648) to afford the **4c**. Boron tribromide demethylation afforded **5a**. Ethyl 3-hydroxy-4-ethylphenylacetate **5b** was prepared by Friedel-Crafts acylation of **4d** which afforded ethyl 4-acetyl-3-methoxyphenylacetate (**4e**). Reduction of the ketone with triethylsilane and TFA produced the corresponding 4-ethyl substituted derivative **4f** which was demethylated with BBr₃ to afford **5b**. Ethyl 3-hydroxy-4-*iso*-propylphenylacetate (**5c**) was prepared by Wittig olefination of **4e** and subsequent catalytic hydrogenation of the 2-propenyl substituent to yield **4h**. Demethylation with boron tribromide produced **5c**.

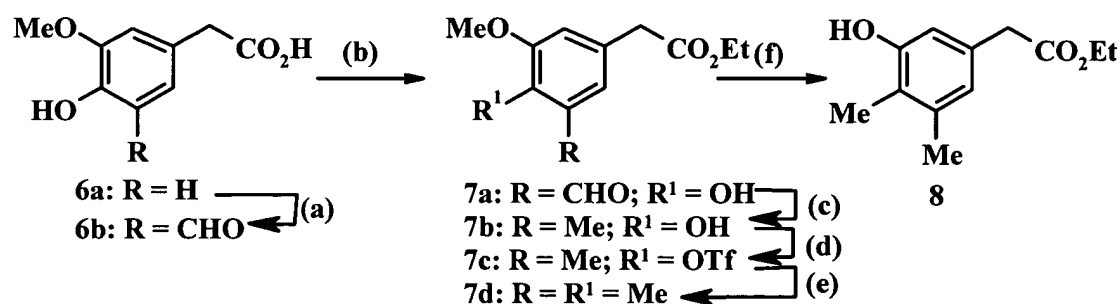
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SCHEME 2



Ethyl 3,4-dimethyl-5-hydroxyphenylacetate (**8**) was prepared by formylation of **6a** and esterification of the resulting carboxylic acid **6b** to produce ethyl 3-formyl-4-hydroxy-5-methoxyphenyl acetate (**7a**). Reduction of the aldehyde and hydrogenolysis the resulting benzyl alcohol afforded **7b**. The second methyl substituent was introduced by sequential treatment of **7b** with triflic anhydride which yielded **7c** and displacement with Me₂Zn, PdCl₂(dppf) and DIBAL-H (*supra*) to produce **7d**. Boron tribromide mediated demethylation afforded **8**. (Scheme 3)

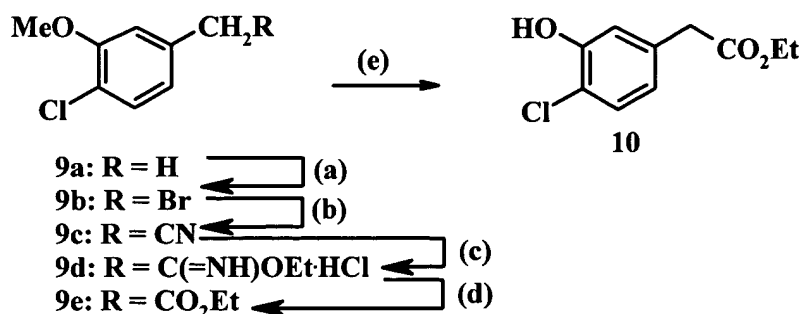
SCHEME 3



Ethyl 4-chloro-3-hydroxyphenyl acetate (**10**) was prepared from 4-chloro-3-methoxytoluene by sequential free radical bromination (**9b**), nucleophilic displacement of the bromine atom with cyanide

(9c) and a two-step hydrolysis of the nitrile to the amidine hydrochloride 9d and subsequently to the ethyl ester 9e. Boron tribromide mediated demethylation as described previously afforded 10. (Scheme 4)

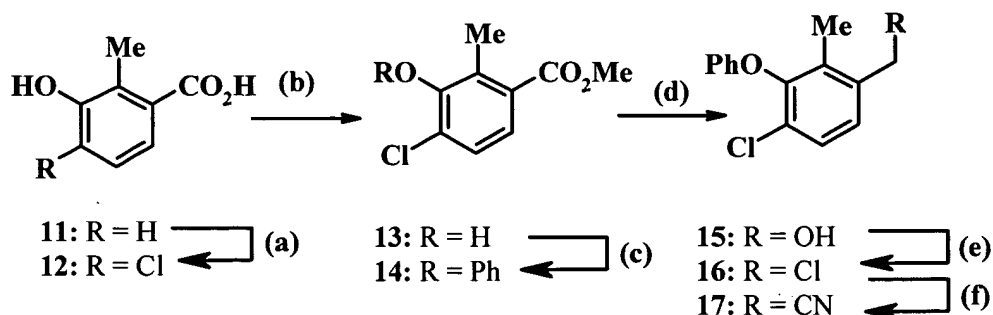
SCHEME 4



(a) NBS, benzoyl peroxide, CCl₄; (b) NaCN, 90% EtOH; (c) HCl, EtOH, Et₂O; (d) H₂O, 40 °C; (e) BBr₃, CH₂Cl₂, -78 °C

- 5 6-Methyl derivatives were prepared from 3-hydroxy-2-methylbenzoic acid (11) which was chlorinated (NaOCl/NaOH) and esterified to afford 13. Cupric acetate mediated coupling (*infra*) of benzenboronic acid provided the diaryl ether 14. The nitrile was introduced by sequential reduction, mesylation and cyanide displacement to afford 17. The mesylate underwent an *in situ* displacement by chloride during the mesylation reaction.

SCHEME 5



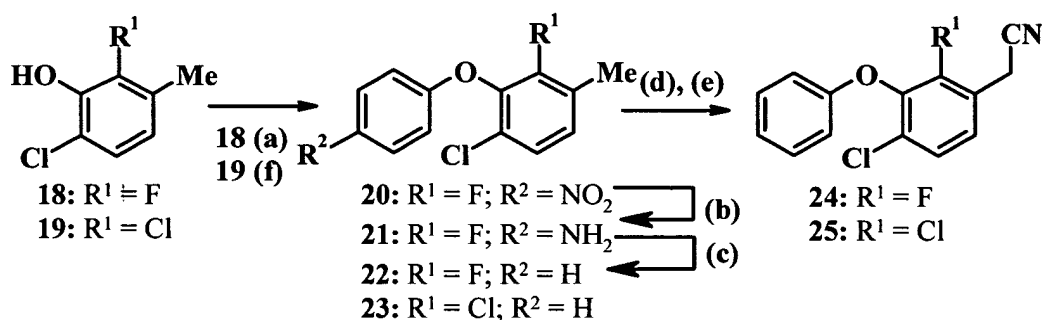
(a) NaOH, NaOCl, H₂O; (b) H₂SO₄, MeOH; (c) benzenboronic acid, Cu(OAc)₂, TEA, molecular sieves, CH₂Cl₂; (d) DIBAL-H; (e) MsCl, TEA; (f) NaCN, EtOH.

10

6- fluoro- and chloro- derivatives were available from 6-chloro-2-fluoro-3-methylphenol (18) and 3-bromo-2,4-dichlorotoluene (19), respectively (Scheme 6). The base-catalyzed reaction of 18 and

p-fluoro-nitrobenzene yielded diaryl ether **20**. Conversion of the nitro substituent to the corresponding amine followed by diazotization and reduction produced 4-chloro-2-fluoro-3-phenoxytoluene (**22**). One skilled in the art will appreciate that the availability of amino-substituted aryl groups affords the possibility to replace the amino substituent with a variety of other substituents utilizing the Sandmeyer reaction. Cupric chloride-mediated coupling (see *infra*) of **19** afforded the corresponding 2,4-dichloro-3-phenoxytoluene (**23**). Elaboration of the acetonitrile sidechain in **24** and **25** was accomplished by benzylic bromination and displacement.

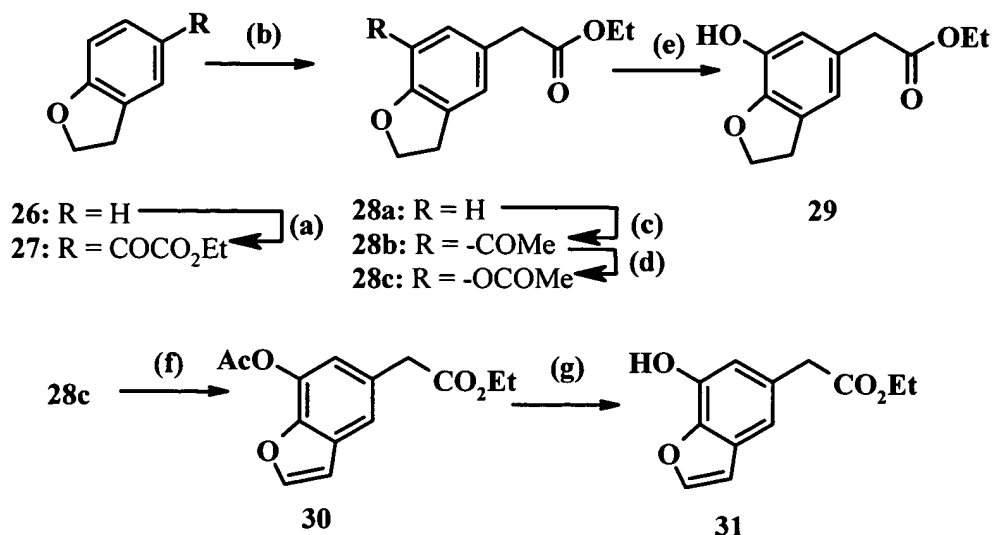
SCHEME 6



(a) K₂CO₃; (b) SnCl₂; (c)(i) NaNO₂, HOAc, HCl, H₂O (ii) FeSO₄·7H₂O, DMF; (d) (i) NBS, benzoyl peroxide, CCl₄ (ii) NaCN, EtOH; (f) Cs₂CO₃, CuCl, NMP.

Benzofuran **31** and dihydrobenzofuran **29** derivatives (Scheme 7) were prepared from dihydrobenzofuran (**26**). Acylation with ethyl chloro oxalate produced the α-ketoester **27** which was reduced to the corresponding phenylacetic acid derivative **28a** under Wolff-Kischner conditions. The preparation of **29** by a Wilgerodt reaction also has been reported (J. Dunn *et al.* *J. Med Chem* **1986** 29:2326). Freidel-Crafts acylation with acetyl chloride afforded the acetyl derivative **28b** which was converted to the acetate **28c** under Baeyer-Villiger conditions and subsequently hydrolyzed to **29**. The corresponding benzofuran analogs were prepared by benzylic bromination and concomitant dehydrohalogenation to yield **31**.

SCHEME 7



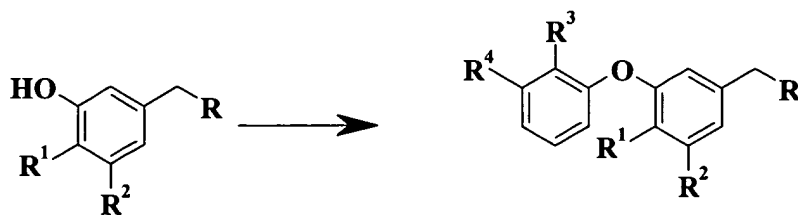
(a) ClC(=O)CO₂Et, AlCl₃, solvent; (b) NH₂NH₂, KOH, ethylene glycol;
 (c) CH₃COCl, AlCl₃, solvent; (d) H₂O₂, solvent; (e) hydrolysis conditions;
 (f) NBS; (g) NaHCO₃, H₂O, EtOH.

Preparation of Aryl Ether Intermediates (Scheme 1; 2; X = O or S)

The preparation of diaryl ethers has been reviewed (J. S. Sawyer, *Recent Advances in Diaryl Ether Synthesis, Tetrahedron* **2000** 56:5045-5065). The diaryl ethers required herein were prepared by three
 5 different methods (Scheme 8): (i) Cu(OAc)₂ catalyzed condensation of substituted benzene boronic acids and phenols (D. A. Evans *et al.*, *Synthesis of Diaryl Ethers through the Copper-Promoted Arylation of Phenols with Aryl Boronic Acids. An Expedient Synthesis of Thyroxine, Tetrahedron Lett.*, **1998** 39:2937-2940 and D. M. T. Chan *et al.*, *New N- and O-Arylations with Phenylboronic Acids and Cupric Acetate, Tetrahedron Lett.* **1998** 39:2933-2936; Scheme 1, conditions (a), (b), (e), (f), (i); (ii) by
 10 variations of the Ullmann diaryl ether synthesis with Cu(I) salts (J.-F. Marcoux *et al.*, *A General Copper-Catalyzed Synthesis of Diaryl Ethers, J. Am. Chem. Soc.* **1997** 119:10539-540; E. Buck *et al.*, *Ullmann Diaryl Ether Synthesis: Rate Acceleration by 2,2,6,6-tetramethylheptane-3,5-dione, Org. Lett.* **2002** 4(9):1623-1626); conditions (c), (d) and (h); or by nucleophilic aromatic displacement reactions (Sawyer *supra* pp 5047-5059; conditions Scheme 1(g) and (j). An alternative process utilizing
 15 palladium-catalyzed coupling procedures also has been reported (G. Mann *et al.*, *Palladium-Catalyzed Coupling Involving Unactivated Aryl Halides. Sterically Induced Reductive Elimination to Form the C-O Bond in Diaryl Ethers, J. Am. Chem. Soc.*, **1999** 121:3224-3225). The preparation of aryl ethers **33a-g** from the appropriate phenol **32a-e**, and **35a-d** from **34a** and **34b**, as described exemplifies these

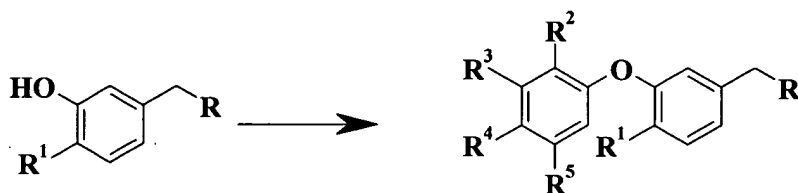
transformations. One skilled in the art will appreciate that optimal procedure will vary depending on the nature and position of substituents on the aryl rings.

SCHEME 8



- | | | |
|---|---|-----|
| 32a: R = CN, R ¹ = Me, R ² = H | 33a: R = CN, R ¹ = Me, R ² = R ⁴ = H, R ³ = Br | (a) |
| 32b: R = CO ₂ Et, R ¹ = Me, R ² = H | 33b: R = CO ₂ Et, R ¹ = Me, R ² = R ⁴ = H, R ³ = Cl | (b) |
| 32c: R = CO ₂ Et, R ¹ = Et, R ² = H | 33c: R = CO ₂ Et, R ¹ = Me, R ² = R ³ = H, R ⁴ = F | (c) |
| 32d: R = CO ₂ Et, R ¹ = <i>i</i> -Pr, R ² = H | 33d: R = CO ₂ Et, R ¹ = Et, R ² = R ⁴ = H, R ³ = Cl | (d) |
| 32e: R = CO ₂ Et, R ¹ = R ² = Me | 33e: R = CO ₂ Et, R ¹ = Et, R ² = R ³ = H, R ⁴ = Cl | (e) |
| | 33f: R = CO ₂ Et, R ¹ = <i>i</i> -Pr, R ² = R ³ = H, R ⁴ = Cl | (f) |
| | 33g: R = CO ₂ Et, R ¹ = R ² = Me, R ³ = H, R ⁴ = Cl | (f) |

(a) 2-bromobenzeneboronic acid, Cu(OAc)₂, pyridine, 4Å molecular sieves, CH₂Cl₂; (b) 2-chlorobenzeneboronic acid, Cu(OAc)₂, pyridine, 4Å molecular sieves, CH₂Cl₂; (c) *m*-fluorobromobenzene, CuCl Cs₂CO₃, TMHD, NMP; (d) 2-iodochlorobenzene; CuCl Cs₂CO₃, TMHD, NMP; (e) 3-chlorobenzeneboronic acid; Cu(OAc)₂, TEA, 4Å molecular sieves, CH₂Cl₂; (f) 3-chlorobenzeneboronic acid, Cu(OAc)₂, TEA, 4Å molecular sieves, CH₂Cl₂.

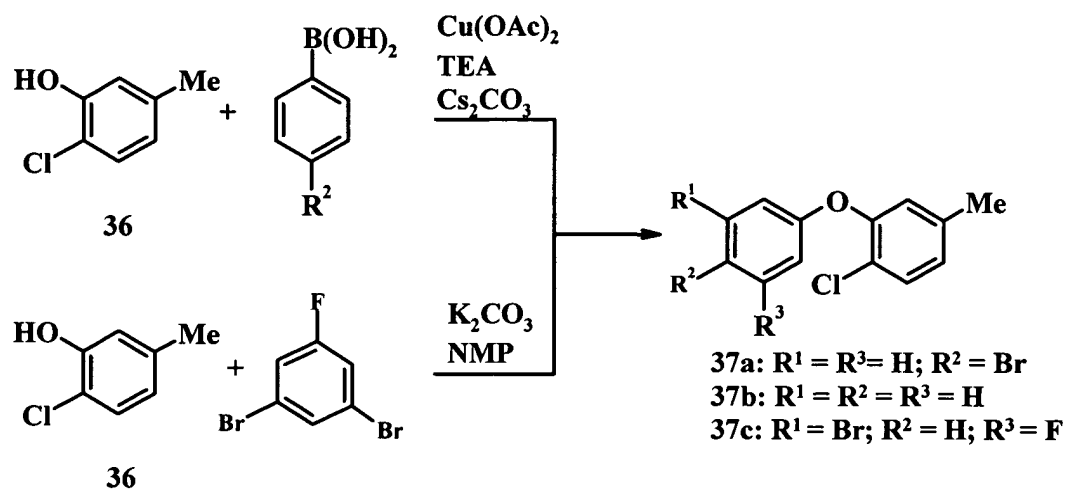


- | | | |
|---|---|-----|
| 34a: R = CO ₂ Et, R ¹ = Cl | 35a: R = CO ₂ Et, R ¹ = Cl, R ² = R ⁴ = H, R ³ = F, R ⁵ = CN | (g) |
| 34b: R = H, R ¹ = Cl | 35b: R = CO ₂ Et, R ¹ = R ² = R ⁵ = Cl, R ³ = R ⁴ = H | (h) |
| | 35c: R = CO ₂ Et, R ¹ = Cl, R ² = R ³ = R ⁵ = H, R ⁴ = Br | (i) |
| | 35d: R = H, R ¹ = Cl, R ³ = Br, R ⁵ = F, R ² = R ⁴ = H | (j) |

(g) 3,5-difluorobenzonitrile, K₂CO₃, NMP, 120 °C; (h) 2,5-dichlorobromobenzene, CuCl, Cs₂CO₃, TMHP, NMP 120 °C; (i) 4-bromobenzeneboronic acid, Cu(OAc)₂, TEA, 4Å molecular sieves, CH₂Cl₂; (j) 3,5-dibromofluorobenzene, Cs₂CO₃, TMHD, NMP.

Substituted *m*-cresol derivatives are also suitable substrates for coupling using these procedures. After introduction of the *meta* substituent the intermediate can be converted to the corresponding phenylacetonitrile derivative by bromination and cyanide displacement (Scheme 9).

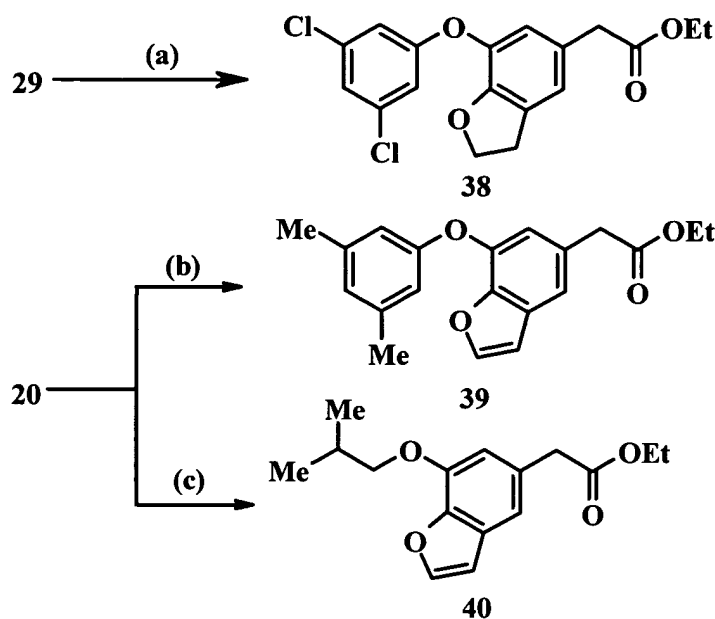
SCHEME 9



coupling of compounds with a fused aryl, heteroaryl or heterocyclic ring to produce diaryl ethers, alkylaryl ethers or arylalkylethers can be carried out by the same procedures. The preparation of alkyl

5 aryloxybenzofuranacetate and aryloxydihydrobenzofuranylacetate derivatives is exemplified in Scheme 10. Alkoxybenzofurans are prepared by Mitsunobu coupling of the alcohol and the hydroxybenzofuranacetic acid.

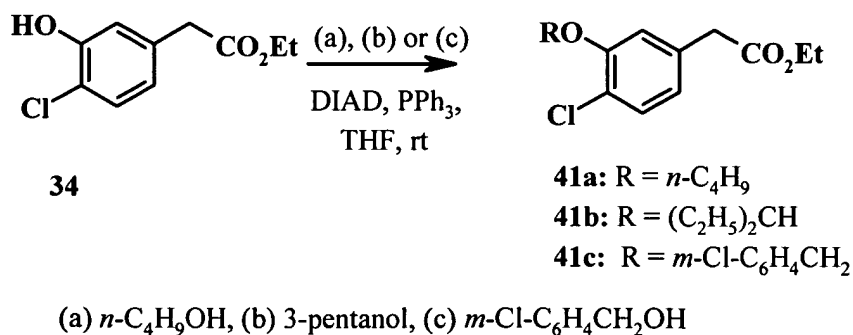
SCHEME 10



(a) 3,5-dichlorobenzeneboronic acid, $\text{Cu}(\text{OAc})_2$, TEA, molecular sieves, CH_2Cl_2 ; (b) 3,5-dimethylbenzeneboronic acid, $\text{Cu}(\text{OAc})_2$, TEA, molecular sieves, CH_2Cl_2 ; (c) $\text{Me}_2\text{CHCH}_2\text{OH}$, DIAD, PPh_3 .

Alkyl aryl and aralkyl aryl ethers were prepared using Mitsunobu conditions (Scheme 11; O. Mitsunobu, *Synthesis* **1981** 1-28). Alternatively alkyl and aralkyl ethers can be prepared via a classical Williamson ether synthesis (J. March, *Advanced Organic Chemistry*; 4th Edition; Wiley & Sons: New York, 1992; pp. 386-87) or utilizing palladium-catalyzed coupling (M. Palucki *et al.*, *Palladium-catalyzed Intermolecular Carbon-Oxygen Bond Formation: A New Synthesis of Aryl Ethers*, *J. Am. Chem. Soc.* **1997** 119:3395-96).

SCHEME 11



Preparation of Diphenylamine Intermediates (Scheme 1; X = NR⁶)

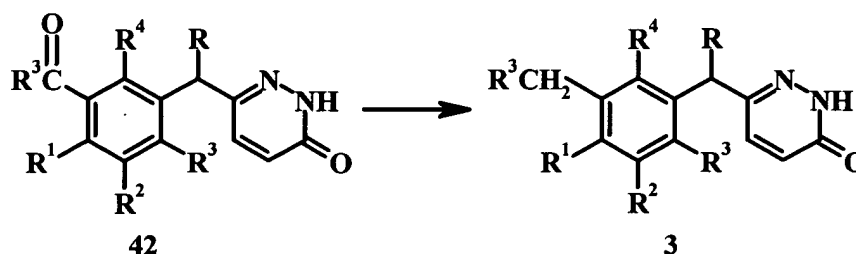
Diphenylamine compounds within the scope of the present invention can be prepared by palladium-catalyzed coupling reactions as described by Hartwig (*Transition Metal Catalyzed Synthesis of Aryl Amines and Aryl Ethers from Aryl Halides and Triflates: Scope and Mechanism, Angew. Chem. Int. Ed.*

5 *Eng.* **1998** 37:2046-67)

Preparation of Diphenyl Methane Intermediates (Scheme 1; 2: X = CH₂ or C=O)

Diphenylmethane compounds of the present invention can be prepared by reduction of the corresponding benzoyl derivatives **42**. While reductions are conveniently carried out with triethylsilylhydride and trifluoroacetic acid, a variety of other procedures to effect this transformation are well known within the art.

10



The preparation of the requisite benzoyl derivatives has been described in U. S. Patent No. 5,886,178 (D. A. Allen *et al.*). The synthesis of benzoyl substituted benzofuran derivatives have also been reported in U. S. Patent No. 4,780,480 (J. P. Dunn) and the scientific literature (J. P. Dunn *et al.* *Analgetic and Antiinflammatory 7-Aroylbenzofuran-5-ylacetic acids and 7-Aroylbenzothiophene-5-ylacetic Acids, J. Med. Chem.* **1986** 29:2326) These references are hereby incorporated by reference in its entirety.

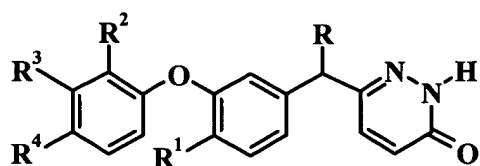
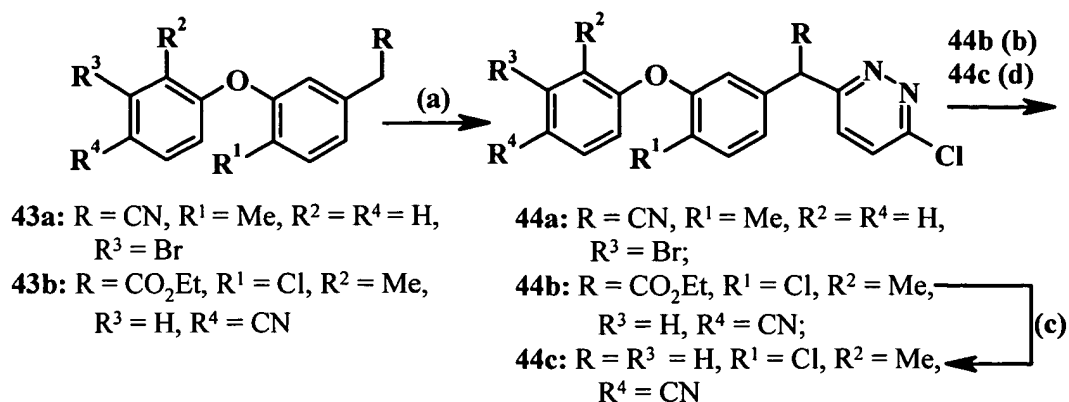
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20 Introduction of the Pyridazinone Ring (Scheme 1; 3)

Introduction of a pyridazinone is accomplished by base-catalyzed condensation of the appropriately substituted phenylacetate **43b** or phenylacetonitrile **43a** and 3,6-dichloropyrazine (Scheme 12). The condensation is accomplished efficiently with NaH and DMF. Hydrolysis of nitrile **44a** under acidic conditions with aqueous hydrochloric acid and acetic acid resulted in hydrolysis, decarboxylation and concomitant hydrolysis of the pyrazine to produce pyridazinone **45a**. Saponification of **44b** under basic conditions resulted in hydrolysis, acidification, and decarboxylation of the of the carboxylic acid to produce chloropyridazine **44c** which was converted to pyridazinone **45a** by exposure to sodium acetate and acetic acid or the more stringent acetic acid, water and hydrochloric acid.

25

SCHEME 12



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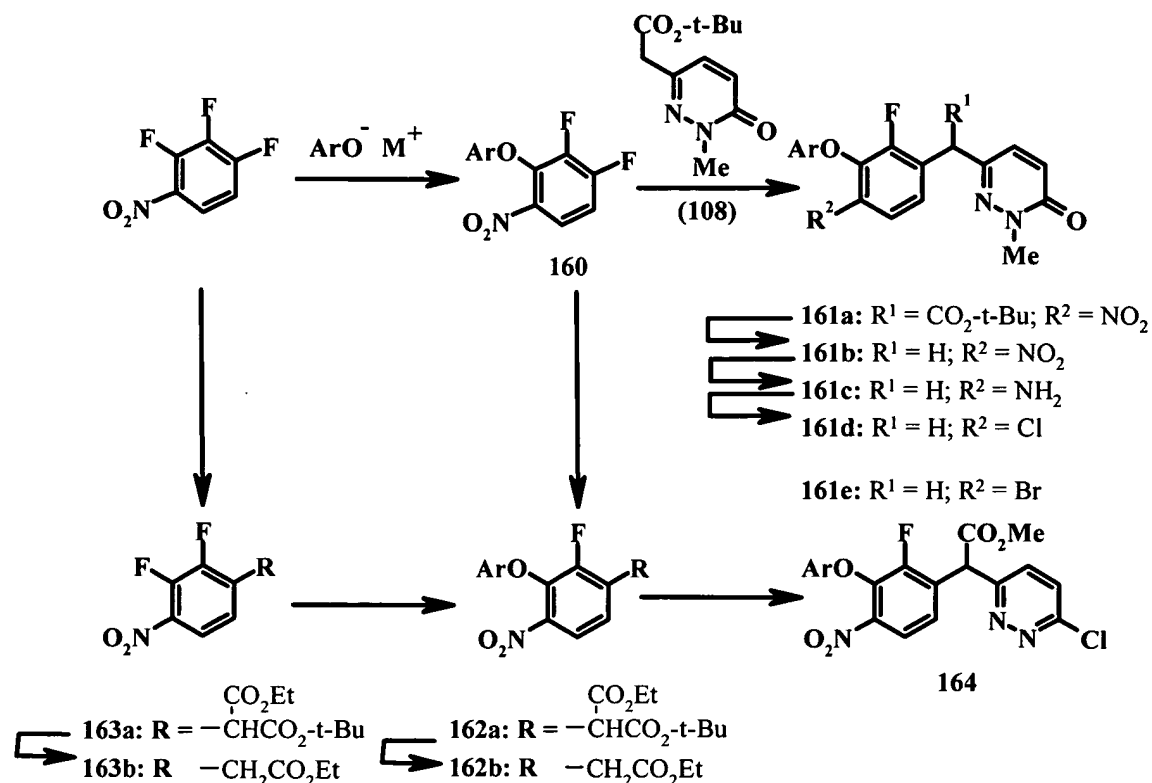
45a: R = R² = R⁴ = H, R¹ = Me, R³ = Br
45b: R = R³ = H, R¹ = Cl, R² = Me, R⁴ = CN

(a) 3,6-dichloropyrazine, NaH, DMF; (b) HOAc, HCl, H₂O; (c) LiOH, MeOH, H₂O;
 (d) NaOAc, HOAc

While the conditions exemplified utilized 3,6-dichloropyrazine, other pyrazine derivatives, e.g. 3-alkoxy-6-halopyridazines, may also be utilized in the preparation of compounds of the present invention
 5 (T. L. Drapier and T. R. Bailey, *J. Org. Chem.* **1995** 60(3):748-50; Druey *et al. Helv. Chim. Acta* **1954** 37:121). Substituted 3,6-dichloropyrazine derivatives can also be utilized. 5-Ethyl-, 5-*iso*-propyl and 5-hydroxymethyl-1,4-dichloropyrazines have been prepared J. G. Samaritoni *Org. Prep. & Procedures Int.* **1988** 20(2):117-121). (3,6-Dichloro-pyridazin-4-yl)-dimethyl-amine can be prepared by literature methods (Tsujiimoto *et al. Chem. Pharm. Bull.* **1979** 27:1169; Fenton *et al. J. Chem. Soc. C*, **1971**
 10 1536). (3,6-Dichloro-pyridazin-4-yl)-methyl-amine was prepared as described in Example 22. 4-(3,6-Dichloro-pyridazin-4-yl)-morpholine was prepared by the method of Fenton *et al. (J. Chem. Soc. Perkin Trans. I* **1972** 2323). While the pyrazine coupling is conveniently accomplished with phenylacetic acid and phenyl acetonitrile derivatives, other carbon nucleophiles can also be used. For example,
 15 deprotonation of an aromatic methyl substituent with strong bases, e.g. *n*-BuLi or lithium di-*iso*-propylamide affords a benzylic organometallic intermediate, ArO-C₆H₄CH₂⁻ M⁺, which can react with electrophiles including (see, e.g., R. B. Bates *et al. J. Org. Chem.* **1989** 54:311-17). Transmetalation of halomethylethers, ArO-C₆H₄CH₂X (March, *supra* pp. 454-456 and 920-931) provides an alternative

method to produce benzylic organometallic intermediates capable of reacting with 3,6-dichloropyrazine or equivalents.

SCHEME 13



Alternatively the pyridazinone can be introduced prior to final elaboration of the aromatic substituents by taking advantage of the propensity for fluoro nitrobenzene compounds to undergo nucleophilic aromatic substitution. Thus 2,3,4-trifluoronitrobenzene can be reacted with a phenoxide salt resulting in the displacement of the 3-fluoro radical to yield the aryl ether **160**. Deprotonation of the side chain of **108** and reaction with **160** results in displacement of the 4-fluoro radical to afford **161b** which can be saponified and decarboxylated as illustrated previously to yield **161b**. The remaining nitro radical allows for convenient introduction of a variety of substituents. Halogens can be introduced by a variant of the classical reduction, diazotization and displacement sequence (*i.e.*, the Sandmeyer reaction). While the reaction is exemplified in Scheme 13 with a chloride substituent, it would be readily apparent that other halogens could be introduced in similar manner. Further one skilled in the art will immediately recognize that **161e** is a versatile substrate for palladium-catalyzed reactions allowing the introduction of a wide variety of other functional groups at the 4-position into the aromatic ring. The sequence of steps can be altered to incorporate an acetic acid side chain which can be used to introduce the pyridazinone by alkylation of 3,6-pyrazine. Alkylation of 2,3,4-trifluoronitrobenzene with methyl *t*-butyl malonate affords **163a** which can be hydrolyzed and decarboxylated to afford **163b**. Alkylation

with a phenoxide affords **162b**. Alternatively the diaryl ether **160** can be subjected to displacement with a malonate ester to afford **162a** which can be hydrolyzed and decarboxylated in an analogous manner.

DOSAGE AND ADMINISTRATION

5 The compounds of the present invention may be formulated in a wide variety of oral administration dosage forms and carriers. Oral administration can be in the form of tablets, coated tablets, dragées, hard and soft gelatine capsules, solutions, emulsions, syrups, or suspensions. Compounds of the present invention are efficacious when administered by other routes of administration including continuous (intravenous drip) topical parenteral, intramuscular, intravenous, subcutaneous, transdermal (which may
10 include a penetration enhancement agent), buccal, nasal, inhalation and suppository administration, among other routes of administration. The preferred manner of administration is generally oral using a convenient daily dosing regimen which can be adjusted according to the degree of affliction and the patient's response to the active ingredient.

15 A compound or compounds of the present invention, as well as their pharmaceutically useable salts, together with one or more conventional excipients, carriers, or diluents, may be placed into the form of pharmaceutical compositions and unit dosages. The pharmaceutical compositions and unit dosage forms may be comprised of conventional ingredients in conventional proportions, with or without additional active compounds or principles, and the unit dosage forms may contain any suitable effective
20 amount of the active ingredient commensurate with the intended daily dosage range to be employed. The pharmaceutical compositions may be employed as solids, such as tablets or filled capsules, semisolids, powders, sustained release formulations, or liquids such as solutions, suspensions, emulsions, elixirs, or filled capsules for oral use; or in the form of suppositories for rectal or vaginal administration; or in the form of sterile injectable solutions for parenteral use. A typical preparation
25 will contain from about 5% to about 95% active compound or compounds (w/w). The term "preparation" or "dosage form" is intended to include both solid and liquid formulations of the active compound and one skilled in the art will appreciate that an active ingredient can exist in different preparations depending on the target organ or tissue and on the desired dose and pharmacokinetic parameters.

30

The term "excipient" as used herein refers to a compound that is useful in preparing a pharmaceutical composition, generally safe, non-toxic and neither biologically nor otherwise undesirable, and includes excipients that are acceptable for veterinary use as well as human pharmaceutical use. The term "excipient" as used herein includes both one and more than one such excipient.

Solid form preparations include powders, tablets, pills, capsules, cachets, suppositories, and dispersible granules. A solid carrier may be one or more substances which may also act as diluents, flavoring agents, solubilizers, lubricants, suspending agents, binders, preservatives, tablet disintegrating agents, or an encapsulating material. In powders, the carrier generally is a finely divided solid which is a mixture with the finely divided active component. In tablets, the active component generally is mixed with the carrier having the necessary binding capacity in suitable proportions and compacted in the shape and size desired. Suitable carriers include but are not limited to magnesium carbonate, magnesium stearate, talc, sugar, lactose, pectin, dextrin, starch, gelatin, tragacanth, methylcellulose, sodium carboxymethylcellulose, a low melting wax, cocoa butter, and the like. Solid form preparations may contain, in addition to the active component, colorants, flavors, stabilizers, buffers, artificial and natural sweeteners, dispersants, thickeners, solubilizing agents, and the like.

Liquid formulations also are suitable for oral administration include liquid formulation including emulsions, syrups, elixirs, aqueous solutions, aqueous suspensions. These include solid form preparations which are intended to be converted to liquid form preparations shortly before use. Emulsions may be prepared in solutions, for example, in aqueous propylene glycol solutions or may contain emulsifying agents such as lecithin, sorbitan monooleate, or acacia. Aqueous solutions can be prepared by dissolving the active component in water and adding suitable colorants, flavors, stabilizing, and thickening agents. Aqueous suspensions can be prepared by dispersing the finely divided active component in water with viscous material, such as natural or synthetic gums, resins, methylcellulose, sodium carboxymethylcellulose, and other well known suspending agents.

The compounds of the present invention may be formulated for parenteral administration (*e.g.*, by injection, for example bolus injection or continuous infusion) and may be presented in unit dose form in ampoules, pre-filled syringes, small volume infusion or in multi-dose containers with an added preservative. The compositions may take such forms as suspensions, solutions, or emulsions in oily or aqueous vehicles, for example solutions in aqueous polyethylene glycol. Examples of oily or nonaqueous carriers, diluents, solvents or vehicles include propylene glycol, polyethylene glycol, vegetable oils (*e.g.*, olive oil), and injectable organic esters (*e.g.*, ethyl oleate), and may contain formulatory agents such as preserving, wetting, emulsifying or suspending, stabilizing and/or dispersing agents. Alternatively, the active ingredient may be in powder form, obtained by aseptic isolation of sterile solid or by lyophilisation from solution for constitution before use with a suitable vehicle, *e.g.*, sterile, pyrogen-free water.

The compounds of the present invention may be formulated for topical administration to the epidermis as ointments, creams or lotions, or as a transdermal patch. Ointments and creams may, for example, be formulated with an aqueous or oily base with the addition of suitable thickening and/or gelling agents.

5 Lotions may be formulated with an aqueous or oily base and will in general also contain one or more emulsifying agents, stabilizing agents, dispersing agents, suspending agents, thickening agents, or coloring agents. Formulations suitable for topical administration in the mouth include lozenges comprising active agents in a flavored base, usually sucrose and acacia or tragacanth; pastilles comprising the active ingredient in an inert base such as gelatin and glycerin or sucrose and acacia; and
10 mouthwashes comprising the active ingredient in a suitable liquid carrier.

The compounds of the present invention may be formulated for administration as suppositories. A low melting wax, such as a mixture of fatty acid glycerides or cocoa butter is first melted and the active component is dispersed homogeneously, for example, by stirring. The molten homogeneous mixture is
15 then poured into convenient sized molds, allowed to cool, and to solidify.

The compounds of the present invention may be formulated for vaginal administration. Pessaries, tampons, creams, gels, pastes, foams or sprays containing in addition to the active ingredient such carriers as are known in the art to be appropriate.

20

The compounds of the present invention may be formulated for nasal administration. The solutions or suspensions are applied directly to the nasal cavity by conventional means, for example, with a dropper, pipette or spray. The formulations may be provided in a single or multidose form. In the latter case of a dropper or pipette, this may be achieved by the patient administering an appropriate, predetermined
25 volume of the solution or suspension. In the case of a spray, this may be achieved for example by means of a metering atomizing spray pump.

The compounds of the present invention may be formulated for aerosol administration, particularly to the respiratory tract and including intranasal administration. The compound will generally have a small
30 particle size for example of the order of five (5) microns or less. Such a particle size may be obtained by means known in the art, for example by micronization. The active ingredient is provided in a pressurized pack with a suitable propellant such as a chlorofluorocarbon (CFC), for example, dichlorodifluoromethane, trichlorofluoromethane, or dichlorotetrafluoroethane, or carbon dioxide or other suitable gas. The aerosol may conveniently also contain a surfactant such as lecithin. The dose of

drug may be controlled by a metered valve. Alternatively the active ingredients may be provided in a form of a dry powder, for example a powder mix of the compound in a suitable powder base such as lactose, starch, starch derivatives such as hydroxypropylmethyl cellulose and polyvinylpyrrolidone (PVP). The powder carrier will form a gel in the nasal cavity. The powder composition may be presented in unit dose form for example in capsules or cartridges of *e.g.*, gelatin or blister packs from which the powder may be administered by means of an inhaler.

When desired, formulations can be prepared with enteric coatings adapted for sustained or controlled release administration of the active ingredient. For example, the compounds of the present invention can be formulated in transdermal or subcutaneous drug delivery devices. These delivery systems are advantageous when sustained release of the compound is necessary and when patient compliance with a treatment regimen is crucial. Compounds in transdermal delivery systems are frequently attached to a skin-adhesive solid support. The compound of interest can also be combined with a penetration enhancer, *e.g.*, Azone (1-dodecylaza-cycloheptan-2-one). Sustained release delivery systems are inserted subcutaneously into to the subdermal layer by surgery or injection. The subdermal implants encapsulate the compound in a lipid soluble membrane, *e.g.*, silicone rubber, or a biodegradable polymer, *e.g.*, polylactic acid.

Suitable formulations along with pharmaceutical carriers, diluents and excipients are described in *Remington: The Science and Practice of Pharmacy 1995*, edited by E. W. Martin, Mack Publishing Company, 19th edition, Easton, Pennsylvania. A skilled formulation scientist may modify the formulations within the teachings of the specification to provide numerous formulations for a particular route of administration without rendering the compositions of the present invention unstable or compromising their therapeutic activity.

The modification of the present compounds to render them more soluble in water or other vehicle, for example, may be easily accomplished by minor modifications (salt formulation, esterification, etc.), which are well within the ordinary skill in the art. It is also well within the ordinary skill of the art to modify the route of administration and dosage regimen of a particular compound in order to manage the pharmacokinetics of the present compounds for maximum beneficial effect in patients.

The term "therapeutically effective amount" as used herein means an amount required to reduce symptoms of the disease in an individual. The dose will be adjusted to the individual requirements in each particular case. That dosage can vary within wide limits depending upon numerous factors such as

the severity of the disease to be treated, the age and general health condition of the patient, other medicaments with which the patient is being treated, the route and form of administration and the preferences and experience of the medical practitioner involved. For oral administration, a daily dosage of between about 0.01 and about 100 mg/kg body weight per day should be appropriate in monotherapy and/or in combination therapy. A preferred daily dosage is between about 0.1 and about 500 mg/kg body weight, more preferred 0.1 and about 100 mg/kg body weight and most preferred 1.0 and about 10 mg/kg body weight per day. Thus, for administration to a 70 kg person, the dosage range would be about 7 mg to 0.7 g per day. The daily dosage can be administered as a single dosage or in divided dosages, typically between 1 and 5 dosages per day. Generally, treatment is initiated with smaller dosages which are less than the optimum dose of the compound. Thereafter, the dosage is increased by small increments until the optimum effect for the individual patient is reached. One of ordinary skill in treating diseases described herein will be able, without undue experimentation and in reliance on personal knowledge, experience and the disclosures of this application, to ascertain a therapeutically effective amount of the compounds of the present invention for a given disease and patient.

15

In embodiments of the invention, the active compound or a salt can be administered in combination with another antiviral agent, such as a nucleoside reverse transcriptase inhibitor, another nonnucleoside reverse transcriptase inhibitor or HIV protease inhibitor. When the active compound or its derivative or salt are administered in combination with another antiviral agent the activity may be increased over the parent compound. When the treatment is combination therapy, such administration may be concurrent or sequential with respect to that of the nucleoside derivatives. "Concurrent administration" as used herein thus includes administration of the agents at the same time or at different times. Administration of two or more agents at the same time can be achieved by a single formulation containing two or more active ingredients or by substantially simultaneous administration of two or more dosage forms with a single active agent.

25

It will be understood that references herein to treatment extend to prophylaxis as well as to the treatment of existing conditions, and that the treatment of animals includes the treatment of humans as well as other animals. Furthermore, treatment of a HIV infection, as used herein, also includes treatment or prophylaxis of a disease or a condition associated with or mediated by HIV infection, or the clinical symptoms thereof.

30

The pharmaceutical preparations are preferably in unit dosage forms. In such form, the preparation is subdivided into unit doses containing appropriate quantities of the active component. The unit dosage

form can be a packaged preparation, the package containing discrete quantities of preparation, such as packeted tablets, capsules, and powders in vials or ampoules. Also, the unit dosage form can be a capsule, tablet, cachet, or lozenge itself, or it can be the appropriate number of any of these in packaged form.

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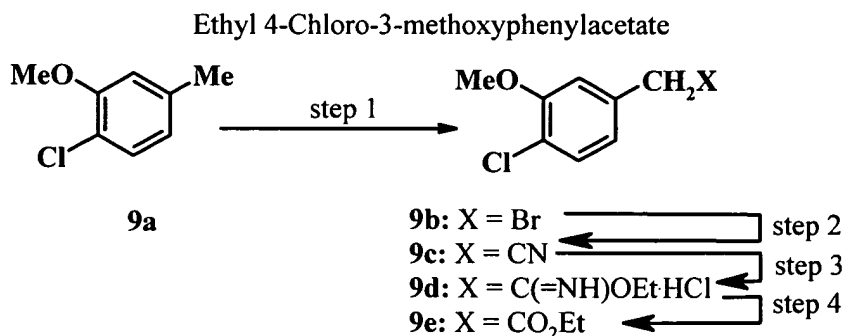
The pharmaceutical compositions in Example 46 are given to enable those skilled in the art to more clearly understand and to practice the present invention. They should not be considered as limiting the scope of the invention, but merely as being illustrative and representative thereof.

- 10 The pharmaceutical preparations are preferably in unit dosage forms. In such form, the preparation is subdivided into unit doses containing appropriate quantities of the active component. The unit dosage form can be a packaged preparation, the package containing discrete quantities of preparation, such as packeted tablets, capsules, and powders in vials or ampoules. Also, the unit dosage form can be a capsule, tablet, cachet, or lozenge itself, or it can be the appropriate number of any of these in packaged
- 15 form.

- The compounds of formula I may be prepared by various methods known in the art of organic chemistry. The starting materials for the syntheses are either readily available from commercial sources or are known or may themselves be prepared by techniques known in the art. The following examples
- 20 (*infra*) are given to enable those skilled in the art to more clearly understand and to practice the present invention. They should not be considered as limiting the scope of the invention, but merely as being illustrative and representative thereof.

EXAMPLE 1

25



Step 1

A solution of 4-chloro-3-methoxytoluene (**9a**; 0.5 g; 3.2 mmol), NBS (0.57 g; 3.2 mmol) and benzoyl peroxide (0.031 g; 0.13 mmol) and 32 mL of DCE were heated at reflux for 3 h. The reaction mixture

was cooled, diluted with CH_2Cl_2 and washed with water and brine. The organic extract was dried filtered and evaporated to yield the bromomethyl compound **9b** which was used without further purification.

Step 2

- 5 The 28 g (0.166 mmol) of **9b** from the previous step, NaCN (28 g; 0.58 mmol; 3.5 equiv.) and 500 mL of 90% aqueous EtOH were stirred at room temperature overnight. The crude residue was partitioned between EtOAc/H₂O (359 mL of each), washed with brine, dried, filtered and evaporated. Silica gel chromatography and elution with a gradient (100% hexane → 90:10 hexane:EtOAc) yielded 21 g of **9c**.

step 3

- 10 Gaseous HCl was slowly bubbled into a cooled solution of 4-chloro-3-methoxyacetonitrile (**9b**) in toluene (10mL), ether (10 mL) and EtOH (1 mL) for about 10 min. The reaction was stoppered and stored at -30 °C for one week. TLC failed to detect any remaining starting material. The solvent was evaporated and the yellow solid was stirred with Et₂O, filtered and washed with Et₂O and dried in a vacuum oven to yield 0.57 g (90%) of ethyl 4-chloro-3-methoxyphenylmethyylimidate (**9d**).

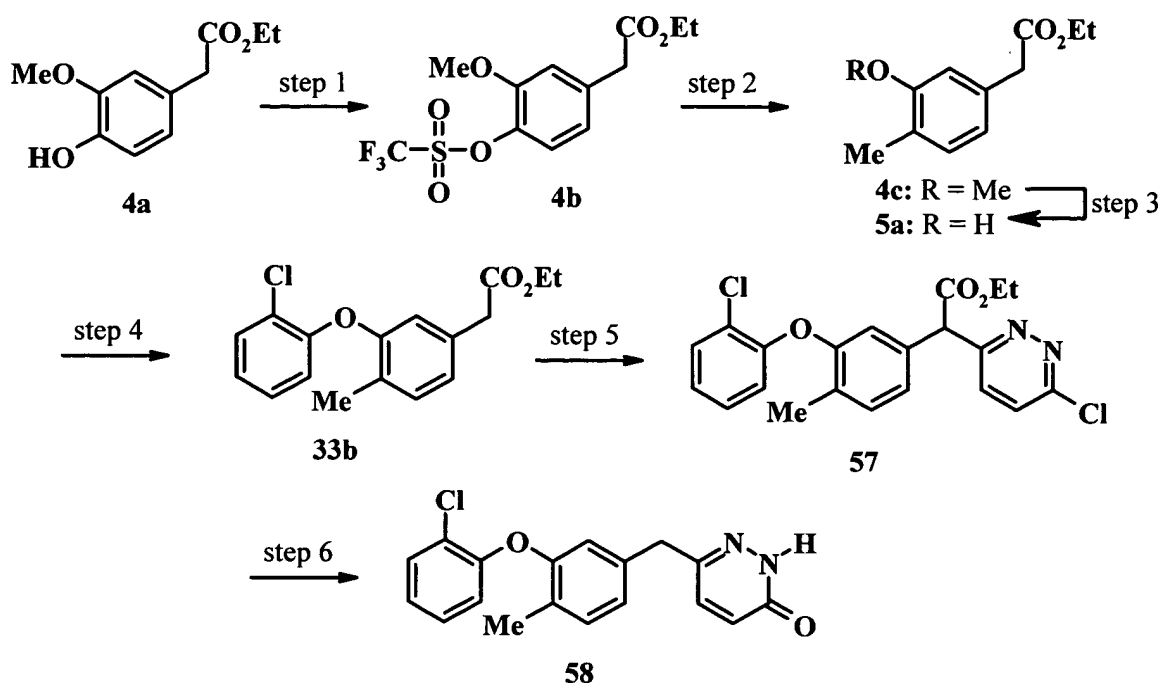
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step 4

A solution of 0.57 g of **9d** and 10 mL of H₂O was heated at 40 °C for 3 h. The reaction was cooled to rt and extracted with EtOAc. The reaction was dried (MgSO₄), filtered and evaporated and the resulting product **9e** was used without further purification.

EXAMPLE 2

6-[3-(2-chloro-phenoxy)-4-methylbenzyl]-2H-pyridazin-3-one



5 step 1

To a cooled solution of ethyl 4-hydroxy-3-methoxyphenylacetate (**4a**; 13.7 g; 65.2 mmol) and 260 mL of CH₂Cl₂ under N₂ atmosphere was added dropwise triflic anhydride (16 mL; 97.9 mmol) followed by dropwise addition of pyridine (8.9 mL; 8.8 mmol). The reaction was stirred in an ice-water bath for 3 h. The solution was transferred to a separatory funnel and washed with water and brine, dried (Na₂SO₄),
 10 filtered and evaporated to yield 21 g (90%) of **4b**.

step 2

To a solution of ethyl 3-methoxy-4-trifluorosulfonyloxyphenylacetate (**4b**) in 4 mL of THF cooled in an ice-water bath was added slowly a solution of Pd(dppf)Cl₂ (0.024 g; 0.029 mmol) and DIBAL-H (6 mL; 0.058 mmol; 1.0M in PhMe) and a small quantity of THF followed by dimethylzinc (0.29 mL; 0.58 mmol; 2.0 M in PhMe). After addition was completed the ice bath was removed and the reaction allowed to warm to rt and then heated to reflux for 1 h. The reaction was carefully quenched with a small quantity of water, filtered through a pad of CELITE® and the solids washed thoroughly with EtOAc. The combined organic extracts were washed with water and brine, dried (MgSO₄) and the
 15 solvent evaporated to afford 0.240 g (85%) of ethyl 3-methoxy-4-methylphenylacetate (**4c**).

step 3

To a solution of **4c** (2.2 g; 8.0 mmol) and 250 mL CH₂Cl₂ cooled to -78 °C was added dropwise *via* syringe BBr₃ (9.8 mL; 0.104 mol). After 1 h at -78 °C the reaction was stirred for 4 h in an ice-water bath. The reaction mixture was re-cooled to -78 °C and the reaction quenched aqueous NaHCO₃ then warmed to rt and the organic phase washed with water, saturated NaHCO₃ and brine. The organic phase was dried (MgSO₄) and the solvent evaporated to afford 1.4 g of ethyl 3-hydroxy-4-methylphenylacetate (**5a**).

step 4

To a suspension of **5a** (4.8 g; 25 mmol), 2-chlorobenzeneboronic acid (7.8 g; 50 mmol), Cu(OAc)₂ (5 g; 27.5 mmol), powdered 4Å molecular sieves (15 g) and 250 mL of CH₂Cl₂. After 4 days starting material was still evident by tlc and an addition 5.0 g of the boronic acid was added. The reaction was stirred for an additional day and the suspension filtered through a pad of CELITE® and silica gel. The solids were washed well with CH₂Cl₂. The combined filtrates were washed sequentially with 2N HCl (2 x 25 mL), NaHCO₃ (25mL), water and brine. The extracts were dried (MgSO₄), filtered and evaporated. The crude product was purified by silica gel chromatography and eluted with 25% EtOAc:hexane to yield 2.2 g (28%) of **33b**.

step 5

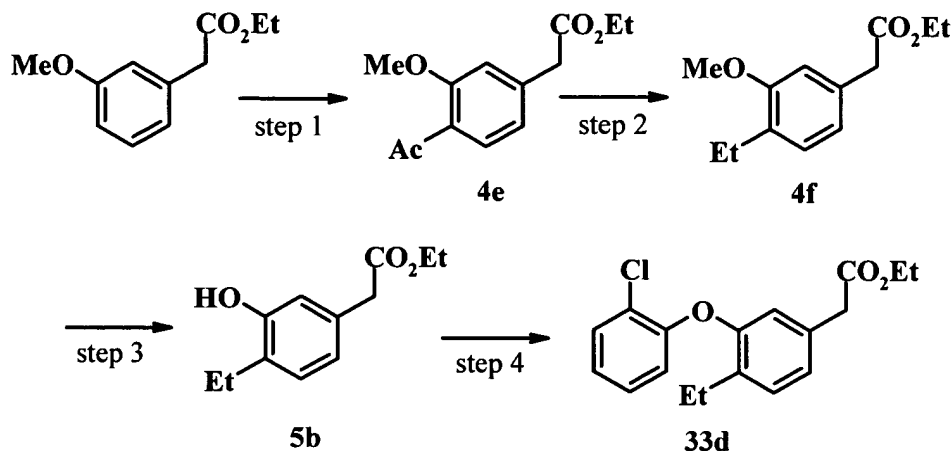
To an ice-cold solution of **33b** (0.7g; 7.2 mmol), 3,6-dichloropyrazine (2.1 g; 14.4 mmol) and 72 mL of dry DMF was added portionwise NaH (0.4 g; 15.2 mmol; 60% in mineral oil). The reaction was stirred at 0°C for 15 min and allowed to warm to rt overnight. The reaction mixture and poured into 100 mL of H₂O containing about 1 g of NaHSO₄. The organic phase was washed with EtOAc and the combined organic extracts washed with 5% LiCl, dried (Na₂SO₄), filtered and evaporated. The product was purified by silica gel chromatography and eluted with 20% EtOAc:hexane to yield 0.72 g (37%) of **57**.

step 6

A mixture of 0.72 g (2.6 mmol) of **57**, HOAc (3.5 mL), HCl (7 mL) and H₂O (3.5 mL) were heated at reflux for 6 h, cooled to rt, diluted with water and extracted with EtOAc. The combined extracts were washed sequentially with water, sat'd NaHCO₃, and brine, dried (Na₂SO₄), filtered and concentrated. The crude product was purified by chromatography on silica gel. The eluted product, which still contained the 3-chloropyridazine was dissolved in HOAc (20mL) and NaOAc (0.2 g) and reisolated to yield 0.4 g (50%) of **58** as a white solid; m.p. 116-118.

EXAMPLE 3

3-(2-Chloro-phenoxy)-4-ethyl-phenyl]-acetic acid ethyl ester



5 step 1

To a stirred solution of ethyl 3-methoxyphenylacetate (16.0 g; 82.38 mmol) in CH_2Cl_2 (200 mL) at rt was added dropwise AcCl (9.88 mL; 138.9 mmol) followed by stannic chloride (16.9 mL; 169 mmol; 1.0 M solution in CH_2Cl_2). The reaction mixture was stirred at rt for 6 h and poured into an ice-water mixture. The aqueous phase was extracted with CH_2Cl_2 and the combined extracts were washed with water, dried (Na_2SO_4) and the solvent removed *in vacuo*. The crude product **4e** was purified by chromatography on silica gel and eluted with CH_2Cl_2 :EtOAc (20:1) to yield 13.96 g (69.5%) of a white solid.

step 2

To a solution of **4e** (19g; 80.42 mmol) and 200 mL of TFA cooled to 0°C was added an excess of Et_3SiH and the reaction allowed to warm to rt for 3 h. Excess TFA was removed *in vacuo* and the residue partitioned between water and CH_2Cl_2 . The crude product was purified by chromatography on silica gel and eluted with CH_2Cl_2 :hexane (3:1) to yield 3.0 g (16%) of **4f**.

20 step 3

A solution of ethyl 4-ethyl-3-methoxyphenylacetate (**4f**; 3.0 g; 13.50 mmol) and CH_2Cl_2 (80 mL) was cooled to -78°C and a solution of BBr_3 (5.10 mL; 53.94 mmol; 1.0 M in CH_2Cl_2) over 30 min. After 1 h at -78°C the reaction was allowed to warm to rt and stirred for 12 h. The reaction was cooled in an ice-water bath and the reaction quenched with 20 mL of water. The aqueous phase was extracted with CH_2Cl_2 :EtOAc (4:1 v/v), dried (Na_2SO_4), filtered and evaporated. The crude product was purified by

silica gel chromatography and eluted with a CH₂Cl₂:EtOAc gradient (100:1 → 100:4) to yield **5b** (2.0 g; 71%); m.s. 209.2 (M+H)⁺.

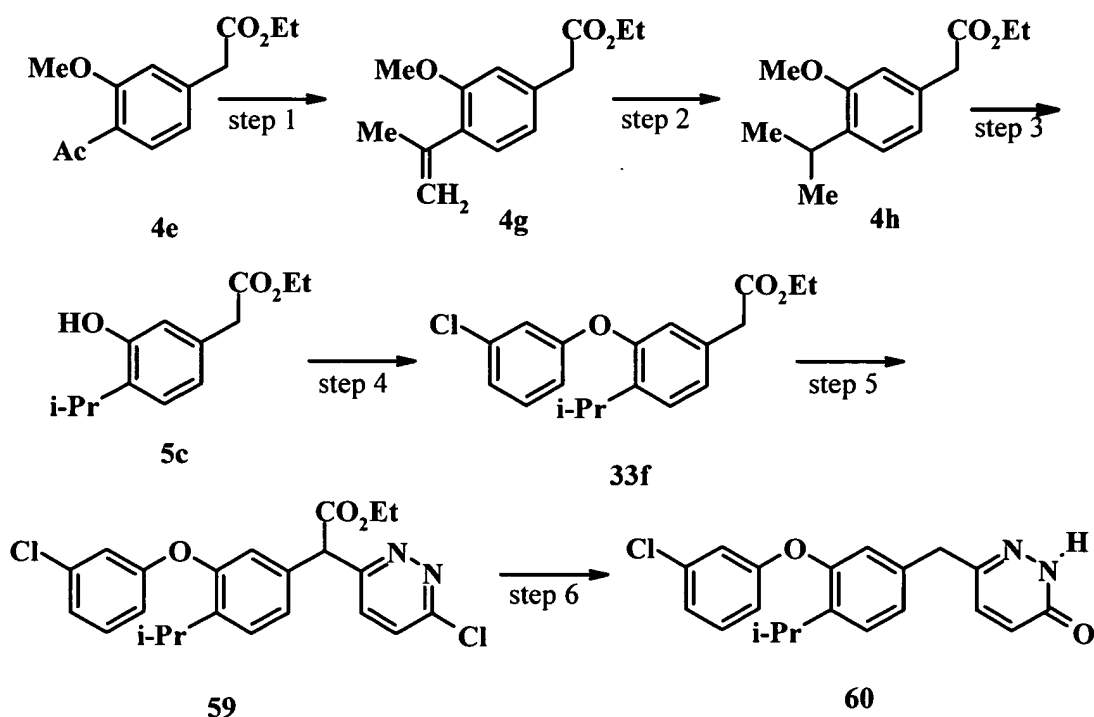
step 4

- 5 A solution of ethyl 4-ethyl-3-hydroxyphenylacetate (**5b**, 0.20 g; 0.96 mmol), 2-iodo-chlorobenzene (0.18 mL; 1.44 mmol), Cs₂CO₃ (0.469 g; 1.44 mmol), TMHD (0.020 mL; 0.096 mmol) and NMP (15 mL) was degassed with a stream of nitrogen for 15 m. Cuprous chloride (0.48 g; 4.8 mmol) was added and the solution was degassed. The reaction mixture was heated to 120 °C for 11 h then cooled to rt. The suspension was filtered through a pad of CELITE® and the solid washed thoroughly with EtOAc.
- 10 The combined filtrate was washed with 2N HCl, dried (Na₂SO₄) and the solvent evaporated. The product was purified by chromatography on silica gel and eluted with EtOAc:hexane (1:10) to yield 0.31 g (39%) of **33d**.

EXAMPLE 4

6-[3-(3-Chloro-phenoxy)-4-isopropyl-benzyl]-2H-pyridazin-3-one

15



step 1

- To a suspension of PPh₃CH₃⁺ Br⁻ (36.29 g; 101.6 mmol) in THF (150 mL) cooled to -40 °C was added dropwise *n*-BuLi (40.6 mL; 1.6M in hexanes) and the resulting solution was allowed to warm to -10 °C for 10 m and re-cooled to -40 °C. To the resulting solution was added in one portion ethyl 4-acetyl-3-
- 20

methoxyphenylacetate (see Example 4; step 1) and the reaction mixture was stirred at 0 °C for 30 m and warmed to rt and stirred for an additional 2 h. The reaction mixture was diluted with hexane filtered through a pad of CELITE ® and the solids washed with hexane:Et₂O (5:1 v/v; 60 mL). The combined organic layers were washed with water (50 mL) and brine (50 mL), dried (Na₂SO₄), filtered and
5 evaporated to yield a yellow oil. The product was purified by silica gel chromatography and eluted with CH₂Cl₂:hexane (1:1 → 2:1) to yield 9.1 g of **4g**.

step 2

A suspension of **4g** (9.0 g; 38.41 mmol), 5% Pd/C (380 mg) in 50 mL HOAc and 50 mL EtOH was
10 shaken under a hydrogen atmosphere (50 psi) for 7 h. The mixture was filtered through a pad of CELITE ® and the filtered catalyst was washed with EtOAc. The solvents were evaporated under reduced pressure and the residue dissolved in MTBE and carefully washed with sat'd NaHCO₃, water and brine. The resulting solution was dried (Na₂SO₄), filtered and evaporated to yield ethyl 4-*iso*-propyl-3-methoxyphenylacetate (**4h**; 9.0 g) as a yellow oil.

15

step 3

A solution of **4h** (3.38 g; 14.30 mmol) and CH₂Cl₂ (150 mL) were cooled to -78 °C and a solution of BBr₃ (5.41 mL; 57.22 mmol) in 130 mL of CH₂Cl₂ were added dropwise over a 30 m period. The reaction mixture was stirred at -78 °C for 1 h, allowed to warm to rt for 4 h and re-cooled to -78 °C and
20 carefully quenched with sat'd. NaHCO₃ (80 mL). The aqueous layer was extracted with CH₂Cl₂ (1 x 100 mL), EtOAc (50 mL) and the combined aqueous layers washed with water and brine, dried (Na₂SO₄) and evaporated to yield a light brown oil. The phenol was purified by silica gel chromatography and eluted with CH₂Cl₂:hexane (3:1) → CH₂Cl₂ → CH₂Cl₂:EtOAc (100:4) to yield ethyl 4-*iso*-propyl-3-hydroxyphenylacetate (**5c**; 3.0 g; 94%)

25

step 4

To a solution of **5c** (1.0 g; 4.5 mmol), 3-chlorobenzenboronic acid (0.844 g; 5.4 mmol), cupric acetate (0.899 g; 4.95 mmol), 4Å molecular sieves (5.0 g) and CH₂Cl₂ (50 mL) was added TEA (3.14 mL; 22.53 mmol) and the reaction was stirred for 3 days. The reaction mixture was filtered through a pad of
30 CELITE®. The top layer containing the molecular sieves was removed and stirred with CH₂Cl₂ and refiltered. The combined organic filtrates were washed with 2N HCl, brine, dried (Na₂SO₄), filtered and evaporated. The crude product was chromatographed with silica gel and eluted with a gradient of hexane/EtOAc (90%hexane/EtOAc) to yield **33f** (1.0 g; 66%).

step 5

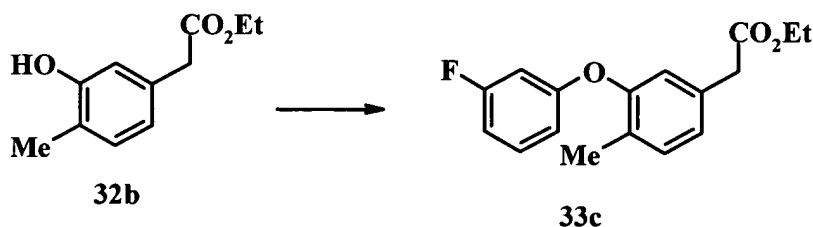
To a ice-cold solution of **33f** (1.0 g; 3.00 mmol), of 3,6-dichloropyrazine (0.895 g; 7.50 mmol) in 15 mL dry DMF was added portionwise 0.300 g NaH(7.50 mmol; 60% in oil). The reaction was allowed to warm to ambient temperature and stirred 6 h. The reaction was poured onto a mixture of ice, water and sodium bisulfate. The mixture was extracted thoroughly with EtOAc and the combined extracts were washed 6 times with water and brine. The extract was dried (MgSO₄), filtered and evaporated and the residue chromatographed on silica gel and eluted with a hexane:EtOAc gradient (15:1 → 8:1) to yield 0.80 g of **59** (purity *ca.* 80%)

step 6

A mixture of 0.64 g (1.44 mmol) of **59**, HOAc (12 mL), HCl (24 mL) and H₂O (12 mL) were heated at reflux for 16 h, cooled to rt and extracted with EtOAc. The combined extracts were washed with H₂O, dried (Na₂SO₄), filtered and concentrated *in vacuo* to afford a brown solid. The crude product was purified by chromatography on silica gel and eluted with a gradient of CH₂Cl₂:EtOAc (15:1 → 8:1) to yield 0.10 g (20%) of **60**.

EXAMPLE 5

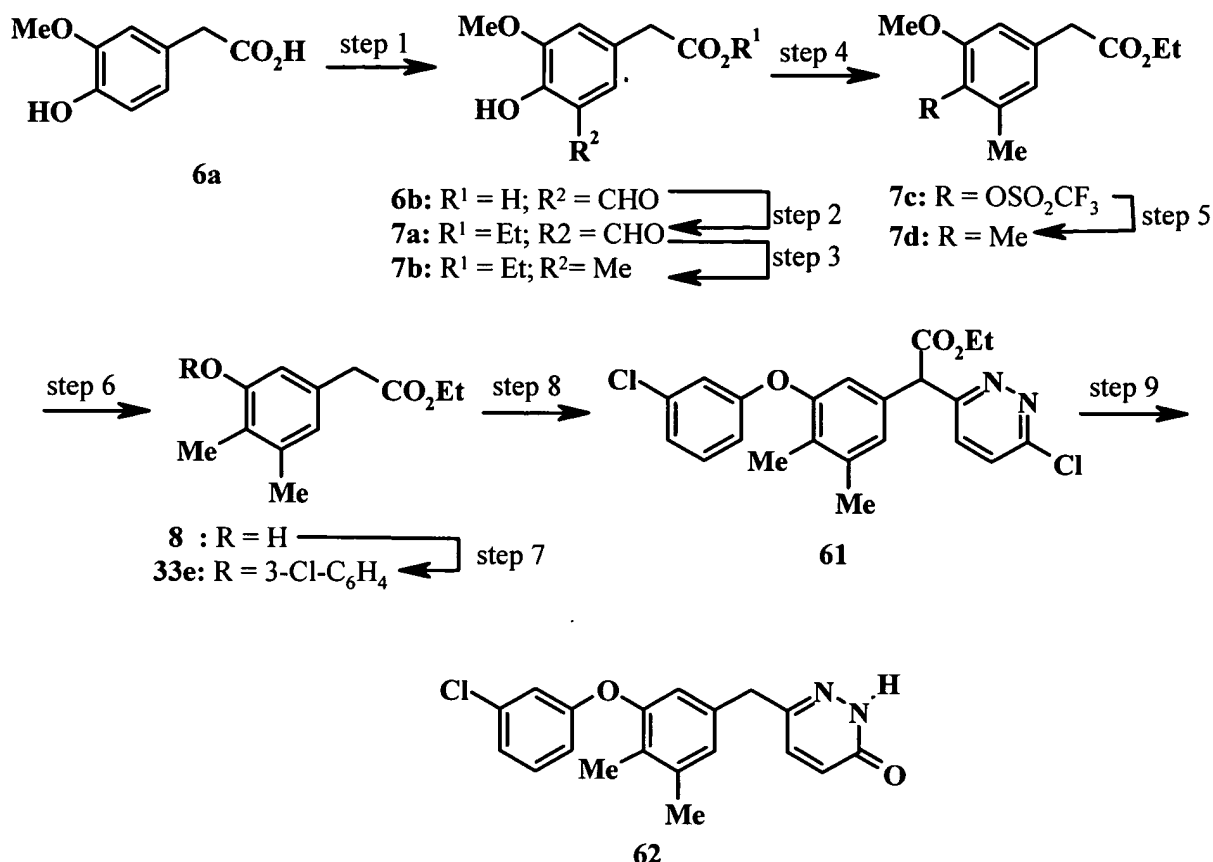
Ethyl 4-methyl-3-(3-fluorophenoxy)phenylacetate



To a stirred solution of **32b** (0.80 g; 4.12 mmol) and 7 mL NMP under a N₂ atmosphere was added 1-bromo-3-fluorobenzene (0.69 mL; 6.18 mmol), TMHD (0.086 mL; 0.41 mmol), Cs₂CO₃ (2.68 g; 8.24 mmol) and Cu(I)Cl (0.204 g; 2.06 mmol). The reaction was heated to 120 °C for 3 h. The reaction mixture was cooled to ambient temperature, and quenched with a mixture of 2 N HCl and EtOAc. The aqueous layer was thrice extracted with EtOAc and the combined organic layers were washed with water and brine, dried (MgSO₄), filtered and evaporated to dryness. The crude product was chromatographed on silica gel and eluted with hexane:Et₂O (9:1) which yielded **33c** (0.60 g; 50%).

EXAMPLE 6

6-[3-(3-chloro-phenoxy)-4,5-dimethyl-benzyl]-2H-pyridazin-3-one



step 1

A mixture of 4-hydroxy-3-methoxyphenylacetic acid (**6a**; 1.0 g; 5.49 mmol) and hexamethylenetetramine (0.808 g; 5.76 mmol) and TFA (7 mL) were stirred and heated at 90 °C for 4 h. The reaction was cooled and excess TFA removed *in vacuo* and 35 mL of ice and water was added to the residue. The resulting dark brown solution was stirred at rt for 20 m. The aqueous solution was extracted with Et₂O (40 mL) and the extract was dried (Na₂SO₄), filtered and evaporated to afford 0.70 g of **6b** (61%; m.s. (M+H)⁺ = 211.13; mw = 210).

step 2

To a solution of **6b** (4.0 g; 19.03 mmol) in EtOH (80 mL) was added con H₂SO₄ (1 mL). The reaction was heated at reflux for 6 h. Approximately 80% of the EtOH was removed *in vacuo* and the residue partitioned between EtOAc/H₂O (1:1) the organic phase residue washed with 10% NaHCO₃, water (100 mL), dried (Na₂SO₄), filtered and evaporated to afford a brown oil **7a** (88%; m.s. (M+H)⁺ = 239.19; mw = 238.3).

step 3

A mixture of **7a** (3.70 g; 15.53 mmol), 5% Pd/C (0.350 g), HOAc (45 mL) were shaken under a H₂ atmosphere (40 psi) for 8 h. TLC showed product and the corresponding benzyl alcohol. An additional
5 300 mg of Pd/C in 25 mL HOAc was added and hydrogenation continued for another 8 h. A second portion of 0.15 g of Pd/C in HOAc (15 mL) was added and reaction continued for another 12 h. The mixture was diluted with EtOAc and filtered through a pad of CELITE®. The catalyst was washed with EtOAc and the combined organic extracts dried (Na₂SO₄) and evaporated. The product was purified by silica gel chromatography and eluted with CH₂Cl₂:hexane (4:1) to afford 2.64 g of **7b** (75.8%).

10

step 4

To a solution of **7b** (5.87 g; 26.175 mmol) in CH₂Cl₂ cooled to 0 °C was added pyridine (3.60 mL; 44.51 mmol) followed by dropwise addition of triflic anhydride (6.605 mL; 39.26 mmol) over about 20 min. The reaction was stirred at 0 °C for 3.5 h. The reaction mixture was extracted with dilute HCl and
15 half-saturated NaHCO₃, dried (Na₂SO₄) and evaporated to yield 9.41 g of **7c** as a brown oil (100%).

step 5

To a suspension of PdCl₂(dppf) (0.650 g; 0.785 mmol) in THF (40 mL) cooled to 0 °C was added dropwise a solution of DIBAL-H (1.0 M in PhMe; 1.57 mL; 1.57 mmol). The resulting mixture was
20 stirred at 0 °C for 5 minutes and a solution of **7c** in 5 mL of THF was added followed by Me₂Zn (23 mL; 46.0 mmol; 1.0 M in PhMe). The mixture was stirred at 0 °C for 5 m and heated at reflux for 2.5 h then cooled to rt for 30 m. The reaction was poured into dilute HCl and extracted with EtOAc (2 x 100 mL), dried (Na₂SO₄), and evaporated. The crude product was purified by silica gel chromatography and eluted with CH₂Cl₂:hexane (1:2 → 1:1 → 2:1 v/v) to yield 5.1 g (87.6%) of **8**.

25

step 6

A solution of ethyl 3,4-dimethyl-5-methoxyphenylacetate (**8**; 0.560 g; 2.519 mmol) and CH₂Cl₂ (40 mL) was cooled to -78 °C and a solution of BBr₃ (10.1 mL; 10.1 mmol; 1.0 M in CH₂Cl₂) dropwise over 10 min. After 1 h at -78 °C the reaction was allowed to warm to rt and stirred for 12 h. The reaction was
30 cooled in an ice-water bath and the reaction quenched with 15 mL of ice/water. The aqueous phase was extracted with CH₂Cl₂:EtOAc (3:1 v/v), dried (Na₂SO₄), filtered and evaporated to yield **8** (0.52 g; 99%; m.s. 209.21 (M+H)⁺).

step 7

To a suspension of ethyl 3,4-dimethyl-5-hydroxyphenylacetate (**8**, 1.0 g; 4.8 mmol), 3-chloro-benzeneboronic acid (0.901 g; 5.762 mmol), Cu(OAc)₂ (0.959 g; 5.28 mmol), powdered 4Å molecular sieves (5 g) and 40 mL of CH₂Cl₂. After 40 h starting material was still evident by tlc and an addition
5 0.35 g of the boronic acid was added. The reaction was stirred for an additional 72 h. The reaction mixture was filtered through a pad of CELITE® and silica gel. The solids were washed well with CH₂Cl₂. The combined filtrates were washed sequentially with 2N HCl (2 x 25 mL), NaHCO₃ (25mL), water and brine. The extracts were dried (Na₂SO₄), filtered and evaporated. The crude product was purified by silica gel chromatography and eluted with EtOAc:hexane (1:15 → 1:10) to yield **33e** (1.0 g;
10 65%; m.s. (M+H)⁺ = 319.34, mw = 318).

step 8

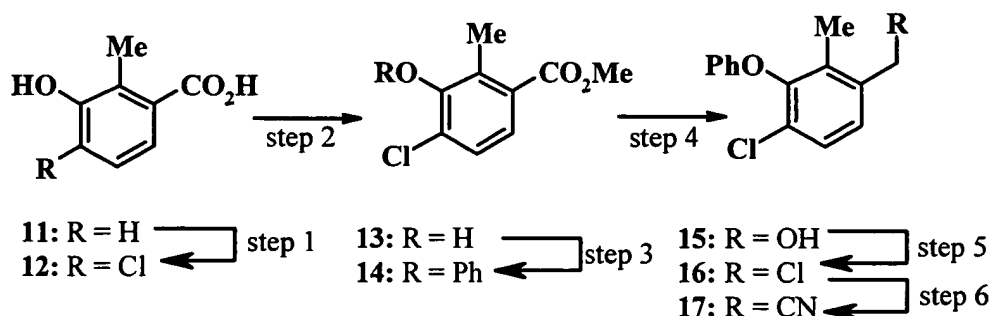
To a solution of 1.0 g of **33e** (3.14 mmol), 0.935 g (6.276 mmol) of 3,6-dichloropyrazine in 10 mL dry DMF cooled in an ice-water bath was added portionwise 0.313 g NaH (7.825 mmol; 60% in oil). The
15 reaction stirred at 0 °C for 5 m then was allowed to warm to ambient temperature and stirred for 14 hour. The reaction was poured onto a mixture of ice, water and sodium bisulfate. The mixture was extracted thoroughly with EtOAc and the combined extracts were washed with 5% LiCl, water and brine. The extract was dried (MgSO₄), filtered and evaporated and the residue chromatographed on silica gel and eluted with hexane:EtOAc (10:1 → 8:1) to yield 1.0 g (73.9%) of **61**: m.s. (M+H)⁺ =
20 431.29)

step 9

A mixture of 1.0 g (2.318 mmol) of **61**, HOAc (12 mL), HCl (24 mL) and H₂O (12 mL) were heated at reflux for 16 h, cooled to rt and extracted with EtOAc. The combined extracts were washed with H₂O,
25 dried (Na₂SO₄), filtered and concentrated *in vacuo* to afford a brown solid. The crude product was purified by chromatography on silica gel and eluted with a gradient of CH₂Cl₂:EtOAc (8:1) to yield 0.150 g (18%) of **62** as a brown solid: m.s. (M+H)⁺ = 341.27; mw = 340.8.

EXAMPLE 7

(4-chloro-2-methyl-3-phenoxy-phenyl)-acetonitrile



5 step 1

To a suspension of 3-hydroxy-2-methylbenzoic acid (**11**; 22.8 g; 0.15 mol) and water (300 mL) cooled in an ice-water bath was added 3 M NaOH to adjust pH to about 10 (*ca.* 60 mL). NaOCl (208 mL; 5.35% aqueous solution; 0.15 mol) was added dropwise over about 30 m while maintaining the temperature between 2-6 °C. After the addition was completed, 90 mL of 3 M HCl was added in one
 10 portion. The resulting precipitate was collected and dried on a sintered glass filter. The crude product was recrystallized from Et₂O:hexane (*ca.* 3:1) to yield a yellow solid **12** (12.24 g; 44%).

step 2

A solution of **12** (12.24 g; 65.6 mmol), MeOH (200 mL) and con H₂SO₄ (3.85 mL) was stirred
 15 overnight at rt then heated to reflux for 6 h. The solution was cooled, concentrated to approximately 10% of the original volume and the residue redissolved in EtOAc. The organic phase was washed with sat'd. NaHCO₃ and brine, dried, filtered and evaporated. The crude product was purified by silica gel chromatography and eluted with a EtOAc:hexane gradient (1:9 → 4:6). The combined fractions were evaporated to yield **13** (8.32 g; 63.2%).

20

step 3

To a solution of methyl 4-chloro-3-hydroxy-2-methylbenzoate (**13**; 1.0 g; 4.98 mmol), benzenboronic acid (1.52 g; 12.5 mmol), cupric acetate (1.00 g; 5.48 mmol), 4Å molecular sieves (1 g), and CH₂Cl₂ (25 mL) was added TEA (3.47 mL; 24.9 mmol) and the reaction was stirred overnight. Starting material
 25 was still detected by tlc and an additional 0.62 g of benzenboronic acid was added and stirred for another 24 h. The reaction mixture was filtered through a pad of CELITE®. The top layer containing the molecular sieves was washed with CHCl₃. The combined organic filtrates were evaporated. The

crude product was chromatographed with silica gel and eluted with hexane/EtOAc gradient (100:0 → 85:15) to yield **14** (0.82 g; 60%).

step 4

- 5 To a solution of methyl 4-chloro-2-methyl-3-phenoxybenzoate (**14**; 0.780g; 2.81 mmol) dissolved in PhMe (20 mL) cooled in an ice-water bath was added dropwise DIBAL-H (7.41 mL; 7.41 mmol; 1.0 M in PhMe) The reaction was quenched by sequentially adding MeOH, H₂O, and con HCl. The organic phase was extracted with Et₂O. The combined organic extracts were washed with sat'd. NaHCO₃, water and brine, dried (Na₂SO₄), filter and evaporated to yield **15** as a colorless oil which was used in the next
10 step without further purification.

step 5

- To a solution of **15** (0.736 g; 2.96 mmol) dissolved in pyridine (10 mL) was added dropwise methanesulfonyl chloride (0.252 μL; 5.92 mmol) over 5 min. After 30 min a small quantity of starting
15 material was evident and an addition 25 μL of methanesulfonyl chloride was added. The reaction was partitioned between Et₂O and 5% HCl. The organic phase was twice washed with 5% HCl, water, sat'd. NaHCO₃ and brine. The organic extract was dried (MgSO₄), filtered and evaporated. The crude product was chromatographed on silica gel eluting with 10% EtOAc:hexane to yield the benzylic chloride **16** (0.220 g) as a colorless oil.

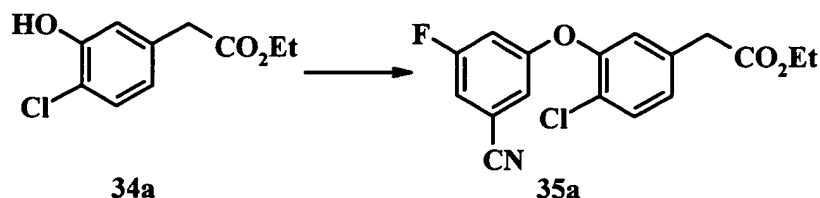
20

step 6

- The benzyl chloride **16** (0.220 g; 0.82 mmol) was dissolved in EtOH (1 mL) and KCN (0.107 g; 1.64 mmol and 1 mL of water. The mixture was heated to reflux and CH₃CN (0.3 mL) was added to produce a homogenous solution which was allow to reflux overnight. The reaction mixture was concentrated *in*
25 *vacuo* and partitioned between water and CH₂Cl₂. The organic phase was washed twice with brine, dried (MgSO₄), filtered and evaporated to yield **17** (0.210 g) sufficiently pure for further processing.

EXAMPLE 8

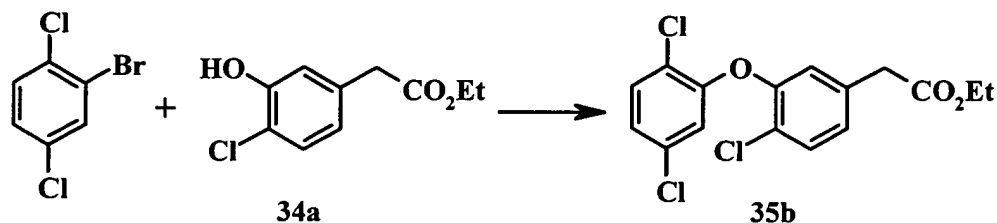
Ethyl 4-chloro-3-(3-cyano-5-fluorophenoxy)phenylacetate



5 To a solution of ethyl 4-chloro-3-hydroxyphenylacetate (**34a**; 1.4 g; 6.5 mmol) and NMP (13 mL) was added potassium carbonate (2.7g; 19.6 mmol) and 1.2 g of 3,5-difluorobenzonitrile (1.2 g; 8.5 mmol). The reaction mixture was heated to 120 °C and monitored by TLC. After 3.5 h an additional 0.9 g of K₂CO₃ was added and at 5.5 h an additional 0.9 g of K₂CO₃ and 0.3 g of 3,5-difluorobenzonitrile was added. After 8 h of heating the reaction was cooled to rt and the reaction mixture was filtered through a
10 pad of CELITE® and the solid cake was washed well with EtOAc. The filtrate was washed with 2 portions of 2N HCl, 1N NaOH, water and brine. The organic extract was dried (MgSO₄), filtered and evaporated to yield 1.3 g of the ether **35a**.

EXAMPLE 9

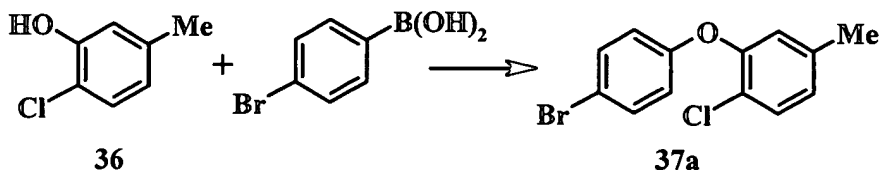
15 Ethyl 4-chloro-3-(2,5-dichlorophenoxy)phenylacetate



A solution of ethyl 4-chloro-3-hydroxyphenylacetate (**34a**; 2.0 g; 9.3 mmol), 2,5-dichloro-
bromobenzene, Cs₂CO₃ (6.0 g; 18.6 mmol), TMHD (0.38 mL; 1.9 mmol) and NMP (15 mL) was
20 degassed with a stream of nitrogen for 15 m. Cuprous chloride (0.5 g; 4.7 mmol) was added and the solution again was degassed. The reaction mixture was heated to 120 °C for 18 h then cooled to rt. The suspension was filtered through a pad of CELITE® and the solid washed thoroughly with EtOAc. The combined filtrate was washed with 2N HCl, dried (Na₂SO₄) and the solvent evaporated. The product was purified by chromatography on silica gel and eluted with EtOAc:hexane (1:10) to yield **35b** (0.554
25 g; 16%).

EXAMPLE 10

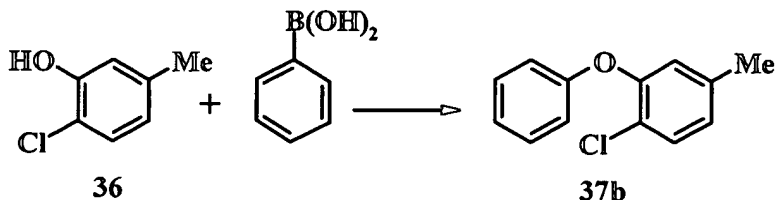
4-Chloro-3-(4-bromophenoxy)toluene



- 5 A solution of 2-chloro-4-methylphenol (36; 3.0 g; 21 mmol), 4-bromobenzeneboronic acid (5.0 g; 24 mmol), cupric acetate (4.2 g; 23.1 mmol), 4Å molecular sieves and CH₂Cl₂ (210 mL) was added TEA (9.8 mL; 70 mmol) and the reaction was stirred for 3 days. The reaction mixture was filtered through a pad of CELITE®. The top layer containing the molecular sieves was removed and stirred with CH₂Cl₂ and refiltered. The combined organic filtrates were washed with 2N HCl, brine, dried (Na₂SO₄), filtered and evaporated. The crude product was chromatographed with silica gel and eluted with a gradient of hexane/EtOAc (100:0 → 90:10) to yield 37a.

EXAMPLE 11

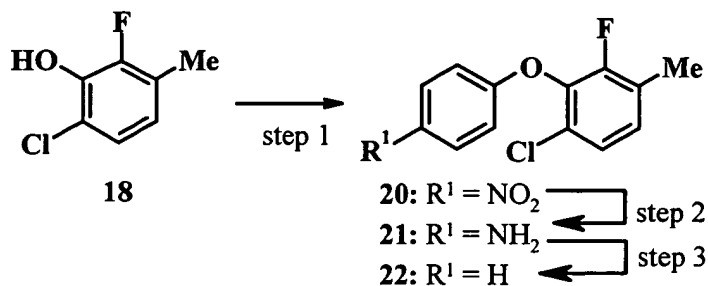
4-chloro-3-phenoxytoluene



- 15 To a solution of benzeneboronic acid (1.9 g; 15.8 mmol) dissolved in CH₂Cl₂ (250 mL) was added 2-chloro-5-methylphenol (36; 2.5 g; 17.5 mmol), cupric acetate (3.5 g; 19.3 mmol), TEA ((12.3 mL; 87.7 mmol) and 12.5 g of 4Å molecular sieves. The reaction was stirred for 24 h and an additional aliquot of benzeneboronic (2.4 g; 19.3 mmol) was added and stirring continued for an additional 48 hr. The reaction mixture was filtered through a bed of CELITE® and the filtered solids were washed thoroughly with CH₂Cl₂. The combined organic extracts were washed with 2N HCl, H₂O, sat'd NaHCO₃, H₂O and brine, dried (MgSO₄) filtered and evaporated. The crude product was purified by silica gel chromatography and eluted with hexane:EtOAc (9:1) to yield 37b (1.6 g; 47.1 %) as a clear oil.

EXAMPLE 12

4-Chloro-2-fluoro-3-phenoxytoluene



step 1

- 5 A solution of 4-chloro-2-fluoro-3-hydroxytoluene (**18**; 0.161 g; 1.0 mmol; N. Imazaki *et al.*, WO 2002100833), *p*-fluoronitro-benzene (0.141 g; 1.0 mmol), K₂CO₃ (0.276 g; 2 mmol) and DMF (4 mL) was heated to reflux for 4 h under a N₂ atmosphere. The reaction was cooled to rt and poured into water and stirred for several minutes. The aqueous solution was extracted twice with CH₂Cl₂ and the combined organic extracts washed with brine, dried (MgSO₄), filtered and evaporated to yield **20**.

10

step 2

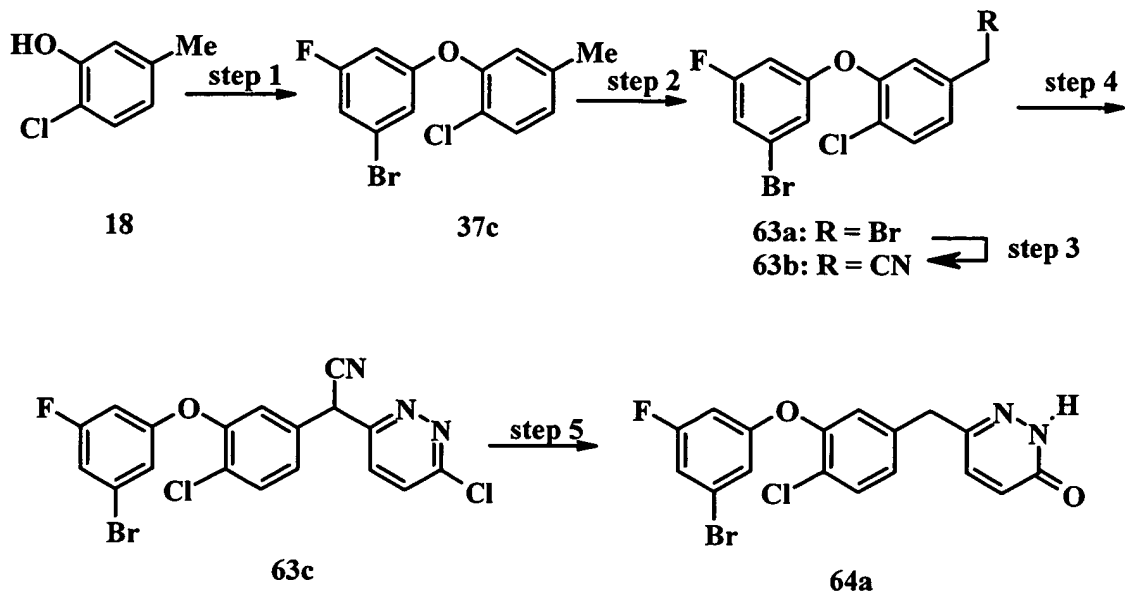
- A solution of **20** (1.58 g; 5.3 mmol), stannous chloride dihydrate (6.0 g; 26.6 mmol) and EtOH (5 mL) were heated to 70 °C stirred overnight. The reaction mixture was added to a small quantity of ice and made basic with 10% Na₂CO₃. The aqueous phase was extracted with EtOAc (5 mL) which resulted in an emulsion. About 7 mL of ethylenediamine was added to chelate remaining tin which resulted in a blue aqueous solution. The EtOAc was washed with water and brine, dried (NaHCO₃), filtered and evaporated to yield 1.35 g of **21** which was carried on to the next step.
- 15

step 3

- 20 A solution of **21** (0.830 g; 3.3 mmol) was dissolved in HOAc (2.25 mL) and added to a solution of ice-water (7.5 mL) and HCl (1.2 mL). A solution of NaNO₂ (0.254 g; 5.6 mmol) and H₂O (1.5 mL) was added over a 10-15 m period. The resulting solution was stirred for several minutes then added dropwise over 15 m to a suspension of FeSO₄·7H₂O (0.917 g; 3.3 mmol) and DMF (10.5 mL). The reaction was stirred for 0.5 h and a mixture of hexanes:EtOAc (1:1; 30 mL) was added. The organic phase was washed thrice with water, dried (MgSO₄), filtered and concentrated *in vacuo*. The dark oil was purified by chromatography on silica gel and eluted with an EtOAc:hexane gradient (0:100 → 20:80) which yielded **22** as a clear oil (0.450 g; 58%).
- 25

EXAMPLE 13

[3-(3-Bromo-5-fluoro-phenoxy)-4-chloro-phenyl]-(6-oxo-1,6-dihydro-pyridazin-3-yl)-acetonitrile, (**64a**)
and 6-[3-(3-Bromo-5-fluoro-phenoxy)-4-chloro-benzyl]-2H-pyridazin-3-one, (**64b**)



5

step 1

Cesium carbonate (11.4 g; 8.79 mmol) was added to a solution of 2-chloro-5-methylphenol (**18**; 2.5 g; 17.53 mmol) and NMP (16 mL). The resulting slurry was degassed and the flask alternately purged and refilled with nitrogen. 1,3-Dibromo-fluorobenzene (3.54 g; 28.13 mmol), TMHD (0.92 mL; 0.81 g; 4.41 mmol) and Cu(I)Cl (0.87 g; 8.79 mmol) were added sequentially and the reaction mixture was heated to 110 °C for 6 h. The reaction mixture was cooled to ambient temperature, filtered through a bed of CELITE® and the filter cake washed thoroughly with EtOAc. The filtrate was washed sequentially with dilute HCl, dilute NaOH, water and brine. The organic extract was dried (Na₂SO₄), filtered and evaporated. The residue was chromatographed on silica gel and eluted with hexane:Et₂O which yielded 1.8 g (32%) of **37c** as a colorless oil.

15

step 2

A mixture of **37c** (1.8 g; 5.704 mmol), NBS (1.066 g; 5.989 mmol), benzoyl peroxide (0.069 g; 0.28 mmol) and CCl₄ (20 mL) was heated to 90 °C for 2.5 h. The reaction mixture was cooled to room temperature and poured into 100 mL of H₂O. The mixture was extracted with CH₂Cl₂ (2 x 80 mL), dried (Na₂SO₄) and evaporated to yield **63a** (2.25 g) as a colorless oil.

20

step 3

A solution of **63a** (2.25 g; 5.704 mmol), NaCN (0.839 g; 17.12 mmol) and 20 mL of 90% aqueous EtOH was stirred at room temperature for 24 h. The solvent was evaporated and the residue partitioned between EtOAc (100 mL) and H₂O (100mL). The EtOAc phase was washed with H₂O and saturated brine. The organic extracts were dried (Na₂SO₄) and evaporated. The crude product was purified by silica gel chromatography and eluted with a hexane/EtOAc gradient (10:1 → 6:1) to yield 1.10 g (56.6%) of **63b** as a colorless oil.

step 4

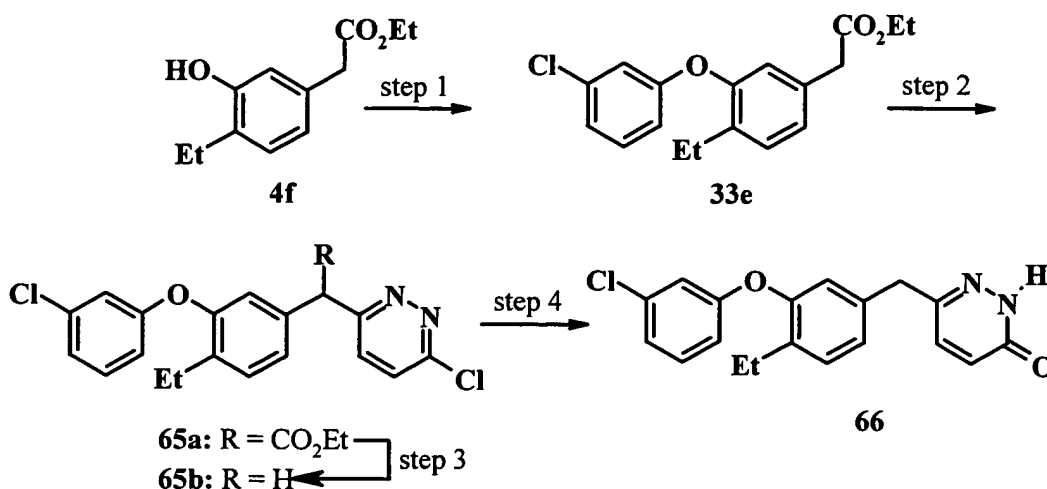
To a mixture of 1.00 g of **63b** (1.0 g; 2.936 mmol) 3,6-dichloropyridazine (0.96 g; 6.5 mmol) and 16 mL of DMF cooled to 0°C was added portionwise 0.30 g of NaH (7.5 mmol; 60% in mineral oil). The reaction mixture was stirred at 0°C for 2 h and allowed to warm to room temperature for 12 h. The reaction was poured into aqueous 10% NaHSO₄ and extracted with EtOAc (2 x 120 mL). The combined organic extracts were washed six times with H₂O, dried (Na₂SO₄), filtered and evaporated. The crude product was purified by silica gel chromatography and eluted with a hexane/EtOAc gradient (15:1 → 8:1) to provide **63c** as an orange oil (0.93 g; 70%).

step 5

A mixture of 0.93 g of **63c**, HOAc (10 mL), HCl (20 mL) and H₂O (10 mL) were heated at reflux for 8 h, cooled to rt and extracted with EtOAc. The combined extracts were dried (Na₂SO₄), filtered and concentrated. The crude product was purified by chromatography on silica gel and eluted with a gradient of CH₂Cl₂:EtOAc (8:1 → 4:1) to yield 40.56 g (67%) of **64a** as a white solid.

EXAMPLE 14

6-[3-chloro-phenoxy)-4-ethyl-benzyl]-2H-pyridazin-3-one



step 1

To a solution of ethyl 4-ethyl-3-hydroxyphenylacetate (**4f**; 1.0 g; 4.81 mmol), 3-chlorobenzeneboronic acid (1.56 g; 10.1 mmol), cupric acetate (0.96 g; 5.29 mmol), 4Å molecular sieves (5 g), and CH₂Cl₂ (48 mL) was added TEA (3.34 mL; 24.05 mmol) and the reaction was stirred for 4 days. The reaction mixture was filtered through a pad of CELITE®. The top layer containing the molecular sieves was removed and stirred with CH₂Cl₂ and refiltered. The combined organic filtrates were washed with 2N HCl, brine, dried (Na₂SO₄), filtered and evaporated. The crude product was chromatographed with silica gel and eluted with hexane/EtOAc (90%hexane/EtOAc) to yield **33e** (0.38 g; 25%).

step 2

To a solution of 0.34 g (1.07 mmol) of **33e**, 3,6-dichloropyrazine (0.30 g; 2.03 mmol) in 15 mL dry DMF cooled in an ice-water bath was added portionwise 0.107 g NaH (2.6 mmol; 60% in oil). The reaction was allowed to warm to ambient temperature and 2 h. The reaction was poured onto a mixture of ice, water and 10% sodium bisulfate. The mixture was extracted thoroughly with EtOAc and the combined extracts were washed with six times with water and then brine. The extract was dried (MgSO₄), filtered and evaporated and the residue chromatographed on silica gel and eluted with hexane:EtOAc (85:15) to yield **65a** (0.40 g; 87%).

step 3

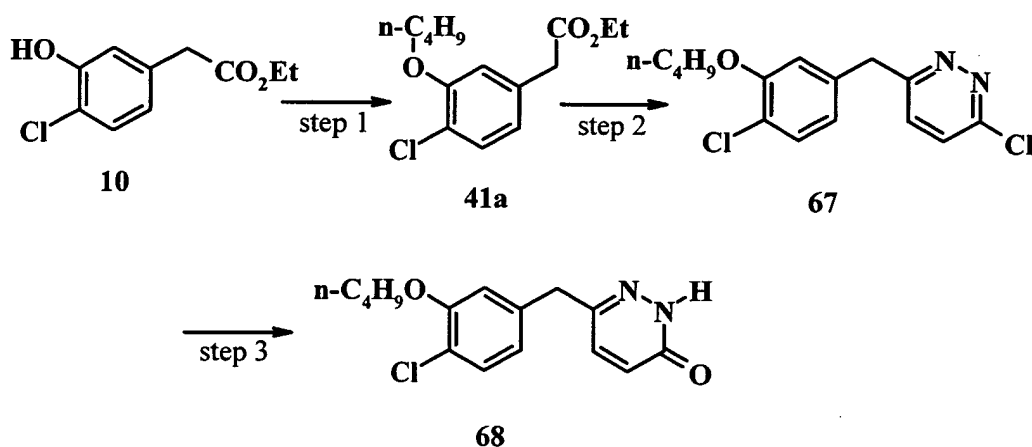
A mixture of **65a** (0.55 g; 1.27 mmol), HOAc (3 mL), HCl (6 mL) and H₂O (3 mL) were heated at reflux for 3 h, cooled to rt and thrice extracted with EtOAc. The combined extracts were washed with brine, dried (MgSO₄), filtered and concentrated *in vacuo*. The crude product was purified by chromatography on silica gel and eluted with a gradient of CH₂Cl₂:EtOAc (7:3 → 3:7) to yield **65b** (0.28 g).

step 4

A mixture of **65b** (0.35 g; 0.97 mmol), HOAc (8 mL) and NaOAc (110 mg) was heated at reflux for 2 h, cooled to rt and thrice extracted with EtOAc. The combined extracts were thrice washed with dilute NaHCO₃ and brine, dried (MgSO₄), filtered and concentrated *in vacuo*. The crude product was purified by chromatography on silica gel and eluted with a gradient of hexane:EtOAc (6:4) to yield **66** (0.0511 g).

EXAMPLE 15

6-(3-Butoxy-4-chloro-benzyl)-2H-pyridazin-3-one



step 1

Ethyl 4-chloro-3-hydroxyphenylacetate (**10**; 0.5 g; 2.33 mmol), *n*-butanol (0.43 mL; 4.66 mmol) and PPh₃ (0.92 g; 3.5 mmol) were dissolved in THF (8 mL) and cooled in an ice bath. DIAD was added dropwise to the cooled solution and the reaction mixture then allowed to warm to rt under N₂ atmosphere for 30 m. The THF was evaporated and the viscous yellow oil adsorbed onto silica gel and eluted with a hexane:EtOAc gradient (9:1 → 8:2) to yield **41a** (0.472 g; 75%).

step 2

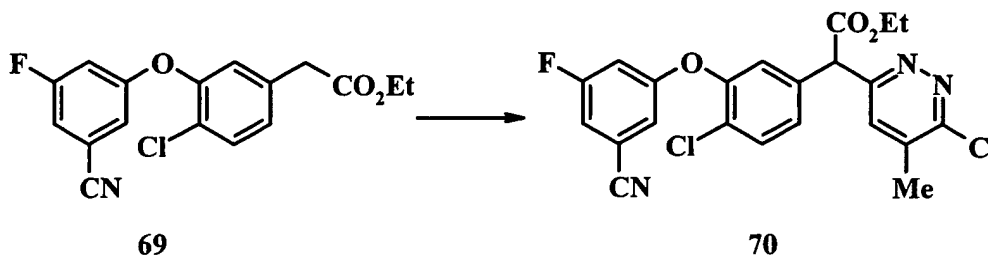
To a solution of **41a** (0.472 g; 1.75 mmol), 3,6-dichloropyrazine (0.7 g; 3.28 mmol) in 8 mL dry DMF cooled in an ice-water bath was added portionwise 0.160 g NaH (4.1 mmol; 60% in oil). The reaction was allowed to warm to ambient temperature and stirred overnight. The reaction was poured onto a mixture of ice, water and sodium bisulfate. The mixture was extracted thoroughly with EtOAc and the combined extracts were washed with water (six times) and brine. The extract was dried (MgSO₄), filtered and evaporated and the residue chromatographed on silica gel and eluted with hexane:EtOAc (8:2) to yield **67** (0.463 g).

step 3

A mixture of **67** (0.465 g), HOAc (5 mL), HCl (10 mL) and H₂O (50 mL) were heated at reflux for 1 h, cooled to rt and extracted with EtOAc. The combined extracts were washed with H₂O, dried (Na₂SO₄), filtered and concentrated *in vacuo*. The crude product was purified by chromatography on silica gel and eluted with a gradient of CH₂Cl₂:EtOAc (7:3 → 3:7) to yield **68** (0.28 g).

EXAMPLE 16

[4-chloro-3-(3-cyano-5-fluoro-phenoxy)-phenyl]-(6-chloro-5-methyl-pyridazin-3-yl)-acetic acid ethyl ester

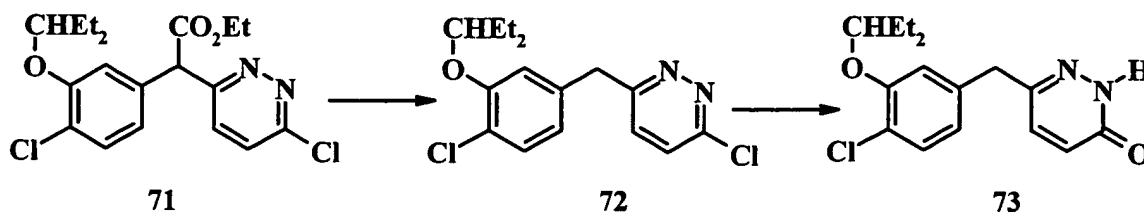


To a solution of **69** (3.9 g), 0.7 g of 3,6-dichloro-4-methylpyrazine in 40 mL dry DMF cooled in an ice-water bath was added portionwise 0.33 g NaH (60% in oil). The reaction was allowed to warm to ambient temperature and stirred for one hour. The reaction was poured onto a mixture of ice, water and sodium bisulfate. The mixture was extracted thoroughly with EtOAc and the combined extracts were washed with 5% LiCl, water (six times) and brine. The extract was dried (MgSO₄), filtered and evaporated and the residue chromatographed on silica gel and eluted with hexane:EtOAc (7:3) to yield **70** (0.65 g)

Hydrolysis to the pyridazinone was carried out as described in step 2 of Example 18.

EXAMPLE 17

6-[4-chloro-3-(1-ethyl-propoxy)-benzyl]-2H-pyridazin-3-one



5 step 1

To a solution of **71** (1.2 g; 3.02 mmol) in MeOH (5 mL) and water (1.5 mL) was added LiOH·H₂O (0.178 g; 4.23 mmol) and stirred for 30 m. The reaction mixture was diluted with EtOAc, and washed with 2N HCl, water and brine. The organic extract was dried (Na₂SO₄), filtered and evaporated to yield **72** which as used in the next step without additional purification.

10

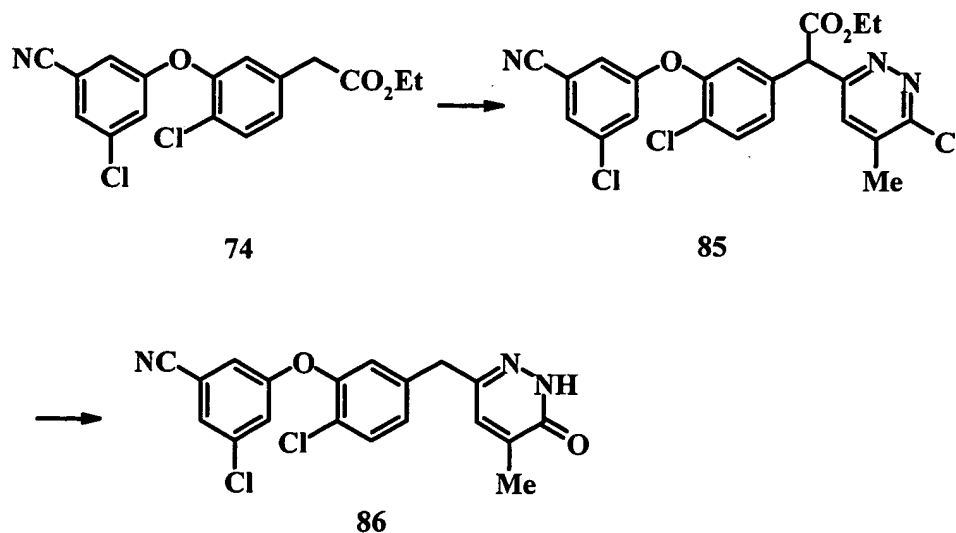
step 2

To a solution of crude **72** (0.80 g) from the previous step and HOAc (15 mL) was added NaOAc (0.40 g) and the reaction heated to reflux for 1 h. The reaction mixture was cooled to rt and diluted with EtOAc, The organic phase was washed with water and brine, dried (Na₂SO₄) and evaporated. The crude product was purified by silica gel chromatography and eluted with a hexane:EtOAc gradient (7:3 → 6:4) to yield pyridazinone **73** (0.350 g).

15

EXAMPLE 18

3-Chloro-5-[2-chloro-5-(5-methyl-6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile



20

step 1

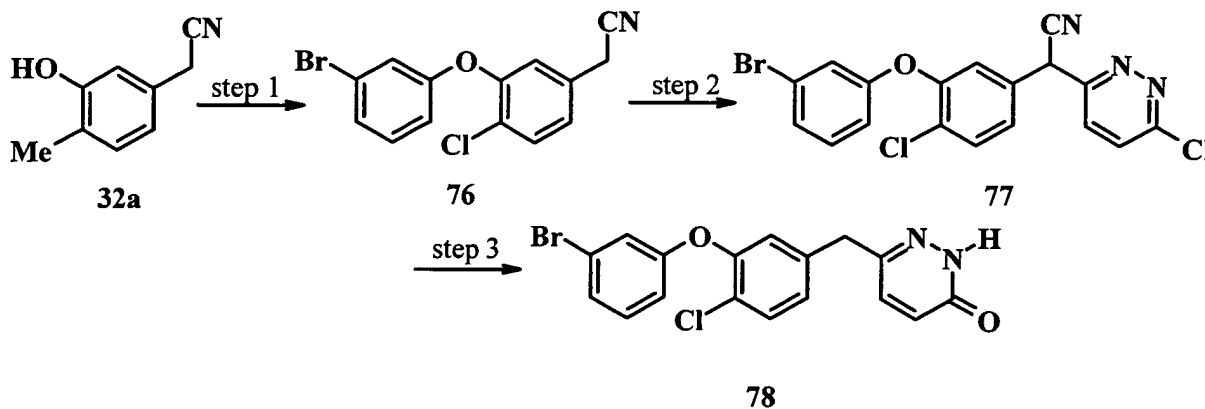
To a ice-cold solution of **74** (1.30 g; 3.712 mmol), 5-methyl-1,4-dichloropyrazine (1.21 g; 7.42 mmol) and DMF (16 mL) under an nitrogen atmosphere was added in one portion NaH (0.297 g; 7.43 mmol; 60% in mineral oil). The reaction mixture was stirred at 0°C for 10 minutes and then allowed to warm to room temperature and stir for 6 h. The mixture was diluted with 75 mL EtOAc and poured into about 100 mL of ice-water containing a small quantity of NaHSO₄. The aqueous layer was separated and washed with 50 mL of EtOAc. The combined organic phases were dried (Na₂SO₄), filtered and evaporated. The residue was purified by silica gel chromatography and eluted with a hexane:EtOAc gradient (12:1 → 8:1) to yield **85** as a yellow oil which partially solidified (350 mg; 20%). A small quantity of the isomeric compound was isolated after elution of the major product.

step 2

A mixture of **85**, LiOH (1.47 mL; 0.734 mmol; 0.5M in MeOH/H₂O) and degassed MeOH/H₂O (4:1 v/v; 8mL) was stirred at 0°C for 5 minutes then allowed to stir 12 h at room temperature under an N₂ atmosphere. The reaction mixture was carefully acidified with 1 N HCl and partitioned between 25 mL of EtOAc and 50 mL of water. The EtOAc layer was dried (Na₂SO₄), filter and evaporated. The crude residue was dissolved in 18 mL of HOAc and 900 mg of NaOAc and refluxed for 3 h under an N₂ atmosphere. The reaction mixture was cooled, diluted with 50 mL of EtOAc and poured into H₂O. The organic layer was dried (Na₂SO₄), filtered and evaporated and the crude product was purified by silica gel chromatography and eluted with Ch₂Cl₂/EtOAc to yield a yellow solid **86** (0.120 g; 42%)

EXAMPLE 19

6-[3-Bromo-phenoxy)-4-chloro-benzyl]-2H-pyridazin-3-one



step 1

To a flask was charged with 3-hydroxy-4-methylphenylacetonitrile (**32a**; 0.92 g; 6.2 mmol), Cu(OAc)₂ (1.3 g; 6.9 mmol), 3-bromobenzenboronic acid (1.1 g; 5.5 mmol) and powdered 4Å molecular sieves, was added CH₂Cl₂ (62 mL) followed by pyridine (2.5 mL; 31 mmol). The reaction was stirred at rt for 3 days. The suspension was filtered through a bed of CELITE®/silica gel and the solid washed with
5 CH₂Cl₂. The combined filtrates were washed sequentially with 2N HCl (2 x 25 mL), NaHCO₃ (25mL), water and brine. The extracts were dried (MgSO₄), filtered and evaporated. The crude product **76** was sufficiently pure to use in the next step.

step 2

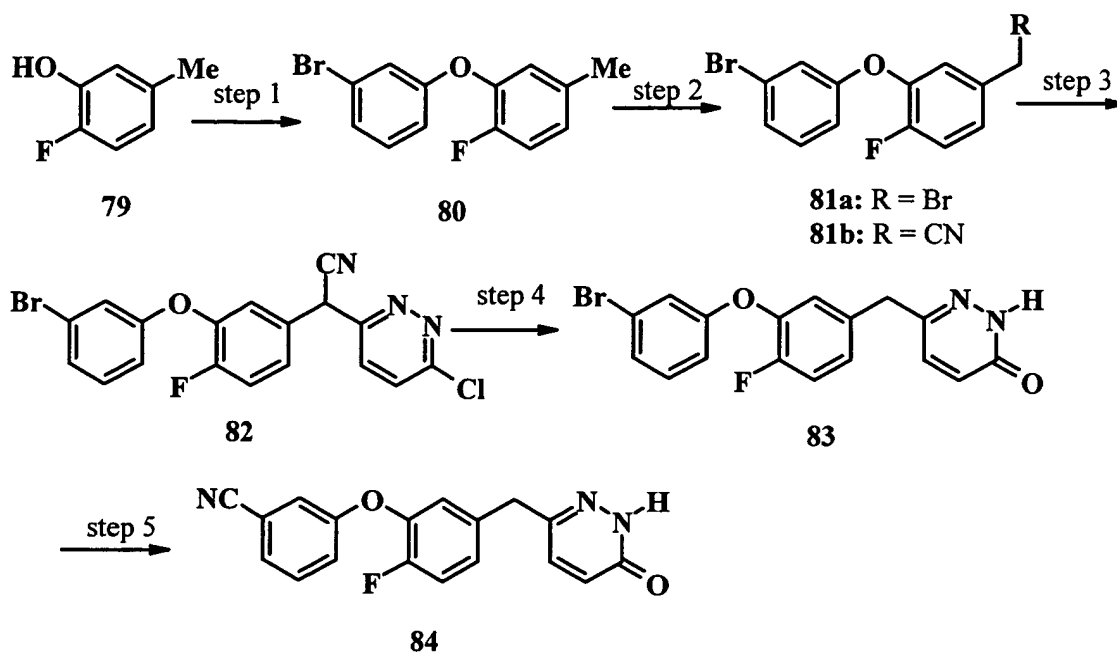
10 To an ice-cold solution of **76** (0.8028 g; 3.6 mmol), 3,6-dichloropyrazine (0.54 g; 3.6 mmol) and 17 mL of dry DMF was added portionwise NaH (0.3 g; 7.6 mmol; 60% in mineral oil). The reaction was stirred at rt for 2 h and poured into 75 mL of H₂O containing about 1 g of NaHSO₄. The aqueous solution as thoroughly extracted with EtOAc and the combined EtOAc extracts washed with six times with water, then with brine, dried (Na₂SO₄), filtered and evaporated. Nitrile **77** was purified by silica
15 gel chromatography and eluted with 25% EtOAc:hexane.

step 3

A solution of **77** (1.0 g; 2.4 mmol), HOAc (4 mL), HCl (8 mL) and water (4 mL) was heated at 100 °C for 12 h. The reaction mixture was cooled and diluted with EtOAc and washed three times with water and
20 once with brine. The organic phase was dried (MgSO₄), filtered and evaporated. The crude product was purified by chromatography on silica and eluted with an EtOAc:hexane gradient (25:75 → 75:25) to yield **78** (62 mg).

EXAMPLE 20

3-[2-Fluoro-5-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-benzonitrile



5

step 1

To a solution of 2-fluoro-4-methylphenol (**79**; 3.0 g; 24 mmol), 3-bromobenzenboronic acid (5.3 g; 24 mmol), cupric acetate (4.8 g; 23.1 mmol), 4Å molecular sieves (15 g) and CH₂Cl₂ (240 mL) was added TEA (17 mL; 120 mmol) and the reaction was stirred for 4 days. The molecular sieves were filtered and washed well with CH₂Cl₂. The combined organic filtrates were washed with 2N HCl, brine, 2N NaOH, water and brine, dried (Na₂SO₄), filtered and evaporated. The crude product was chromatographed with silica gel and eluted with hexane:EtOAc (90%hexane:EtOAc) to yield **80** (5.7 g; estimated purity 72%).

10

step 2

A solution of **80** (4.1 g; 14.6 mmol), NBS (2.6 g; 14.6 mmol), AIBN (0.25 g; 1.50 mmol) and 146 mL of CCl₄ was heated at reflux for 5.0 h, cooled to rt and the precipitated succinimide filtered through a pad of CELITE®. The filtrate was evaporated and the crude product **81a** was sufficiently pure to use in the next step.

15

The crude bromomethyl compound **81a** from the previous step was dissolved in 73 mL of 90% aq. EtOH and 2.5 g of NaCN (49.01 mmol) was added. The reaction mixture was stirred overnight at rt. The solid material was filtered through a pad of CELITE® and the filtrate was evaporated. The crude

20

product purified by silica gel chromatography and eluted with 30% EtOAc:hexane to yield the nitrile **81b** (2.4 g; 54%).

step 3

- 5 To a ice-cold solution of **81b** (2.4 g; 7.4 mmol), 3,6-dichloropyrazine (2.2 g; 14.4 mmol) and 74 mL of dry DMF was added portionwise NaH (0.62 g; 15.5 mmol; 60% in mineral oil). The reaction was stirred at 0°C for 15 min and allowed to stir at rt for 2 h and poured into 75 mL of H₂O containing about 1 g of NaHSO₄. The aqueous solution as thoroughly extracted with EtOAc and the combined EtOAc extracts washed with six times with water, once with 5% LiCl, dried (Na₂SO₄), filtered and evaporated.
- 10 The product was purified by silica gel chromatography to yield **82** (2.9 g; 94%).

step 4

- A mixture of **82** (0.209 g; 2.6 mmol), HOAc (10 mL), HCl (20 mL) and H₂O (10 mL) were heated at reflux for overnight, cooled to rt, diluted with water and EtOAc. The aqueous phase was washed with
- 15 EtOAc and the combined extracts were washed sequentially with water and sat'd NaHCO₃, dried (Na₂SO₄), filtered and evaporated to yield **83** (2.0 g; 77%).

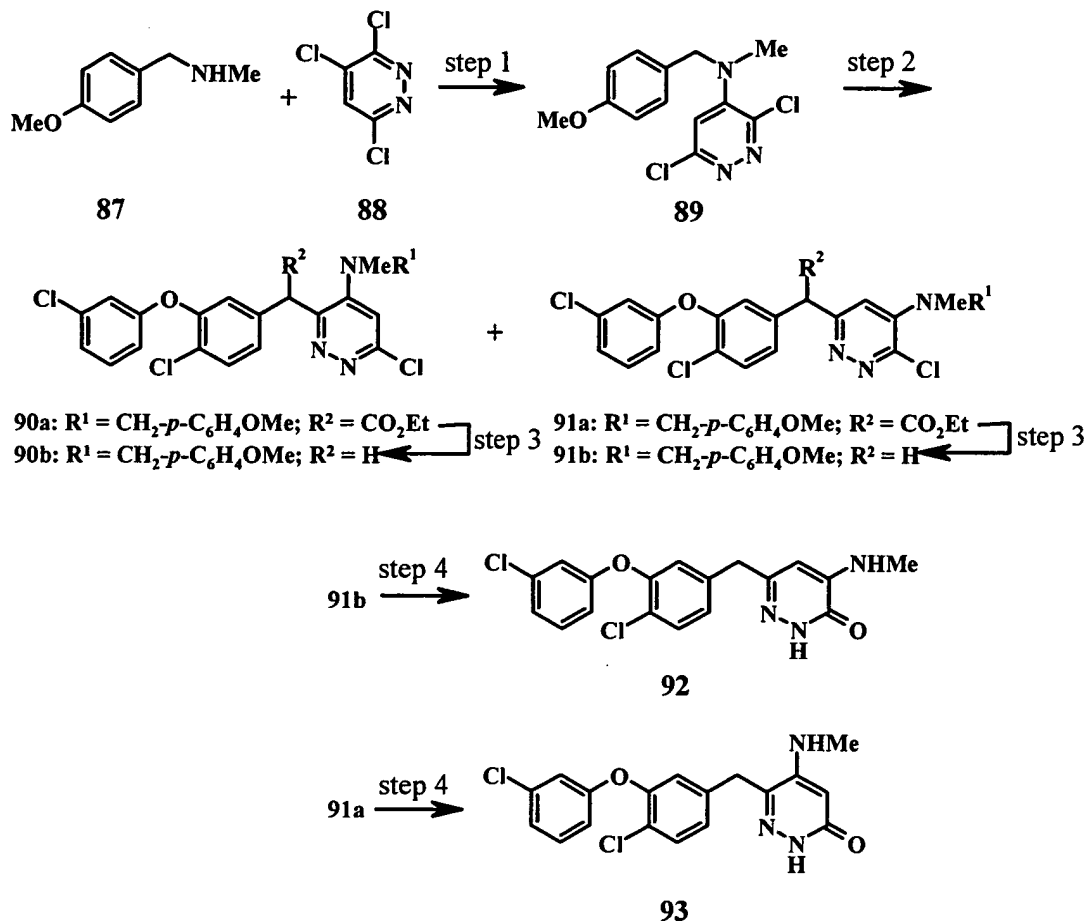
step 5

- A solution of **83** (0.750 g; 0.20mmol) and CH₃CN (10 mL) was degassed for 10 m with a stream of N₂
- 20 and Zn(CN)₂ (0.14 g; 1.2 mmol) and Pd(PPh₃)₄ (0.35 g; 0.30 mmol) and CuI (0.039 g; 0.2 mmol) were added. The reaction mixture was heated to 120 °C for 12 h, cooled to rt and diluted with EtOAc. The resulting suspension was filtered through a bed of CELITE®, the solids washed with EtOAc and the combined organic filtrates evaporated. The crude product was purified by silica gel chromatography to yield nitrile **84** (0.060 g).

EXAMPLE 21

6-[4-Chloro-3-(3-chloro-phenoxy)-benzyl]-4-methylamino-2H-pyridazin-3-one (**92**) and 6-[4-Chloro-3-(3-chloro-phenoxy)-benzyl]-5-methylamino-2H-pyridazin-3-one (**93**)

5



step 1

A solution of *p*-methoxybenzylmethylamine (1.11 g; 7.34 mmol) and EtOH (10 mL) was added to a solution of 3,4,6-trichloropyridazine (1.28 g; 6.99 mmol), TEA (0.983 mL; 6.99 mmol) and anhydrous EtOH (20 mL) and the reaction stirred overnight at room temperature. The solvent was evaporated and the residue partitioned between EtOAc and H₂O. The phases were separated, the water washed with EtOAc, and the combined EtOAc solutions washed with water and brine. The solution was dried (MgSO₄), filtered and evaporated. The product was purified by silica gel chromatography and eluted with an EtOAc:hexane mixture (5:95 → 30:70) to yield **89** as a white solid (1.3 g; 62%).

15

step 2 and step 3

The condensation of **89** and [4-Chloro-3-(3-chloro-phenoxy)-phenyl]-acetic acid ethyl ester to yield **90a** and **91a** was carried out as described in step 4 of Example 14 and the subsequent saponification and decarboxylation step to yield **90b** and **91b** was carried out as described in step 1 of Example 18. The product was a mixture of isomers. The isomers were separated by silica gel chromatography and eluted with an EtOAc:hexane gradient (10:90 → 50:50). The first isomer to elute was **90b**.

step 4

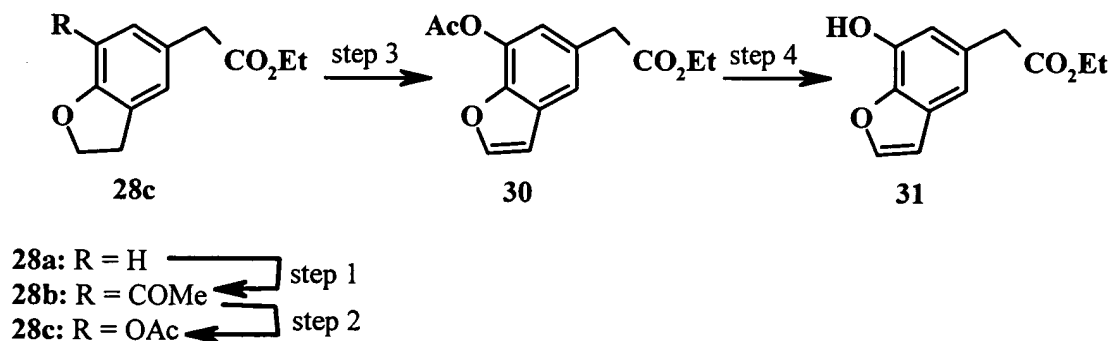
A solution of **90b**, NaOAc (0.25 g) and HOAc (5 mL) was heated for 8 h. The solution was cooled and the solvent removed *in vacuo*. The residue was partitioned between EtOAc and H₂O. The phases were separated and the H₂O phase washed with EtOAc. The combined organic layers were washed with brine, dried and evaporated. The residue was purified by silica gel chromatography and eluted with a MeOH:CH₂Cl₂ gradient (1:99 → 10:90) to yield 16 mg of **93**.

The other isomer was prepared in identical manner except the product was purified by silica gel chromatography and eluted with a EtOAc:hexane gradient (30:70 → 70:30) to yield 64 mg of **92**.

The overall yield of steps 2-4 was 28%.

EXAMPLE 22

(7-Hydroxy-benzofuran-5-yl)-acetic acid ethyl ester



step 1

To a solution of **28a** (5.0g; 24.2 mmol) and anhydrous CH₂Cl₂ (75 mL) was added sequentially acetyl chloride ((2.42 mL; 33.9 mmol) and SnCl₄ (5.39 mL; 46.1 mmol; 1 M solution in CH₂Cl₂). The

reaction was stirred at room temperature for 50 minutes and poured into a mixture of ice and 2 N HCl (200 mL). The organic phase was separated and diluted with about 50 mL of CH₂Cl₂ and thrice washed with water (100 mL) and once with brine (100 mL). The organic phase was dried (MgSO₄), filtered and evaporated to yield **28b** (6.0 g) which contained about 10% of **28a**. The crude product was used without further purification.

step 2

To an ice-cold solution of **28b** (6.01 g; 24.2 mmol) and CH₂Cl₂ (100 mL) under a nitrogen atmosphere was added sequentially a solution of MCPBA (11.9 g; 48.4 mmol) and CH₂Cl₂ (12 mL) followed by TFA (2.14 mL; 27.8 mmol). The reaction mixture was stirred at rt overnight. The reaction mixture was cooled to 0°C and a 5% aqueous Na₂SO₃ solution (150 mL) was added slowly with stirring. The mixture was stirred for 5 minutes after addition was completed and precipitated *m*-chlorobenzoic acid was filtered. The solid was washed with CH₂Cl₂ and the combined filtrates were washed with 10% NaOH (2 x 250 mL), 2 N HCl (200 mL), water and brine. The resulting solution was dried (MgSO₄), filtered through a pad of CELITE and concentrated *in vacuo* to yield **28c** (4.1 g).

step 3

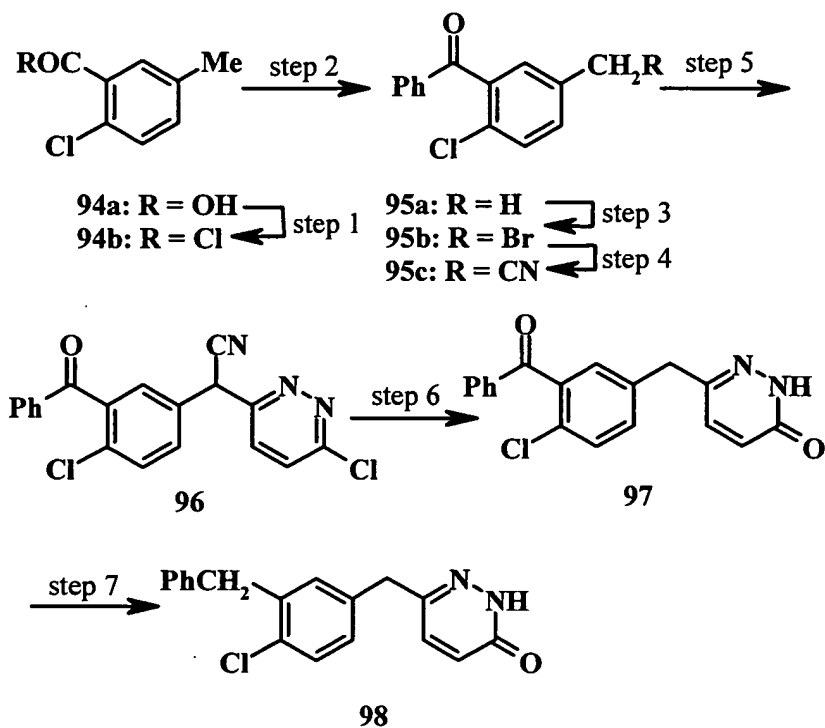
To a solution of dihydrofuran derivative **28c** (14.6 g; 0.0553 mol) and CCl₄ (500 mL) was added NBS (10.3 g; 0.0580 mol) and AIBN (1.4 g). The reaction was heated to reflux for 30 minutes under a nitrogen atmosphere. The reaction was cooled, the solid succinimide filtered, and the organic phase was washed with 0.5 M NaHSO₄ (150 mL) and brine. The product was dried (Na₂SO₄), filtered and evaporated to yield 15.2 g of a yellow syrup. The crude product was purified by silica gel chromatography and eluted with a EtOAc:hexane gradient (3:97 → 10:90) to yield 10.3 g (78.1%) of **30**.

step 4

A solution of **30** (10.3 g; 39.3 mmol), EtOH (250 mL) and saturated NaHCO₃ (100 mL) were heated to reflux for 1 h. The reaction mixture was cooled to room temperature and the EtOH removed *in vacuo*. Ice was added to the residue aqueous solution and the reaction carefully acidified to about pH 2 with 2 N HCl. The resulting mixture was extracted with EtOAc (2 x 300 mL) and the combined organic phase washed with brine, dried (Na₂SO₄), filtered and evaporated to yield a brown oil (8.8 g). The crude product was run through a silica gel column with 15% EtOAc:hexane to yield **31** (5.44g; 62.9%) as a white solid.

EXAMPLE 23

6-(3-Benzyl-4-chloro-benzyl)-2H-pyridazin-3-one



5

step 1

To a ice-cold solution of **94a** (4.0 g; 23.44 mmol) and CH_2Cl_2 (60 mL) was added oxalyl chloride (10.26 mL; 0.117 mol) and one drop of DMF. The reaction mixture was allowed to warm to rt and stirred overnight. The volatile solvents were removed *in vacuo* and oxalyl chloride removed by thrice adding CH_2Cl_2 (30 mL) and re-evaporating the solvent to yield **94b** (4.4 g) as a yellow oil which was used directly in step 2.

step 2

The acid chloride **94b** from step 1 (4.4g; 23.27 mmol) was dissolved in benzene (80 mL) and the solution cooled to 0°C . Solid AlCl_3 was added in portions to the solution and after the addition was complete the reaction was warmed to room temperature and stirred for two days. The reaction mixture was poured into ice containing a small quantity of con HCl and the aqueous mixture extracted with CH_2Cl_2 which was washed sequentially with water, 2N HCl and brine. The resulting solution was dried (MgSO_4) filtered and evaporated to yield an orange oil which was purified by silica gel chromatography and eluted with a EtOAc:hexane (1:20) to yield 3.46 g (64%) of **95a** as an orange oil.

step 3

A solution of **95a** (3.46 g; 0.15 mol), NBS (2.67 g; 0.015 mol), AIBN (0.27 g; 0.0015 mol) and 60 mL of CCl₄ (60 mL) was heated at reflux overnight. The reaction was cooled to room temperature and filtered.

- 5 The filtrate was diluted with 50 mL of CH₂Cl₂, washed with H₂O (100 mL) and brine (100 mL), dried (Na₂SO₄), filtered and evaporated. The crude product was purified by silica gel chromatography and eluted with a EtOAc:hexane (1:20) to yield 1.23 g (29%) of **95b** as an oil.

step 4

- 10 A solution of **95b** (1.23 g; 3.98 mmol), NaCN (0.293 g; 5.97 mmol) and 8 mL of DMSO was stirred at room temperature for two days. The reaction mixture was diluted with EtOAc (100 mL), washed with water and brine, dried (Na₂SO₄), filtered and evaporated to yield **95c** (0.93 g) sufficiently pure to carry onto the next step.

15 step 5

- To a ice-cold solution of **95c** (0.930 g; 3.63 mmol), 3,6-dichloropyrazine (1.08 g; 7.266 mmol) and 8 mL of dry DMF was added portionwise NaH (0.303 g; 9.082 mmol; 60% in mineral oil). The reaction was stirred at 0°C for 15 min and allowed to stir at rt for 24 h and poured into an ice cold solution of 144 mL of H₂O containing about 9 g of NaHSO₄. The aqueous solution as thrice extracted with EtOAc
20 (50 mL). The combined extracts were with three times with water, twice with brine, dried (Na₂SO₄), filtered and evaporated to yield 2.48 g of a brown oil which was . chromatographed on silica gel and eluted with EtOAc:hexane (3:1) to yield **96** (0.744 g; 56%).

step 6

- 25 A mixture of **96** (0.744 g; 2.02 mmol), HOAc (8 mL), HCl (16 mL) and H₂O (8 mL) was heated at reflux for 14 h,, cooled to rt partitioned between water and EtOAc. The aqueous phase was thrice extracted with EtOAc and the combined extracts were washed with water, dried (Na₂SO₄), filtered and evaporated to yield **97** (0.356 g; 55%).

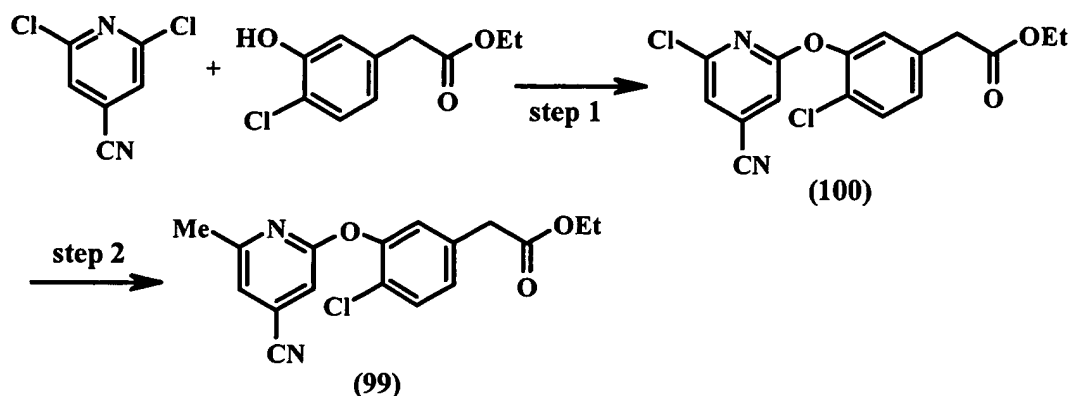
step 7

- 30 To an ice-cold solution of **97** (0.529 g; 1.633 mmol) and TFA (8 mL) was added triethylsilane (0.8 mL; 4.898 mmol) and the resulting mixture was stirred at 0°C for 2 h then at room temperature overnight. The TFA was remove *in vacuo* and the residue diluted with EtOAc, washed with water and brine, dried (MgSO₄), filtered and evaporated. The crude product was purified by silica gel chromatography and

eluted with a CH₂Cl₂:EtOAc (2:1 → 1:1) to yield **98** (0.250 g; 49%) as a yellow oil. The product was maintained at 35° C in a vacuum oven for final drying.

EXAMPLE 24

[3-(6-Methyl-4-cyano-pyridin-2-yloxy)-4-chloro-phenyl]-acetic acid ethyl ester (**99**)



step 1

A 100 mL round bottom flask was charged with 2,6-dichloro-4-cyanopyridine (2.50 g, 14.45 mmol), ethyl (4-chloro-3-hydroxyphenyl)-acetate (3.10 g, 14.45 mmol), and anhydrous K₂CO₃ (2.10g, 15.20 mmol) under nitrogen. DMA (50 mL) was added *via* syringe, and the heterogeneous mixture was heated to 100° C for 2 h. The solution was cooled to RT, poured into 2M NH₄Cl (150 mL), and extracted with EtOAc (3 x 50 mL). The combined organics were washed with brine (50 mL), dried with MgSO₄, and the volatile materials were evaporated. The crude product was purified by flash chromatography (SiO₂, 0% to 35% EtOAc/Hexanes) to afford 2.30 g (45% yield) of **100**

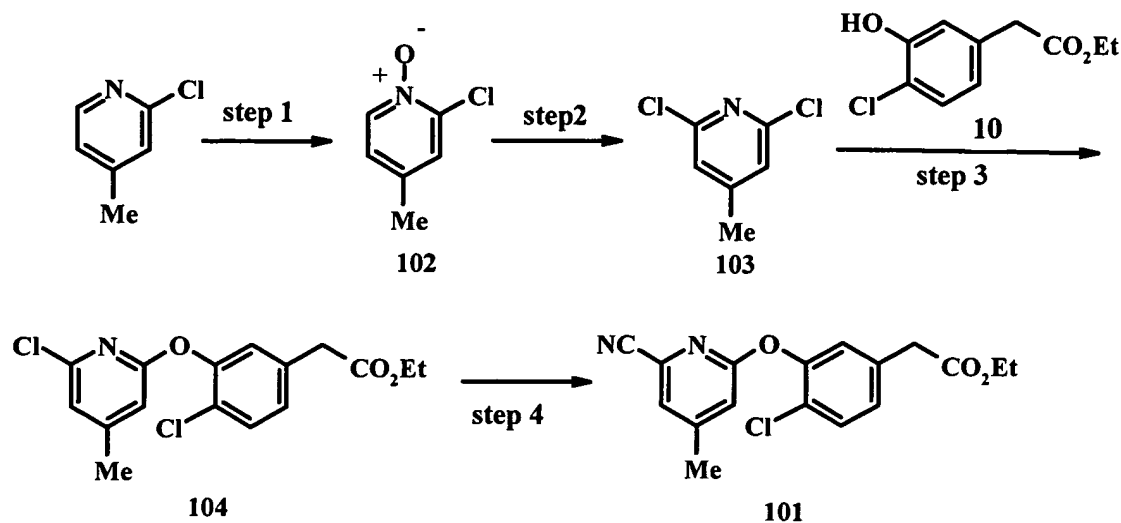
step 2

To a stirred solution of **100** (2.12 g, 6.05 mmol), dichlorobis(triphenylphosphine) palladium(II) (425 mg, 0.61 mmol), and 2-dimethylaminoethanol (122 µL, 1.21 mmol) cooled to 0° C was added dimethylzinc (6.05 mL of a 2.0 M solution in toluene, 12.10 mmol). The mixture was warmed to RT, heated to 85° C for 1.5 h, and then cooled to RT. This solution was then slowly added to 2 M NH₄Cl (100 mL) cooled to 0° C. The resulting mixture was extracted with EtOAc (3 x 50 mL), and the combined organics were washed with brine (50 mL), dried with MgSO₄, and evaporated. The crude oil was purified by flash chromatography (SiO₂, 0% to 20% EtOAc/hexanes) to afford 1.25 g (62%) of [3-(6-chloro-4-cyano-pyridin-2-yloxy)-4-ethyl-phenyl]-acetic acid ethyl ester (**99**) as an oil that slowly crystallized.

The introduction of the pyridazinone ring and saponification and decarboxylation of the ester was accomplished utilizing the procedure described in steps 2-4 of Example 14.

EXAMPLE 25

5 [3-(6-Cyano-4-methyl-pyridin-2-yloxy)-4-ethyl-phenyl]-acetic acid ethyl ester (101)



step 1

A solution of 2-chloro-4-methylpyridine (5.00g, 39.2 mmol), MCPBA (9.02g; assay 75% pure, 39.2 mmol), and chloroform (80 mL) was heated to reflux for 5 h. The reaction mixture was cooled to RT, and approximately 20 mL of the solvent was evaporated (without heating). The benzoic acid that precipitated from the solution was removed by filtration. The remaining filtrate was washed with saturated aqueous K₂CO₃ (50 mL), 1 M NaOH (50 mL), brine (25 mL), and dried with anhydrous Na₂SO₄. The solvent was evaporated, and the remaining material was purified by flash chromatography on silica gel (0% to 5% MeOH/CHCl₃) to provide 3.60 g (64%) of 102

step 2

A 50 mL round bottom flask was charged with 2-chloro-4-methylpyridine-N-oxide, and 20 mL of POCl₃ was slowly added. (WARNING!) A sudden exotherm occurred (the N-oxide should be added slowly to the POCl₃). An additional 10 mL of POCl₃ was added to the brown reaction mixture, and the solution was heated to 95°C for 5 h. The reaction mixture was cooled to RT, and the volatile materials were evaporated. The remaining mixture was slowly added to a saturated aqueous solution of NaHCO₃ (150 mL), and extracted with EtOAc (3 x 50 mL). The combined organic fractions were washed with brine (50 mL) and dried over MgSO₄. The solvent was evaporated, and the remaining brown solid was

purified by flash chromatography on silica gel (0% to 7% EtOAc/ hexanes) to provide 1.56 g (43%) of **103**.

step 3

- 5 A solution of 2,6-dichloro-4-methylpyridine (1.04 g, 6.42 mmol), ethyl (4-chloro-3-hydroxy-phenyl)-acetate (1.38 g, 6.42 mmol), and anhydrous Cs_2CO_3 (2.20 g, 6.74 mmol) in DMA (15 mL) under nitrogen was heated to 120° C for 15 h. The solution was cooled to RT, and poured into a saturated aqueous solution of NH_4Cl (50 mL). The resulting mixture was extracted with 1:1 EtOAc/hexanes (3 x 50 mL). The combined organic fractions were washed with water (3 x 30 mL), brine (30 mL), and dried
- 10 with anhydrous MgSO_4 . The volatile materials were evaporated and the remaining oil was purified by flash chromatography on silica gel (0% to 10% EtOAc/hexanes to provide 975 mg (45%) of **104**.

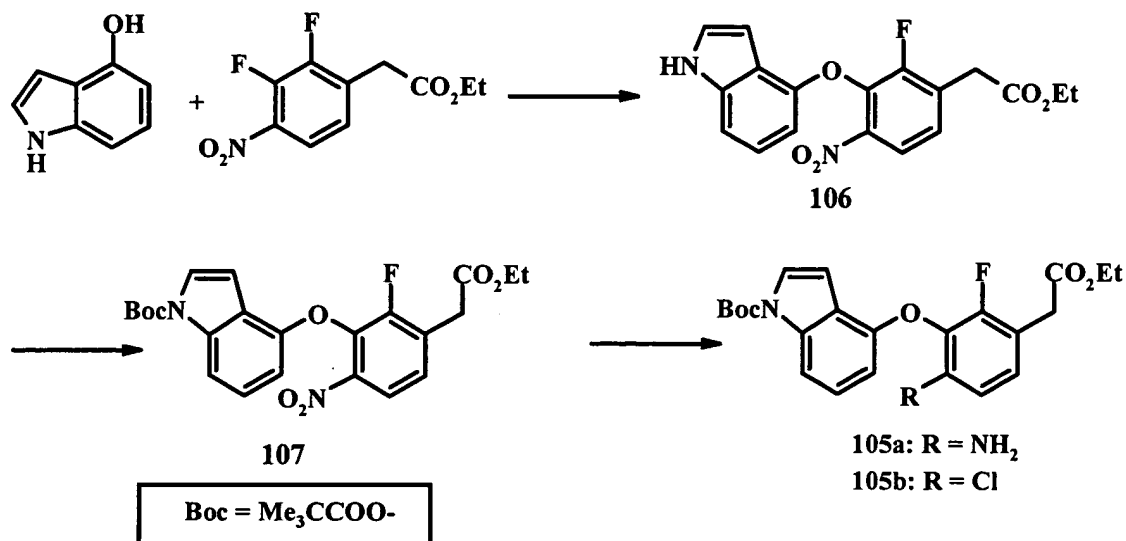
step 4

- A mixture of **104** (1.01 g, 2.97 mmol), zinc cyanide (209 mg, 1.78 mmol), zinc dust (116 mg, 1.78
- 15 mmol), bis(diphenylphosphino)ferrocene (329 mg, 0.59 mmol), $\text{Pd}(\text{dba})_2\text{-CHCl}_3$ (307 mg, 0.30 mmol) and in DMA (7 mL) under nitrogen was heated to 120° C for 4 h. The mixture was cooled to RT, and diluted with 1:1 EtOAc/hexanes (150 mL). This solution was washed with saturated aqueous NH_4Cl (2 x 25 mL), water (25 mL), brine (25 mL), and dried with anhydrous MgSO_4 . The solvents were
- 20 evaporated, and the remaining material was purified by flash chromatography on silica gel (0% to 25% EtOAc/hexanes) to provide 730 mg of **101** (74%).

The introduction of the pyridazinone ring and saponification and decarboxylation of the ester was accomplished utilizing the procedure described in steps 2-4 of Example 14.

EXAMPLE 26

4-(6-Chloro-3-ethoxycarbonylmethyl-2-fluoro-phenoxy)-indole-1-carboxylic acid tert-butyl ester



step 1

- 5 Solid sodium *tert*-butoxide (400 mg, 4.16 mmol) was added to a solution of 7-hydroxyindole (610 mg, 4.58 mmol) in THF (5 mL) under a nitrogen atmosphere, and the mixture was stirred for 10 m. A solution of ethyl 2,3-difluoro-4-nitrophenylacetate (1.02 g, 4.16 mmol in 5 mL of anhydrous THF) was added to the solution of the phenoxide *via* syringe. The resulting purple mixture was stirred overnight at RT, and then poured into a mixture of H₂O (30 mL) and brine (30 mL). The mixture was extracted with
- 10 2:1 EtOAc/hexanes (3 x 40 mL), and the combined organics were washed with H₂O (3 x 30 mL), brine (20 mL), and dried over anhydrous MgSO₄. The solvents were evaporated, and the remaining material was purified by flash chromatography on silica gel (0% to 40% EtOAc/hexanes) to afford 106 in 82% yield.

15 step 2

- tert*-Butoxycarbonylanhydride (533 mg, 2.44 mmol) was added to a solution of 106 (878 mg, 2.44 mmol) in anhydrous THF (12 mL) under a nitrogen atmosphere. The solution was cooled to 0° C, and 4-dimethylaminopyridine (30 mg, 0.24 mmol) was added. After 0.5 h, the solution was warmed to RT and stirred for an additional 2 h. The mixture was poured into H₂O (25 mL) and extracted with EtOAc
- 20 (3 x 30 mL). The combined organic fractions were washed with brine (20 mL), and dried over anhydrous MgSO₄. The volatile materials were evaporated, and the remaining oil was purified by flash chromatography on silica gel (0% to 25% EtOAc/hexanes) to provide 560 mg (50%) of the protected indole 107 and 260 mg (30%) of recovered starting material.

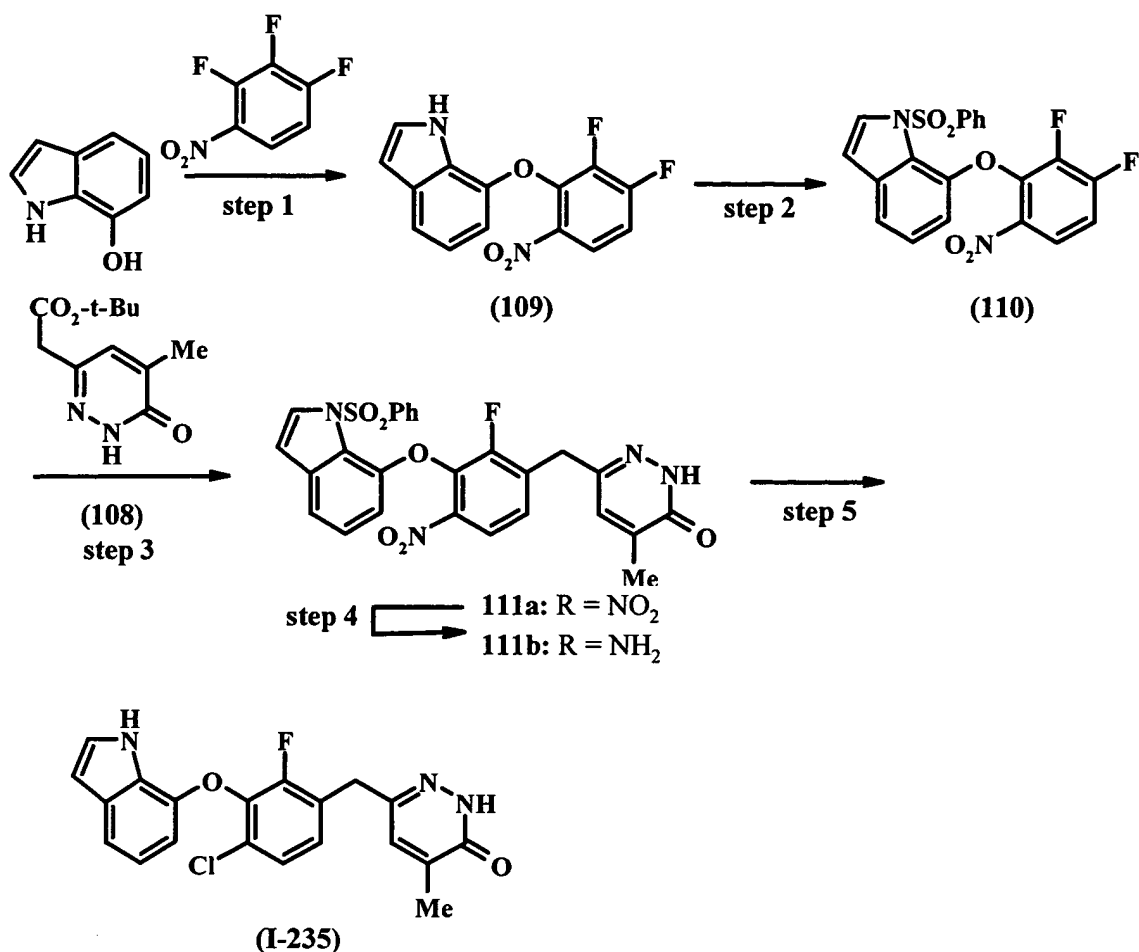
step 3

Protected indole **107** (790 mg, 1.72 mmol), 5% palladium on carbon (79 mg), and EtOH (15 mL) were added to a thick-walled bottle. The bottle was evacuated and pressurized with 50 psi of H₂ gas. After 4 hrs, the pressure was released, and the mixture was filtered through CELITE®. The EtOH was
5 evaporated, and the remaining oil was purified by flash chromatography on silica gel (0% to 25% EtOAc/hexanes) to afford **105a** 609 mg (83%).

Conversion of the amine **105a** to the corresponding chloride **105b** was effected with CuCl₂ and *tert*-butyl nitrite as described in step 3 of Example 27 (for the one step preparation of aryl chlorides and
10 bromides from aryl amines see Doyle *et al.*, *J. Org Chem.* **1977** 42:2426). The introduction of the pyridazinone ring and saponification and decarboxylation of the ester was accomplished utilizing the procedure described in steps 2-4 of Example 14. Deprotection of the Boc protecting group occurred spontaneously during the pyridazine hydrolysis.

EXAMPLE 27

6-[4-Chloro-2-fluoro-3-(1H-indol-7-yloxy)-benzyl]-4-methyl-2H-pyridazin-3-one (I-235)



5 step 1

Solid sodium *tert*-butoxide was added to a ice cold solution of 4-hydroxyindole (1.23 g, 9.24 mmol; *Synthetic Communications* **2003** 33:507) in anhydrous THF (145 mL) under a nitrogen atmosphere. The mixture was stirred for 10 m, and 2,3,4-trifluoronitrobenzene (1.06 mL, 9.24 mmol) was added dropwise. The brown solution was stirred for 2 h, and then added to a saturated aqueous solution of

10 NH₄Cl (150 mL). The aqueous layer was extracted with EtOAc (3 x 100 mL), and the combined organic fractions were washed with H₂O (100 mL), brine (75 mL), and dried over anhydrous MgSO₄. The solvents were evaporated, and the remaining oil was purified by flash chromatography on silica gel (0% to 30% EtOAc/hexanes) to afford 2.26 g (84%) of **109**.

step 2

Phenyl sulfonyl chloride (1.05 mL, 8.18 mmol), powdered NaOH (4 g), and Bu₄NHSO₄ (400 mg) were added sequentially to a solution of **109** (2.26 g, 7.79 mmol) in anhydrous CH₂Cl₂ (25 mL). The mixture was stirred for 3 h, and then filtered through CELITE[®]. The filtrate was washed with H₂O (25 mL), and dried over anhydrous MgSO₄. The solvents were evaporated, and the remaining material was recrystallized from EtOAc. The impure filtrate was purified by column chromatography on silica gel (25% to 40% EtOAc/hexanes), and combined with the crystallized material to afford 2.08 g (62 %) of **110**.

10 step 3

A solution of sodium hexamethyldisilazane (15.5 mL of a 1 M solution in THF, 15.5 mmol) was added slowly to a solution of **110** (2.08 g, 4.83 mmol) and **108** (1.14 g, 5.07 mmol) in anhydrous THF (25 mL) under nitrogen at 0° C. The reaction mixture was stirred for 3 h, and then added to a saturated aqueous solution of NH₄Cl (200 mL). The aqueous mixture was extracted with EtOAc (3 x 70 mL). The combined organic fractions were then washed with brine (50 mL), and dried over anhydrous MgSO₄. Evaporation of the solvents afforded a red oil which was dissolved in acetic acid (100 mL) and heated to reflux for 5 h. The solvent was removed, and the remaining material was dissolved in EtOAc (100 mL). The organic layer was washed with H₂O (40 mL), brine (25 mL), and dried over anhydrous MgSO₄. The solvents were evaporated and the crude product purified by flash chromatography on silica gel (20% to 100% EtOAc/hexanes) to afford **111a** (1.79 g, 69%) as a solid that was only slightly soluble in EtOAc.

step 4

A mixture of pyridazinone **111a** (1.79 g, 3.36 mmol), Fe powder (845 mg, 15.12 mmol), and NH₄Cl (809 mg, 15.12 mmol) in EtOH (60 mL) and H₂O (15 mL) was heated to reflux for 3 h. The reaction mixture was cooled to RT and filtered through CELITE[®]. The filter cake was washed with EtOAc (150 mL), and the combined organic fractions were washed with brine (75 mL), and dried over anhydrous MgSO₄. The solvents were evaporated to provide an oil. The oil was dissolved in CH₂Cl₂ (100 mL), and the organic layer was washed with brine (50 mL), and dried over anhydrous MgSO₄. Evaporation of the solvent provided **111b** (1.50 g; 88% theory).

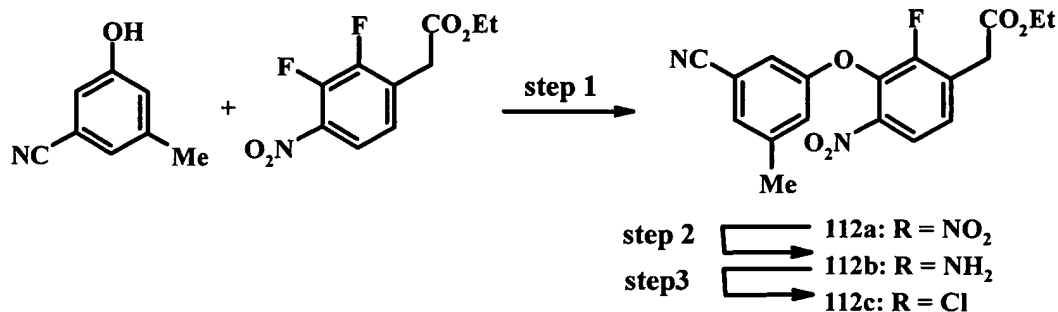
step 5

The aniline **111b** (700 mg, 1.39 mmol) and CuCl₂ (381 mg, 2.77 mmol) were suspended in anhydrous CH₃CN (14 mL) under a nitrogen atmosphere. *tert*-Butylnitrite (0.33 mL, 2.77 mmol) was added

dropwise, and the reaction mixture was warmed to 60°C for 1 h. The solution was cooled to RT, and a 5% aqueous HCl solution (20 mL) was added. The layers were separated, and the aqueous layer was extracted with EtOAc (3 x 30 mL). The combined organic fractions were washed with brine (30 mL) and dried over anhydrous MgSO₄. The solvents were evaporated, and the remaining solid was purified by flash chromatography over silica gel (20% to 100% EtOAc/Hexanes) to provide 500 mg of a solid. The solid was dissolved in anhydrous THF (10 mL) under nitrogen, and TBAF was added dropwise (5.73 mL of a 1.0 M solution, 5.73 mmol). The solution was heated to reflux for 1 h and then cooled to RT. The mixture was quenched with saturated aqueous NaHCO₃, and the aqueous solution was extracted with CH₂Cl₂ (3 x 30 mL). The combined organic fractions were washed with H₂O (30 mL), brine (30 mL), and dried over anhydrous MgSO₄. The solvents were evaporated, and the remaining solid was purified by repeated flash chromatography on silica gel (1% to 3% MeOH/CH₂Cl₂) to afford **I-235** (135 mg; 25% theory).

EXAMPLE 28

[4-Chloro-3-(6-cyano-4-methyl-pyridin-2-yloxy)-phenyl]-acetic acid ethyl ester



Synthesis of ethyl 2,3-difluoro-4-nitrophenylacetate

To an ice-cold solution of *tert*-butyl ethylmalonate (Alfa Aesar) (31.2g, 166mmole) in NMP (300 mL) cooled to 0°C under a nitrogen atmosphere was added NaH (60% oil dispersion, 13.1g, 218 mmole) while maintaining the temperature below 20°C. After addition complete, the solution was allowed to age for 20 minutes. To this solution was added dropwise 2,3,4-trifluoronitrobenzene (Oakwood Products Inc.) (26.6g, 163 mmole) in NMP (50 ml), while maintaining the temperature below 20°C (highly exothermic). Upon completion of addition the reaction aged at room temperature for 2 hours. The solution was added to an aqueous solution of NH₄Cl (1.5 L), extracted with ethyl acetate (3 x 200 mL), and washed 5 times with water (400 mL), dried (MgSO₄) and evaporated. The crude product was used without further purification.

The substituted malonic ester was dissolved in dichloromethane (400 mL) and TFA was added (100mL), this solution heated at 40°C for 16 hours. The reaction mixture was cooled to RT, and the

solvents evaporated. The crude product dissolved in EtOAc (400 mL), washed sequentially with aqueous NaHCO₃, water, and brine, dried (MgSO₄) and evaporated. The remaining oil was purified by flash chromatography on silica gel (5% EtOAc/hexane) to give the product as a golden oil (11.9g) (30%) which crystallizes upon sitting.

5

step 1

Solid sodium *tert*-butoxide (1.32g, 13.71 mmol) was added in 3 portions to a solution of 3-cyano-5-methylphenol (2.01 g, 15.08 mmol, prepared as described in WO 2002085860) in anhydrous THF (50 mL) under a nitrogen atmosphere. The resulting heterogeneous solution was stirred at RT for 15 m and then cooled to 0° C. A solution of the ethyl 2,3-difluoro-4-nitrophenylacetate (3.36 g, 13.71 mmol) in THF was added dropwise over 1 h. The purple mixture was warmed to RT, stirred for 16 h, and then added to a saturated aqueous solution of NH₄Cl (150 mL). The mixture was extracted with ethyl ether (3 x 100 mL), and the combined organics were washed with brine (1 x 100 mL), and dried over anhydrous MgSO₄. The solvents were evaporated, and the remaining oil was purified by flash chromatography on silica gel (0% to 50% EtOAc/hexanes) to provide 4.28 g (90%) of **112a**.

step 2

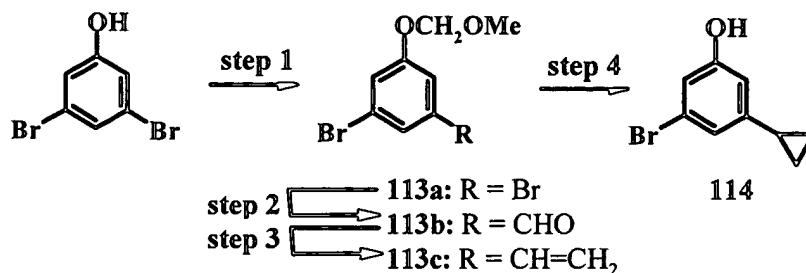
A mixture of **112a** (4.28 g, 12.31 mmol), Fe powder (2.89 g, 51.69 mmol), and NH₄Cl (2.76 g, 51.69 mmol) in EtOH (40 mL) and H₂O (40 mL) was heated to reflux for 3 h. The solution was cooled to RT, and filtered through CELITE®. The filter cake was washed with EtOAc (200 mL), the aqueous and organic layers were separated, and the organic layer was washed with brine (2 x 60 mL). The solution was dried over anhydrous MgSO₄ and evaporated to provide 3.81 g (100%) of the aniline **112b**.

step 3

A solution of **112b** (3.81g, 12.28 mmol) in anhydrous CH₃CN (40 mL) under a nitrogen atmosphere was added slowly to a mixture of *tert*-butyl nitrite (2.63 mL, 22.10 mmol) and CuCl₂ (2.48 g, 18.42 mmol) that had been prepared under a nitrogen atmosphere and was warmed to 60° C. The reaction temperature was maintained at 60° C for 1 h then aged at RT for an additional 2 h. The solution was cooled to 0° C, and an aqueous 5% HCl solution (80 mL) was added. The mixture was extracted with 1:1 EtOAc/hexanes (3 x 75 mL) and the combined organics were washed with brine (75 mL), dried over anhydrous MgSO₄, and evaporated. The remaining oil was purified by flash chromatography on silica gel (0% to 30% EtOAc/hexanes) to provide 2.20 g (52%) of **112c**.

The introduction of the pyridazinone ring and saponification and decarboxylation of the ester were accomplished utilizing the procedure described in steps 2-4 of Example 14.

EXAMPLE 29



step 1

Chloromethylmethyl ether (1.45 mL, 19.13 mmol) was added dropwise to a solution of 3,5-dibromophenol (4.38 g, 17.39 mmol) and diisopropylethylamine (3.63 mL, 20.90 mmol) in CH₂Cl₂ (40 mL) at 0° C. The mixture was warmed to RT, stirred for 16 h, and added to H₂O (50 mL). The layers were separated, and the aqueous fraction was extracted with CH₂Cl₂ (2 x 50 mL). The combined organic fractions were washed with brine (25 mL), dried over anhydrous MgSO₄, and the solvent was evaporated to provide 5.20 g (100%) of 1,3-dibromo-5-methoxymethoxy-benzene (113a).

step 2

n-BuLi (8.34 mL of a 1.6 M solution in Et₂O, 13.34 mmol) was added dropwise to a solution cooled to -78° C of 1,3-dibromo-5-methoxymethoxy-benzene (3.59 g, 12.13 mmol) in anhydrous Et₂O (40 mL) under a nitrogen atmosphere. The solution was aged at -78° C for 45 m, anhydrous DMF (1.03 mL, 13.34 mmol) was added, and the reaction mixture was slowly warmed to RT. The mixture was poured into H₂O (50 mL), and the aqueous phase was extracted with Et₂O (2 x 75 mL). The combined organic fractions were washed with brine (50 mL) and dried over anhydrous MgSO₄. The solvents were evaporated, and the remaining oil was purified by flash chromatography over silica gel (0% to 10% EtOAc/hexanes) to afford 2.18 g (74%) of 3-bromo-5-methoxymethoxy-benzaldehyde (113b).

step 3

A solution of *n*-BuLi (5.65 mL of a 1.6 M solution in Et₂O, 9.04 mmol) was added dropwise to a solution of methyltriphenylphosphonium bromide (3.23 g, 9.05 mmol) in THF (15 mL) that was cooled to 0° C. The resulting yellow solution was stirred for 30 m, cooled to -78° C, and a solution of 3-bromo-5-methoxymethoxy-benzaldehyde (1.58 g, 6.46 mmol) in anhydrous THF (15 mL) was added dropwise. The mixture was slowly warmed to RT, stirred for 16 h, and then added to a saturated aqueous solution of NaHCO₃ (60 mL). The layers were separated, and the aqueous layer was extracted with Et₂O

(2 x 50 mL). The combined organic layers were washed with brine (30 mL), and dried with anhydrous MgSO_4 . The volatile materials were evaporated, and the remaining oil was purified by flash chromatography on silica gel (0% to 3% EtOAc/hexanes) to provide 600 mg (76%) of 1-bromo-3-methoxymethoxy-5-vinyl-benzene (**113c**).

5

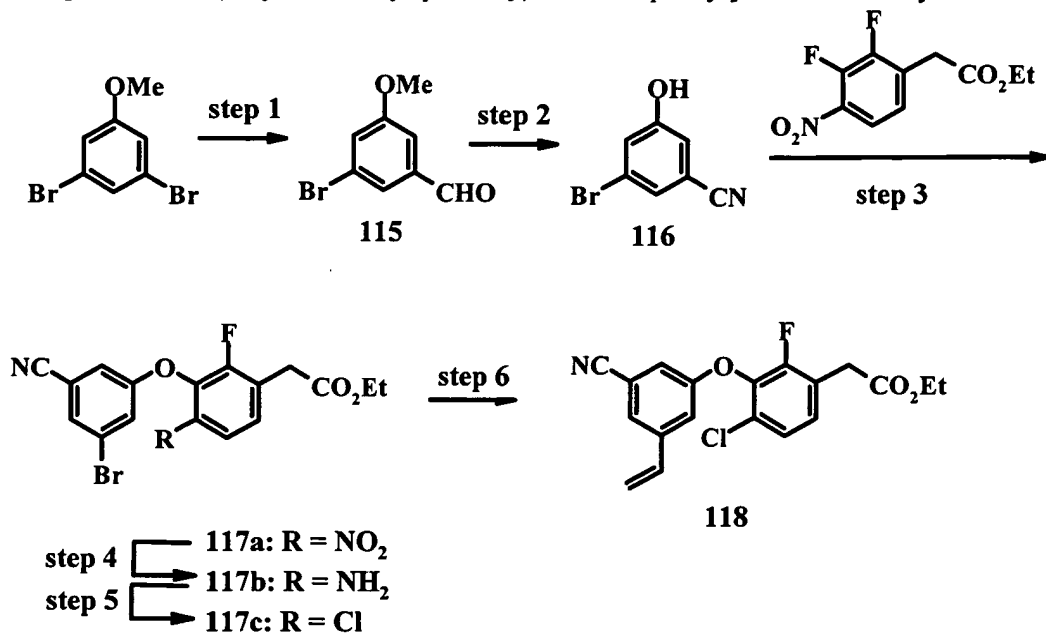
step 4

To anhydrous CH_2Cl_2 (10 mL) was added ZnEt_2 (12.34 of a 1.0 M in heptane, 12.34 mmol) under nitrogen. The solution was cooled to 0°C , and a solution of trifluoroacetic acid (0.95 mL, 12.34 mmol) in anhydrous CH_2Cl_2 (4 mL) was added very slowly. After stirring the reaction mixture for 20 m, a
10 solution of CH_2I_2 (0.99 mL, 12.34 mmol) in CH_2Cl_2 (4 mL) was added. After an additional 20 m stirring, a solution of the 1-bromo-3-methoxymethoxy-5-vinyl-benzene (1.20 g, 4.94 mmol) in CH_2Cl_2 (6 mL) was added, and the reaction was allowed to warm to RT. After 1.5 h, the reaction was quenched with saturated aqueous NH_4Cl (30 mL) and hexanes (50 mL), and the layers were separated. The aqueous layer was extracted with Et_2O (2 x 40 mL), and the combined organic layers were washed with
15 H_2O (30 mL), brine (30 mL), and dried with anhydrous MgSO_4 . The solvents were evaporated, and the remaining oil was partially purified by flash chromatography on silica gel (0% to 5% EtOAc/hexanes). The partially purified material was placed in *i*-PrOH (10 mL), and 2 mL of 1.0 M HCl was added. The mixture was aged at 50°C for 14 h, cooled to RT, and added to H_2O (25 mL). The aqueous mixture was extracted with Et_2O (3 x 40 mL), and the combined organics were washed with
20 brine (30 mL), and dried over anhydrous MgSO_4 . The solvents were evaporated, and the remaining oil was purified by flash chromatography on silica gel (0% to 10% EtOAc/hexanes) to afford 325 mg (31 %) of 3-bromo-5-cyclopropyl-phenol (**114**).

Phenols **113c** and **114** were condensed with ethyl 2,3-difluoro-4-nitrophenyl acetate as described in
25 Example 28. Substitution of the bromo radical with a cyano radical was accomplished by palladium-mediated displacement by $\text{Zn}(\text{CN})_2$ as described in step 5 of Example 20.

EXAMPLE 30

[4-Chloro-3-(3-cyano-5-vinyl-phenoxy)-2-fluoro-phenyl]-acetic acid ethyl ester



step 1

- 5 A solution of 1,3-dibromo-5-methoxybenzene (15 g, 56.40 mmol) in anhydrous Et₂O (200 mL) under nitrogen was cooled to -78° C. *n*-BuLi (38.80 mL of a 1.6 M solution in Et₂O, 62.00 mmol) was added dropwise, and the solution was aged at -78° C for 45 m. To the resulting heterogeneous mixture was added anhydrous DMF (4.78 mL, 62.00 mmol), and the solution was slowly warmed to RT. The mixture was poured into H₂O (200 mL), and the aqueous phase was extracted with Et₂O (3 x 125 mL).
- 10 The combined organic fractions were washed with brine (100 mL) and dried over anhydrous MgSO₄. Evaporation of the solvent provided 11.70 g (96 %) of 3-bromo-5-methoxybenzaldehyde (**115**).

step 2

- 15 A solution of 3-bromo-5-methoxybenzaldehyde (4.02 g, 18.7 mmol) and hydroxylamine hydrochloride (6.50 g, 93.5 mmol) in pyridine (50 mL) and EtOH (50 mL) was heated to 65° C for 16 h. The solvent was removed, and the remaining materials were partitioned between 1:1 EtOAc/hexanes (150 mL) and H₂O (75 mL). The organic layer was washed with brine (60 mL), and the solvents were evaporated. The remaining oil was dissolved in anhydrous dioxane (50 mL), and trifluoroacetic anhydride (5.1 mL, 37.4 mmol) and pyridine (9.07 mL, 112.2 mmol) were added. The mixture was heated to 60° C for 3 h
- 20 and then cooled to RT. CHCl₃ (100 mL) was added, and the organic layer was washed with H₂O (2 x 50 mL), 5% aqueous HCl solution (30 mL), brine (30 mL), and dried with anhydrous MgSO₄. The solvents were removed to provide a white solid. This solid was placed in a 150 mL flask that was flushed with nitrogen. Collidine (40 mL) and LiI (7.92 g, 59.10 mmol) were added, and the mixture was heated to 180° C for 5 h. The reaction mixture was cooled to RT, and partitioned between H₂O

(400 mL) and EtOAc (100 mL). The layers were separated, and the aqueous layer was acidified with 10% aqueous HCl solution, and extracted with 2:1 EtOAc/hexanes (3 x 125 mL). The combined organic layers were washed with H₂O (100 mL), 10% aqueous HCl solution (2 x 50 mL), brine (75 mL), and dried with anhydrous MgSO₄. The solvents were evaporated and the resulting solid was purified
5 by flash chromatography on silica gel (10% to 40% EtOAc/hexanes) to provide 3.40 g (92%) of 3-bromo-5-hydroxybenzonitrile (**116**).

step 3

Solid sodium *tert*-butoxide (1.67 g, 16.50 mmol) was added in three portions to a solution of 3-bromo-
10 5-hydroxybenzonitrile (3.27 g, 16.50 mmol) in anhydrous THF (40 mL) under nitrogen. The resulting heterogeneous solution was aged at RT for 15 m. The solution was cooled to 0 °C, and a solution of the ester (4.04 g, 13.71 mmol) was added dropwise over 30 m. The purple mixture was warmed to RT, stirred for 16 h, and then added to a saturated aqueous solution of NH₄Cl (100 mL). The mixture was extracted with ethyl ether (3 x 100 mL), and the combined organics were washed with brine (1 x 100
15 mL) and dried over anhydrous MgSO₄. The solvents were evaporated, and the remaining solid was purified by crystallization from 1:1 EtOAc/hexanes to provide 4.34 g (62%) of [3-(3-bromo-5-cyano-phenoxy)-2-fluoro-4-nitro-phenyl]-acetic acid ethyl ester (**117a**).

step 4

A solution of ester (**117a**, 4.34 g, 10.26 mmol), Fe powder (2.40 g, 43.07 mmol), and NH₄Cl (2.30 g, 43.07 mmol) in EtOH (80 mL) and H₂O (40 mL) was heated to reflux for 2 h. The solution was cooled to RT and filtered through CELITE®. The filter cake was washed with EtOAc (200 mL), the aqueous and organic layers were separated, and the organic layer was washed with brine (2 x 60 mL). The solution was dried with anhydrous MgSO₄, and evaporated to provide 3.98 g (98 %) of the [4-amino-3-
25 (3-bromo-5-cyano-phenoxy)-2-fluoro-phenyl]-acetic acid ethyl ester (**117b**).

step 5

The aniline **117b** (3.98g, 10.10 mmol) was dissolved in anhydrous CH₃CN (35 mL) under nitrogen. This solution was then added dropwise to a mixture of *tert*-butylnitrite (2.40 mL, 20.20 mmol) and
30 CuCl₂ (2.72 g, 20.20 mmol) that had been prepared under nitrogen and warmed to 60 °C. The reaction temperature was maintained at 60 °C for 2 h, then cooled to 0 °C. A 5% aqueous HCl solution (80 mL) was added, and the mixture was extracted with 1:1 EtOAc/hexanes (3 x 75 mL). The combined organics were washed with brine (75 mL), dried over anhydrous MgSO₄, and evaporated. The remaining oil was

purified by flash chromatography on silica gel (0% to 30% EtOAc/Hexanes) to provide 2.67 g (64 %) of the [4-chloro-3-(3-bromo-5-cyano-phenoxy)-2-fluoro-phenyl]-acetic acid ethyl ester (**117c**).

step 6

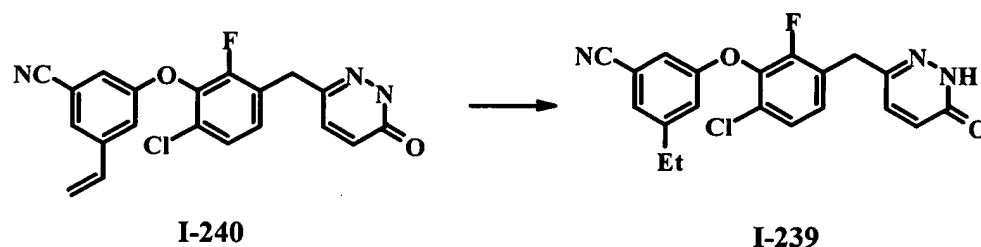
- 5 A 2-necked flask was equipped with a reflux condenser and charged with **117c** (1.14 g, 2.78 mmol) and tetrakis(triphenylphosphine)palladium (308 mg, 0.28 mmol). Anhydrous toluene (15 mL) and tributylvinyltin (0.85 mL, 2.91 mmol) were added, and the reaction was heated to reflux under nitrogen for 16 h. The solution was cooled to RT and EtOAc (50 mL) was added. The mixture was washed with H₂O (25 mL) and brine (25 mL), and dried over anhydrous MgSO₄. The solvents were evaporated, and
- 10 the resulting oil purified by flash chromatography over silica gel(0% to 25% EtOAc/hexanes) to provide 770 mg (78%) of [4-chloro-3-(3-cyano-5-vinyl-phenoxy)-2-fluoro-phenyl]-acetic acid ethyl ester (**118**).

The introduction of the pyridazinone ring into **118** and saponification and decarboxylation of the ester were accomplished utilizing the procedure described in steps 2-4 of Example 14. Introduction of other

15 alkyl, alkenyl or alkynyl is accomplished as described in step 6 with the appropriate tributyltin derivative.

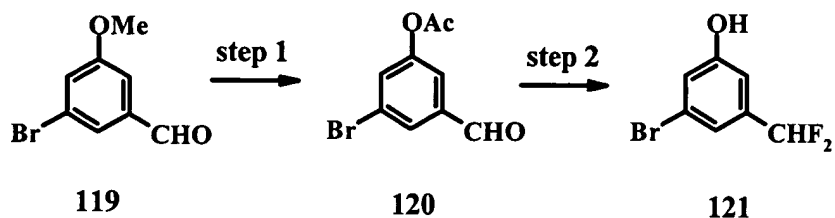
EXAMPLE 31

- 20 3-[6-Chloro-2-fluoro-3-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-5-ethyl-benzonitrile



- A round-bottom flask was charged with **I-240** (100 mg, 0.26 mmol) and 5% palladium on carbon (20 mg) and EtOAc (5 mL) was added. The flask was evacuated, backfilled with H₂ (balloon pressure), and
- 25 the solution was stirred for 3 h. The solution was filtered through CELITE[®], and the filter cake was washed with EtOAc (20 mL). Removal of the solvent provided 90 mg (90 %) of 3-[6-chloro-2-fluoro-3-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-5-ethyl-benzonitrile (**I-239**).

EXAMPLE 32



5 step 1

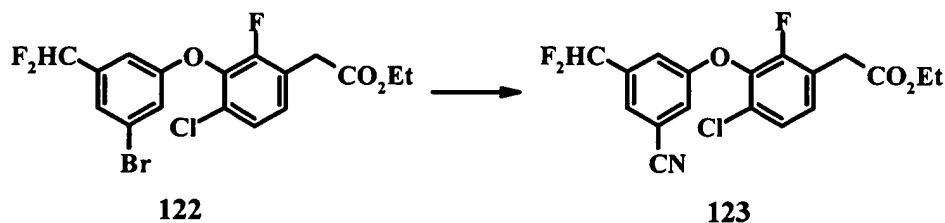
A solution of BBr_3 (29.1 mL of a 1.0 M solution in CH_2Cl_2 , 29.1 mmol) was added slowly to a solution of **119** (2.5 g, 11.62 mmol) in anhydrous CH_2Cl_2 (25 mL) under nitrogen at -78°C . The orange solution was warmed to RT, stirred for 2 h, and poured onto ice. The mixture was extracted with CH_2Cl_2 (100 mL), and the organic layer was washed with H_2O (50 mL) and brine (50 mL). The solvents were
 10 evaporated, and the remaining oil was purified by flash chromatography on silica gel (0% to 20% EtOAc/hexanes) to provide the desired phenol. To a solution of this phenol in pyridine (10 mL) under argon was slowly added acetic anhydride (0.6 mL, 6.33 mmol). After 2 h, the volatile materials were removed to provide 3-bromo-5-formyl-phenyl acetate (**120**, 1.02 g, 40 %).

15 step 2

Diethylaminosulfur trifluoride (1.02 mL, 7.69 mmol) was added to a solution of the 3-bromo-5-formyl-phenyl acetate (**120**, 1.1 g, 4.52 mmol) in CH_2Cl_2 (5 mL) under nitrogen contained in a NALGENE[®] bottle. EtOH (0.013 mL, 0.23 mmol) was added, and the mixture was stirred for 16 h. The reaction mixture was then added slowly to an aqueous solution of saturated NaHCO_3 . After the bubbling was
 20 finished, CH_2Cl_2 (50 mL) was added and the layers were separated. The organic layer was washed with brine (30 mL) and dried with anhydrous MgSO_4 . The solvent was removed to provide a yellow oil that was placed in a mixture of THF (15 mL) and H_2O (4 mL). LiOH monohydrate (474 mg, 11.3 mmol) was added, and the reaction mixture was stirred at RT for 2 h. The solution was then added dropwise to 5% aqueous HCl (50 mL), and the mixture was extracted with EtOAc (3 x 30 mL). The combined
 25 organic fractions were washed with brine (30 mL), and dried with anhydrous MgSO_4 . Evaporation of the volatile materials gave an oil that was purified by flash chromatography on silica gel (0% to 25% EtOAc/hexanes) to provide 800 mg (79%) of 3-bromo-5-difluoromethylphenol (**121**).

The phenol **121** was condensed with ethyl 2,3-difluoro-4-nitro-phenyl acetate as described in step 1 of
 30 Example 28. Reduction of the nitro group and diazotization and displacement of the diazonium salt by chloride were carried out as described in steps 2 and 3 of Example 28 to afford **122**.

step 3

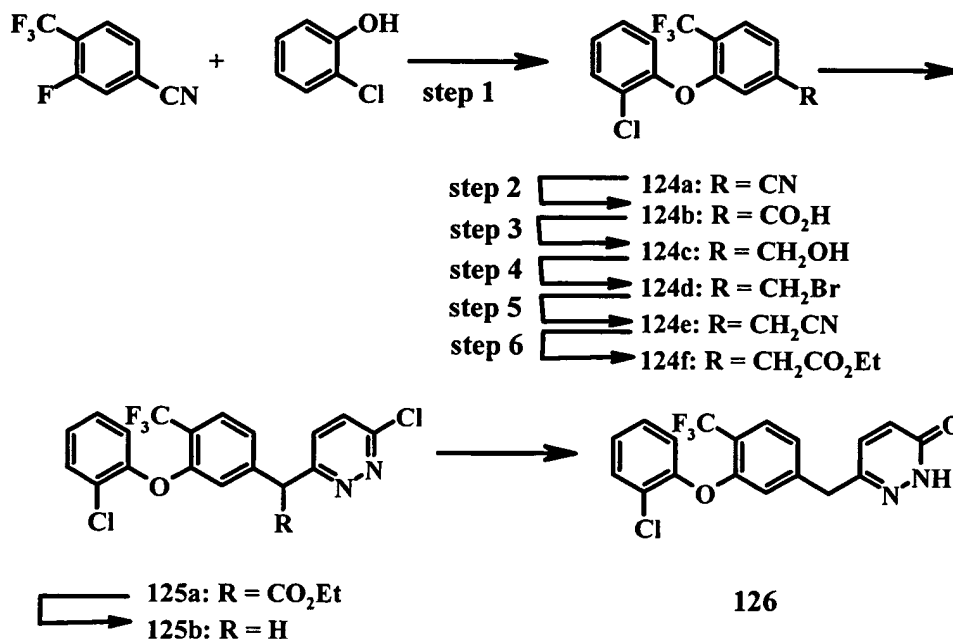


5 A solution of **122** (757 mg, 1.73 mmol), Pd[P(Ph)₃]₄(0) (300 mg, 0.26 mmol), and zinc cyanide (122 mg, 1.04 mmol) in DMF (8 mL) under nitrogen was heated to 80° C for 4 h. The reaction mixture was cooled to RT and added to 2 M aqueous NH₄OH. The solution was extracted with 1:1 EtOAc/hexanes (3 x 30 mL), and the combined organic fractions were washed with H₂O (3 x 20 mL) and dried over anhydrous MgSO₄. The solvent was evaporated, and the remaining oil was purified by flash chromatography on silica gel (0% to 25% EtOAc/hexanes) to provide 580 mg (87%) of [4-chloro-3-(3-cyano-5-difluoromethyl-phenoxy)-2-fluoro-phenyl]-acetic acid ethyl ester (**123**).

10

EXAMPLE 33

6-[3-(2-Chloro-phenoxy)-4-trifluoromethyl-benzyl]-2H-pyridazin-3-one



step 1

15 A round-bottom flask was charged with sodium hydride (254 mg, 6.35 mmol; 60% in mineral oil) and a solution of 2-chlorophenol (658 μ L, 6.35 mmol) in 20 mL of NMP. After stirring for 30 m, a solution of 3-fluoro-4-trifluoromethylbenzonitrile in 5 mL of NMP was added and the reaction was heated at 120° C for 17 h. The reaction mixture was cooled and partitioned between dichloromethane and water.

The aqueous phase was twice extracted with DCM and the combined organic phases were washed with water and brine, dried (MgSO₄), filtered and evaporated. The crude product was purified by flash chromatography on silica gel (EtOAc:hexane 1:30) to afford 1.64 g of 3-(2-chlorophenoxy)-4-trifluoromethylbenzonitrile (**124a**).

5

step 2

To a solution of 3-(2-chlorophenoxy)-4-trifluoromethylbenzonitrile (**124a**, 1.64 g, 5.51 mmol) in MeOH (25 mL) was added a solution of NaOH (450 mg) in water (5 mL). The reaction mixture was heated at reflux for 16 h. The mixture was cooled to RT and partitioned between water and EtOAc. The EtOAc
10 was removed and the aqueous phase was acidified with 1N HCl and thrice extracted with EtOAc (25 mL). The combined organic extracts were sequentially washed with water and brine, dried over MgSO₄, filter and evaporated to yield a white solid which was washed with hexane and dried to afford 1.44 g of carboxylic acid **124b**.

15 step 3

To an ice-cold solution of **124b** (1.44 g, 4.55 mmol) and anhydrous THF (25 mL) was added dropwise BH₃-THF (31.8 mL of a 1.0 M solution in THF) and the solution then heated at reflux for 1.5 h. The reaction was cooled to RT and methanol was added very slowly. The reaction mixture was diluted with EtOAc and washed sequentially with 1N HCl, saturated NaHCO₃ and water. The organic phase was
20 dried MgSO₄, filtered and evaporated. The crude product was purified by flash chromatography on silica gel (EtOAc:hexane 1:4) to afford 1.23 g of **124c**.

step 4

To a solution of the alcohol **124c** (1.37 g, 4.53 mmol) was dissolved 25 mL of THF was added
25 sequentially CBr₄ (3.09 g, 9.05 mmol) and triphenylphosphine (2.37 g, 9.05 mmol). After 30 min the reaction was diluted with 30 mL EtOAc, washed with brine, dried (MgSO₄), filtered and evaporated. The crude product was purified by flash chromatography over silica gel (hexane) to afford 1.69 g **124d**.

step 5

30 To a solution of **124d** (1.69 g, 4.62 mmol) in 25 mL of EtOH was added a solution of KCN (793 mg, 16.2 mmol) in 3 mL of water. The reaction mixture was stirred for 6 h at RT and the volatile solvents removed *in vacuo*. The crude product was partitioned between EtOAc and saturated aqueous NaHCO₃. The organic phase was washed with brine, dried (MgSO₄), filtered and evaporated. The crude product was purified by flash chromatography on silica gel to afford the nitrile **124e** (790 mg)

step 6

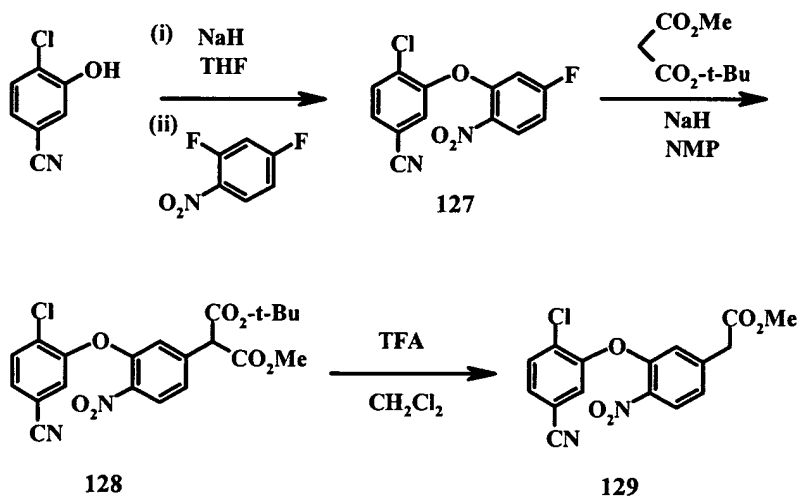
A solution of **124 e** (0.790 g, 2.53 mmol), 48% HBr (4 mL) and glacial HOAc (4 mL) was heated at 110° C for 4 h. The reaction was cooled and diluted with EtOAc and washed twice with brine. The organic phase was evaporated *in vacuo* and the resulting oil was dissolved in 10 mL of EtOH and 1 mL of con H₂SO₄ was added and the resulting solution heated at 75° C for 15 h. The reaction mixture was cooled and partitioned between water and Et₂O. The Et₂O was washed sequentially with saturated NaHCO₃, water and brine, dried MgSO₄, filtered and evaporated to afford **124f** (578 mg, 1.61 mmol).

step 7

To a solution of **124f** (250 mg, 0.7 mmol) and 3,6-dichloropyridazine (261 mg, 1.4 mmol) in 10 mL of DMF was added NaH (56 mg, 1.4 mmol, 60% in mineral oil). The reaction mixture was stirred for 1 h. The workup that afforded **125a** and subsequent conversion to the pyridazinone is carried out as described in steps 1 and 2 of Example 18.

EXAMPLE 34

3-(2-Chloro-5-cyano-phenoxy)-4-nitro-benzoic acid methyl ester



step 1

To an ice-cold solution under argon atmosphere of 2-chloro-5-cyanophenol (7.36 g, 48.10 mmol) in anhydrous THF (100 mL) was added potassium *tert*-butoxide (52.92 mL of 1 M solution in THF, 52.92 mmol) with stirring over 30 m. The cooling bath was then removed and the resulting mixture was stirred for 40 m at RT. The reaction mixture was cooled down to 0° C. A solution of 2, 4-difluoronitrobenzene (8.41 g, 52.92 mmol) in THF (10 mL) was then added to the potassium phenoxide at 0° C over a 20 m

period. The resulting yellow slurry was then heated at 50° C for 16 h. The final reaction mixture was poured into ice-water (500 mL) and the mixture was extracted with EtOAc (4 × 100 mL). Organic layer was washed with water (100 mL), dried (MgSO₄), and purified by flash chromatography on silica gel (hexane:EtOAc 9:1) to afford **127** (13.3 g; 94.6% theory)as pale yellow oil.

5

step 2

To a well stirred solution at 0° C under argon atmosphere of *tert*-butyl methyl malonate (8.18 g, 46.98 mmol) in anhydrous NMP (120 mL) was added carefully NaH (3.76 g, 94 mmol; 60% on mineral oil) over 40 m. The resulting mixture was then stirred at 0° C for 30 m and a solution of **127** (12.5 g, 42.71 mmol) in anhydrous NMP (50 mL) was added dropwise over 1.5 h with stirring while maintaining the temperature at 0° C under argon atmosphere. The cooling bath was removed and the reaction mixture was stirred at RT for 2.5 h. The reaction mixture was poured into 10% NaHSO₄ (400 mL) and the mixture was extracted with EtOAc (4 × 200 mL). The combined organic layers were washed with water (3 × 100 mL) and brine (1 × 100 mL) and dried (MgSO₄). After filtration of the MgSO₄ the solvent was evaporated *in vacuo* to afford **128** as yellow residue.

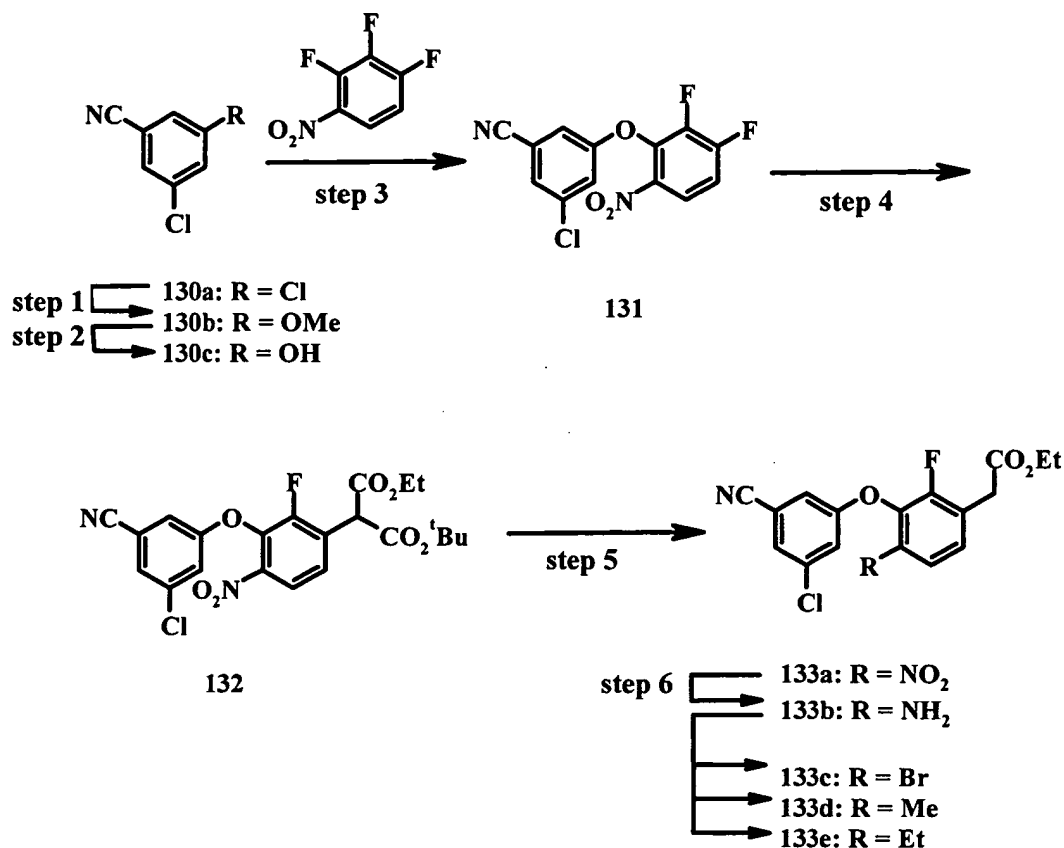
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step 3

The crude product was dissolved in DCM (30 mL) and TFA (80 mL) was added with stirring. The resulting mixture was refluxed for 2 hours. TFA and DCM were removed *in vacuo* and the residue was mixed with water (100 mL). The mixture was adjusted to a pH between 7 and 8 with 10% NaHCO₃ and the resulting mixture was extracted with EtOAc (4 × 100 mL). The combined organic layers were washed with water (2 × 100 mL), dried (MgSO₄) and purified by flash chromatography on silica gel (hexane:EtOAc 3:1) to yield 12.42 g (84.3%) of 3-(2-chloro-5-cyano-phenoxy)-4-nitro-benzoic acid methyl ester (**129**) as yellow oil.

25

EXAMPLE 35



step 1

- 5 To a 250 mL round bottom flask charged with 3,5-dichlorobenzonitrile (**130a**, 7.31g; 34.90 mmol) and maintained under an argon atmosphere was added DMF (70 mL). The flask was cooled to 0° C and powdered sodium methoxide (1.88 g; 34.90 mmol) was added in two portions 15 m apart. The homogeneous mixture was allowed to warm to room temperature and stirred for 24 h. The solution was cooled to 0° C and aqueous 10% HCl (20mL) was added dropwise *via* an addition funnel after which the
- 10 reaction was warmed to RT. The mixture was extracted with EtOAc and the combined extracts washed sequentially with water and brine. The organic phase was dried (Na₂SO₄), filtered, and volatile solvents were removed *in vacuo*. The resulting solid was recrystallized from hexanes to afford 3-chloro-5-methoxybenzonitrile (**130b**, 4.2g; 72%).

15 step 2

A 250 mL round bottom flask was charged 3-chloro-5-methoxybenzonitrile (4.2g; 25.05 mmol) and 2,4,6-collidine (60 mL) was added. The mixture was stirred under an argon atmosphere until the solution was homogeneous. Anhydrous lithium iodide (10.06 g; 75.18 mmol) was added and the

mixture was heated to 175° C for 3 h. The reaction mixture was cooled to RT and partitioned between 10% HCl and EtOAc. The EtOAc phase was washed sequentially with 10 % HCl and brine, dried (Na₂SO₄), filtered and evaporated *in vacuo* to afford a oil which was crystallized from hexanes to afford 3-chloro-5-hydroxybenzonitrile (**130c**, 3.5g, 91% theory).

5

step 3

To an ice-cold solution of 3-chloro-5-hydroxybenzonitrile (**130c**, 3.5g; 22.80 mmol) and dry THF (50 mL) maintained under an argon atmosphere was sodium *tert*-butoxide (2.2g; 22.80 mmol) in two portions 15 m apart. The reaction mixture was stirred until the mixture was homogeneous. To the ice-cold solution was added dropwise 2,3,4-trifluoronitrobenzene (4.0 g; 22.80 mmol) over 30 m. The reaction was stirred at 0° C for 3 h and then allowed to warm to RT. The reaction was cooled to 0° C and quenched by addition of 10% HCl *via* addition funnel. The resulting mixture was extracted with EtOAc and the combined organic phases washed sequentially with 10% HCl and brine. The EtOAc was dried (Na₂SO₄), filtered and the volatile solvent removed *in vacuo* to yield a yellow oil which was crystallized from hexanes to yield **131** (6.3g, 89% theory)

15

step 4

To an ice-cold solution of *tert*-butyl ethyl malonate (3.8g; 20.28mmol) and dry NMP maintained under an argon atmosphere was added NaH (1.2 g, 48.67 mmol, 60% in mineral oil) over a 45 m interval. The reaction was stirred for an additional 30 m after which **131** (6.3 g, 20.28mmol) was added dropwise and the resulting solution stirred for 4 h. The reaction mixture was cooled to 0° C and quenched by dropwise addition of a saturated NaHSO₄ solution. The mixture was EtOAc and the combined organic extracts washed sequentially with water and brine. The EtOAc solution was dried (Na₂SO₄), filtered and the volatile solvents removed *in vacuo* to afford **132** as a purple oil that was used without further purification.

25

step 5

The crude mixed ester **132** from the previous step (8.9 g; 18.60 mmol) was dissolved in DCM (100 mL) and 50 mL of TFA was added and the solution was to heated to 60° C for 24 h. The reaction mixture was cooled to 0° C and saturated NaHCO₃ was added dropwise to the stirred reaction mixture. The resulting solution was extracted with EtOAc and washed sequentially with saturated NaHCO₃, water and brine. The organic phase was dried (Na₂SO₄), filter and the volatile solvents removed *in vacuo*. The resulting dark oil was recrystallized from hexanes to afford **133a** (6.5 g, 92% theory).

30

step 6

To a solution of **133a** (6.5 g; 17.20 mmol) and absolute EtOH (100mL) was added NH₄Cl (1.84 g, 34.39 mmol) dissolved in water (20 mL). The resulting mixture was heated at 60° C until the reaction was homogeneous. Fe(0) (1.44 g, 25.80 mmol) was then added and the mixture stirred vigorously at 60° C for 6 h. When reduction was complete the hot reaction mixture was filtered through a pad of CELITE® which subsequently was washed with hot EtOAc. The resulting filtrate was cooled and extracted with EtOAc and the combined extracts washed sequentially with water and brine. The EtOAc extract was dried (Na₂SO₄), filtered and the volatile solvent was removed *in vacuo* to afford a pale orange oil which was recrystallized from hexanes to yield **133b** (5.0 g, 83% theory).

Introduction of 5-bromo substituent

A 150 mL three-neck round bottom flask was charged with MeCN (50 mL), CuBr (2.8 g, 12.61 mmol) and *t*-butyl nitrite (1.4 g, 13.76 mmol), degassed and maintained under an Ar atmosphere and heated to 70° C. To the mixture was added dropwise a solution of **133b** (4.0 g, 11.47 mmol) dissolved MeCN (20 mL). The reaction mixture was stirred at 70° C for 4 h and then cooled to 0° C. The reaction was quenched by addition of 10 % HCl (30 mL) and extracted with EtOAc. The combined extracts were sequentially washed with 10% HCl and brine. The organic extract was dried (Na₂SO₄), filtered and the volatile solvents removed *in vacuo* to yield a black oil which was purified by flash chromatography on silica gel (hexanes:EtOAc 95:5) to afford **133c** (2.5 g, 52.8% theory).

Introduction of 5-methyl substituent

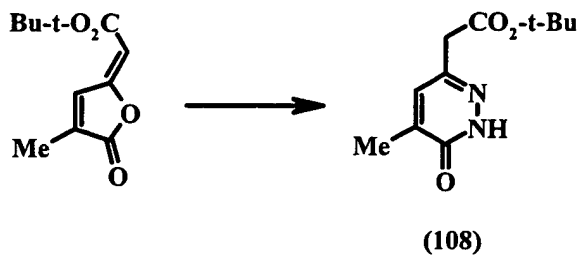
To a degassed ice-cold solution of THF (15mL), Pd(dppf)Cl₂ (0.09 g, 0.121 mmol) was added DIBAL-H (0.012 mmol; 1M in toluene). The reaction mixture was allowed to warm to RT. A solution of **133b** (1.0 g, 2.42 mmol) was added followed by dimethyl zinc (1M in THF, 4.240 mmol). The reaction was heated to 65° C for 4 h, cooled to RT and quenched with aqueous NH₄Cl. The resulting mixture was extracted with EtOAc and washed sequentially with NH₄Cl and brine. The EtOAc extract was dried (Na₂SO₄), filtered and the volatile solvent removed *in vacuo* to yield a dark brown oil that was purified by flash chromatography on silica gel (hexanes:EtOAc 95:5) to yield **133d** (0.50 g, 59% theory).

Introduction of 5-ethyl substituent

133e was prepared in by an identical procedure except diethylzinc was substituted for dimethyl zinc. The product was purified by flash chromatography on silica gel (hexanes:EtOAc 95:5) to yield **133e** (0.65 g, 74% theory).

EXAMPLE 36

(5-Methyl-6-oxo-1,6-dihydro-pyridazin-3-yl)-acetic acid tert-butyl ester (**108**)

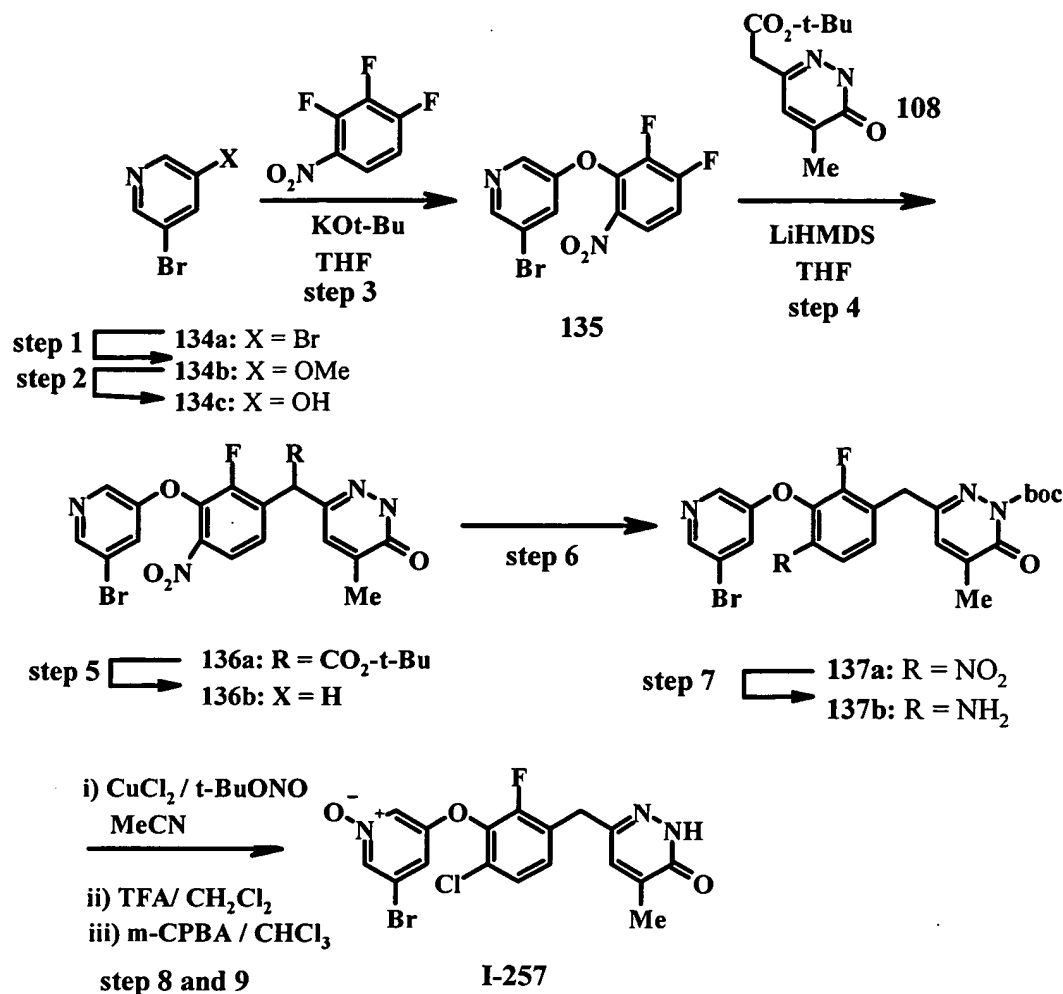


The alkylidene lactone was synthesized from citraconic anhydride via a Wittig reaction using the
5 procedure previously described in the literature (Massy-Westropp, R. A. and Price, M. F., *Aust. J. Chem.* **1980**, 33, 333-341). To a solution of the lactone (9.02 g, 42.9 mmol) in 100 mL of ethanol was added 4.5 mL (144 mmol) of hydrazine hydrate. The reaction mixture was refluxed for 6 h, cooled and concentrated. Successive crystallizations of the crude reaction mixture from hexanes yielded **108** as a clear crystalline solid (8.02 g, 83 %): mp = 113.0-113.9° C, ms: $[\text{M} + \text{H}]^+ = 225$.

10

EXAMPLE 37

6-[3-(5-bromo-1-oxy-pyridin-3-yloxy)-4-chloro-2-fluoro-benzyl]-4-methyl-2H-pyridazin-3-one



5 step 1

A solution of 3,5-dibromopyridine (**134a**, 20g, 84.4 mmol) in DMF (200mL) was stirred at RT under nitrogen atmosphere and then 21.3 mL of sodium methoxide (25% by wt. in methanol (92.8 mmol) was added slowly. The reaction mixture was stirred overnight at 70° C under N₂. The reaction was cooled to RT and quenched with water (200mL) and extracted with Et₂O (2x 200mL). The combined organic
10 extracts was washed with brine, dried (MgSO₄) and concentrated *in vacuo*. The crude 3-bromo-5-methoxypyridine (**134b** 14.8g, 93% theory) afforded a colorless oil after purification by flash chromatography on silica gel (EtOAc:hexane 1:10).

step 2

A solution of 3-bromo-5-methoxy-pyridine (**134b** 18.8 g, 0.1 mol), HBr (80 mL, 48%) and glacial HOAc (60 mL) was stirred overnight at 120° C. The reaction mixture was cooled to RT and then poured into the ice. The pH was adjusted to about 6 by adding 6N NaOH and then extracted with EtOAc (2 x 200 mL). The organic layer was dried over Na₂SO₄ and concentrated *in vacuo*. The crude product was stirred in CH₂Cl₂ (150 mL) and the resulting precipitate was filtered. The product was washed with CH₂Cl₂ to afford 3-bromo-5-hydroxypyridine (**134c** 15.2 g, 87.4% theory) as a white solid.

step 3

A solution of 3-bromo-5-hydroxypyridine (**134c** 7.4g, 42.5 mmol) in anhydrous THF (40 mL) was stirred at 0° C under Ar atmosphere and potassium *tert*-butoxide (46.8 mL, 1M solution in THF) was added slowly. After 1 h at 0° C, 2,3,4-trifluoronitrobenzene (7.91g, 44.6 mmol) in 15 mL of THF was added very slowly. The reaction mixture was stirred at RT for 2 h, quenched with water (80mL) and extracted with EtOAc (2 x 80 mL). The combined organic extracts were dried (MgSO₄) and concentrated *in vacuo*. The crude product was purified by flash chromatography on silica gel (EtOAc:hexane 1:15) to afford **135** (11 g, 78%) as a light orange oil.

step 4

A solution of (5-methyl-6-oxo-1,6-dihydro-pyridazin-3-yl)-acetic acid *tert*-butyl ester (**108**, 7.1g, 31.7 mmol) and **135** (11g, 33.3 mmol) in anhydrous THF (30 mL) was stirred at -78° C under an Ar atmosphere and 112 mL of LiHMDS (1.0M solution in THF) was added very slowly. The reaction mixture was stirred in the cold bath (dry-ice/IPA) for 3 h then in an ice bath for 2 h. The reaction was quenched with a solution of NaHSO₄.H₂O (5% by wt) and extracted with EtOAc (2x 100mL). The combined organic extracts were dried (MgSO₄) and concentrated *in vacuo*. The product was isolated by a flash chromatography on silica gel (EtOAc:hexane 1:2 to 2:1) to afford **136a** as a yellow solid (10.2 g, 60% yield).

step 5

A solution of **136a** (10.2g, 19.1 mmol) in HOAc (120 mL) under a nitrogen atmosphere was heated to reflux overnight. It was cooled to RT and the HOAc was evaporated *in vacuo*. A saturated NaHCO₃ solution (70mL) was added and the aqueous mixture extracted with EtOAc (2 x 80mL). The combined organic fractions were dried (MgSO₄) and concentrated *in vacuo*. The crude product was isolated by a flash chromatography. on silica gel (EtOAc:hexane 1:2 to 2:1) to afford **136b** as a light yellow solid (4.6g, 55.3% theory): ms (M+H)⁺ = 436.

step 6

A solution of the **136a** starting material (1.8g, 4.4 mmol), di-*tert*-butyl dicarbonate (1.16g, 5.3mmol), and 4-dimethylaminopyridine (0.2g) in anhydrous THF (30 mL) was maintained under an Ar atmosphere and stirred at RT overnight. The reaction mixture was quenched with water and extracted with EtOAc (2x30 mL). The combined organic fractions were dried (MgSO₄) and concentrated *in vacuo*. The product was isolated by a flash chromatography on silica gel (1:10 to 2:1 EtOAc:hexane) to afford **137a** as a white solid compound (0.85 g; 38% theory).

step 7

The reduction was carried out as described in step 6 of Example 35. From **137a** (4g, 9.19 mmol) there was obtained 1.8g (4.44 mmol) of **137b** as an off-white solid (48.3% yield).

step 8

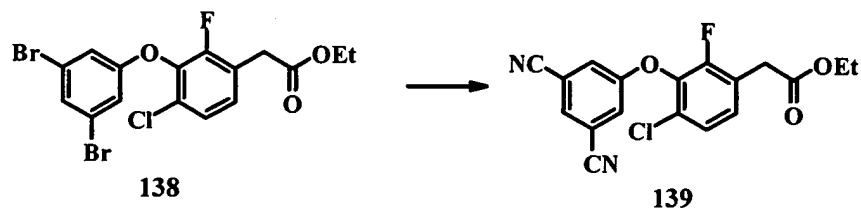
Reduction of the nitro group and diazotization and displacement of the diazonium salt by chloride were carried out as described in step 3 of example 26. The Boc group was removed with trifluoroacetic acid and DME. From 0.85g (1.69 mmol) of **137a** there was obtained 290 mg of the aryl chloride (49.9% theory for the two steps) as a white solid: mp 184.9-188° C, ms [M+H]⁺ = 424.

step 9

A solution of the pyridinyl compound (0.2g, 0.47 mmol) and MCPBA (0.09g, 0.52 mmol) in anhydrous chloroform (10 mL) was heated at reflux for 6 hours. The reaction mixture was cooled to RT, and diluted with 0.05N NaOH (5mL) and extracted with chloroform (2 x 10 mL). The combined organic fractions were dried (MgSO₄) and concentrated *in vacuo*. The crude product was purified by a flash chromatography on silica gel (MeOH:CH₂Cl₂ 0.1 to 1:10) to afford 6-[3-(5-bromo-1-oxy-pyridin-3-yloxy)-4-chloro-2-fluoro-benzyl]-4-methyl-2H-pyridazin-3-one (**I-257**, 60 mg; 32% theory) as a white solid: mp 197.9-198.9° C, ms (M+H)⁺ = 440.

EXAMPLE 38

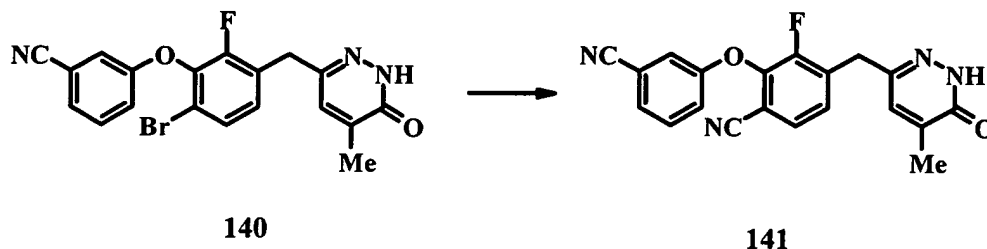
[4-Chloro-3-(3,5-dicyano-phenoxy)-2-fluoro-phenyl]-acetic acid ethyl ester



5 A mixture of ethyl 4-chloro-3-(3,5-dibromo-phenoxy)-2-fluoro-phenyl-acetate (**138**, 41.45g, 88.8 mmol), zinc cyanide (12.5g, 106 mmol), Pd(PPh₃)₄(0) (10.26g, 8.88 mmol) and anhydrous DMF (500 mL) was evacuated under house vacuum and purged with argon three times. The mixture was stirred at 80° C under an argon atmosphere. After 4 h the mixture was cooled to RT and filtered through a pad of silica gel. The filtrate was extracted with 1:1 EtOAc-hexanes (3 x 200 mL). The combined organic
10 phases were sequentially washed with water and brine and dried (MgSO₄). The product (37.4g) was purified by flash chromatography on 500g of silica gel (EtOAc-hexanes 1:10 to 2:10). The resulting material was recrystallized from isopropanol to afford **139** (28.3 g, 89% theory).

EXAMPLE 39

15 2-(3-Cyano-phenoxy)-3-fluoro-4-(5-methyl-6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-benzonitrile

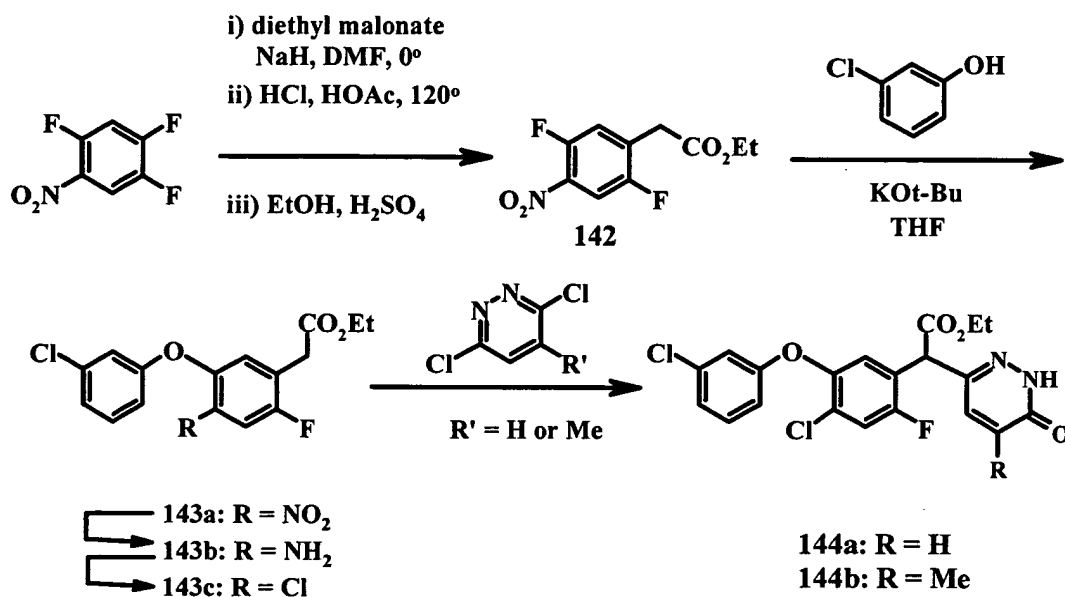


Pyridazinone **140** (100 mg, 0.24 mmol), ZnCl₂ (22 mg, 0.19 mmol), and Pd(PPh₃)₄ (62 mg, 0.05 mmol) were combined in a dry round bottom flask, purged of atmospheric oxygen with argon and subsequently
20 charged with dry DMF (2.7 mL). The mixture was heated to 80° C and allowed to stir for two h. The mixture was allowed to cool to room temperature, diluted with 1:1 EtOAc/hexanes (50 mL) and washed with water (4x50 mL). The organic phase was dried (MgSO₄) and concentrated *in vacuo*. The crude oil was purified by flash chromatography on silica gel (5% methanol/dichloromethane) followed by preparative HPLC to afford **141** (36 mg, 39%) as a white solid.

25

EXAMPLE 40

6-[4-Chloro-5-(3-chloro-phenoxy)-2-fluoro-benzyl]-2H-pyridazin-3-one (**144a**) and 6-[4-Chloro-5-(3-chloro-phenoxy)-2-fluoro-benzyl]-4-methyl-2H-pyridazin-3-one (**144b**)



5 step 1

An ice-cooled slurry of NaH (2.2g, 54.9 mmol, 60% in mineral oil) in dry DMF (60 mL) was prepared under an N₂ atmosphere in a septum-equipped flask. Diethyl malonate (4.2 mL, 27.5 mmol) was added dropwise by syringe over 10m, and stirring at 0° C was continued for another 30 m.

Trifluoronitrobenzene (Aldrich, 3.0 mL, 26.2 mmol) was then added dropwise over 20 m and the mixture was stored at -6° C for 16h. The mixture was diluted with water and extracted with a 3:2 mixture of EtOAc and hexane. The organic extracts were washed with H₂O, dried (Na₂SO₄) and the solvents evaporated to afford the crude product which was purified by flash chromatography on silica gel (acetone:hexane 1:10) to afford the diethyl malonate adduct (7.9g, 95% theory) as a yellow oil. A mixture of substituted malonate ester (7.8g, 24.6 mM) with glacial HOAc (80mL) and HCl (6N, 80mL) was heated under nitrogen at 120° for 2.5 h and then allowed to cool and stirred for 16 h. Most of the solvent was removed by evaporation and then water was added to the residue which produced a precipitate. The slurry was cooled in ice and the precipitate collected by filtration, further washed with water and dried under vacuum to afford 2,5-difluoro-4-nitrophenylacetic acid (4.37g, 82% theory) as a pale yellow solid.

20

To a solution of the carboxylic acid (4.26 g, 19.6 mM) in absolute EtOH (40 mL) was added conc. H₂SO₄ (4 mL) and the mixture was heated at reflux for 5 h. The mixture was then diluted with water and

extracted with EtOAc affording ester **142** as an oil (4.75 g, 98.5% theory) which crystallized on standing

step 2

- 5 To a ice-cooled solution of 3-chlorophenol (2.8 mL, 26.5 mM) in dry THF (25mL) under N₂ was added a solution of potassium *tert*-butoxide (1M in THF, 5.2 mL, 5.2 mM). After 30 m stirring at 0° C a solution of **142** (1.23g, 5.0 mM) in THF (5mL) was added dropwise over 3m. The mixture was then heated at reflux for 1h, after which the reaction was complete. The reaction mixture was partitioned between EtOAc and aqueous NH₄Cl and the EtOAc phase was dried and the crude product was purified
- 10 by flash chromatography on silica gel (5% EtOAc-35% hexane-60% toluene) to afford **143a** (1.28g, 72% theory) as a yellow oil.

step 3

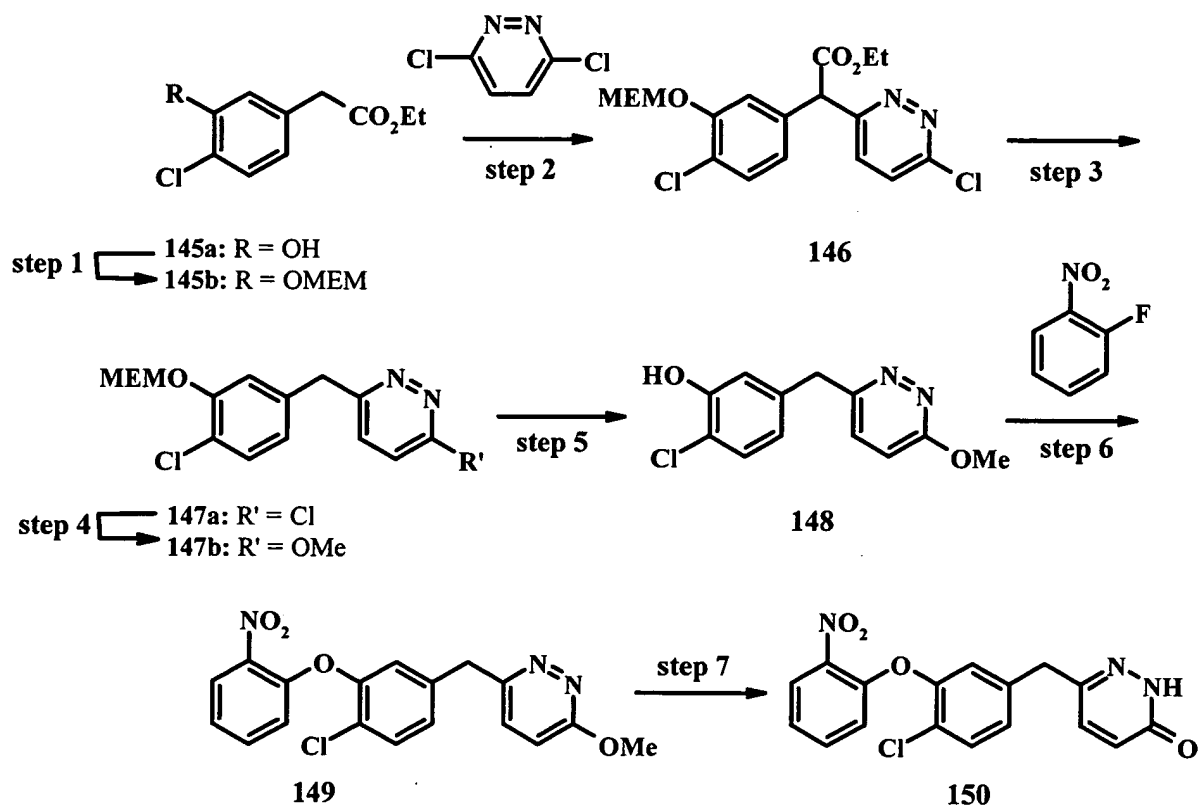
- To a solution of **143a** (945 mg, 2.67 mM) in absolute EtOH (50 mL) was added ammonium chloride
- 15 (850 mg, 16 mM) iron powder (900 mg, 16 mM) and water (20 mL), and the mixture was stirred vigorously while heating at 80° C for 8h. The mixture was cooled and filtered through CELITE[®], the filter cake was washed with EtOH, and most of the solvent was then removed by evaporation. The residue was diluted with EtOAc, washed with water and dried (Na₂SO₄). The organic solution was evaporated to give a crude product that was purified by flash chromatography on silica gel (8% EtOAc-
- 20 25% methylene chloride-67% hexane) to afford **143b** (776 mg, 90% theory) as a violet colored oil.

step 4

- To a solution of **143b** (776 mg, 2.41 mmol) in dry MeCN (18 mL) was added CuCl₂ (390 mg, 2.89 mmol) followed by *tert*-butyl nitrite (0.35 mL, 2.65 mM). The mixture was stirred under an argon
- 25 atmosphere for 2.5 hrs, and was then stored at -6° C for 16 h. The mixture was diluted with EtOAc, washed sequentially with dilute aqueous HCl twice with water and dried (Na₂SO₄). The solvent was evaporated and the crude product purified by flash chromatography on silica gel (EtOAc:hexane 7:93) to afford **143c** (530 mg, 64% theory) as a colorless oil.
- 30 Using the procedure in steps 5 and 6 of example 2 there was prepared pyridazinone **144a** as an amorphous foam: ms [M+H]⁺ = 365: anal; calc'd for C₁₇H₁₁C₁₂FN₂O₂: C, 55.91; H, 3.04; N, 7.67; found: C, 55.67; H, 3.06; N, 7.63. Using the procedure in example 16 there was prepared pyridazinone **144b** as an amorphous foam: ms [M+H]⁺ = 379: anal; calc'd for C₁₇H₁₁C₁₂FN₂O₂:

EXAMPLE 41

6-[4-Chloro-3-(2-nitro-phenoxy)-benzyl]-2H-pyridazin-3-one



5

step 1

To a solution of ethyl 4-chloro-3-hydroxyphenylacetate (**145a**, 12.66 g; 59.0 mmol), *N,N*-diisopropylethylamine (12.3 ml; 70.8 mmol) and CH_2Cl_2 (170 ml) at 0 °C was added MEMCl (7.4 ml; 64.9 mmol) dropwise. The reaction mixture was allowed to warm to room temperature and stirred overnight. The reaction was poured into H_2O and extracted with CH_2Cl_2 . The combined organic extracts were dried (Na_2SO_4), filtered and evaporated to yield of **145b** (13.5 g; 76%).

10

step 2

A solution of **145b** (10.00 g; 33.0 mmol) and 3,6-dichloropyridazine (10.33 g; 69.4 mmol) in DMF (100 mL) was degassed and the flask alternately purged and refilled with N_2 . NaH (3.3 g, 82.6 mmol; 60% in mineral oil) was added portionwise at 0 °C and the reaction was allowed to warm to room temperature and stirred for 1.5 h. The reaction was poured into aqueous 10% $NaHSO_4$ and extracted with EtOAc. The combined organic extracts were washed six times with H_2O and brine, dried (Na_2SO_4), filtered and evaporated. The crude product was purified by flash chromatography on silica gel (hexane/EtOAc 85:15 to 75:25) to afford **146** (9.3 g, 68%).

20

step 3

A solution of **146** (9.27 g; 22.3 mmol) in THF (70 mL) and H₂O (18 mL) was degassed and the flask alternately purged and refilled with nitrogen. LiOH (1.07 g; 44.6 mmol) was added. The reddish-orange reaction mixture was stirred for 3 h and then acidified with HCl (10%) to pH 2. The aqueous solution
5 was thrice extracted with CH₂Cl₂, and the combined extracts washed with H₂O and brine, dried, filtered and evaporated to yield **147a** (7.6 g, 99%).

step 4

To a solution of **147a** (7.58 g; 22.1 mmol) and MeOH (180 mL), was added NaOMe (5.97 g; 110.5
10 mmol) and the solution heated to reflux for 3 h under a N₂ atmosphere. The reaction mixture was concentrated and CH₂Cl₂ was added. The mixture was washed with H₂O (3x) and brine, dried (Na₂SO₄), filtered and concentrated. The crude product was then purified by flash chromatography on silica gel (hexane:EtOAc 90:10 to 60:40) to yield **147b** (7.1 g, 95%).

15 step 5

A mixture of **147b** (7.10 g; 21 mmol), 15.9 mL of 10% HCl, and MeOH (80 mL) was heated at 50 °C overnight. The reaction was cooled to room temperature and saturated NaHCO₃ was added. The mixture was extracted with EtOAc, washed with brine, dried (MgSO₄) and concentrated to yield **148**
(5.20 g, 99%).

20

step 6

To a solution of **148** (0.20 g; 0.8 mmol), K₂CO₃ (0.33 g; 2.4 mmol) and DMF (2 mL) was added 2-fluoronitrobenzene (0.11 mL; 1.04 mmol) and the reaction mixture was heated to 40° C overnight. The reaction was cooled to RT and 10% NaHSO₄ was added. The aqueous phase was extracted with EtOAc.
25 The combined organic extracts were washed with H₂O and brine, dried (Na₂SO₄), filtered and evaporated. The crude product was purified by flash chromatography on silica gel (hexane/EtOAc 90:10 to 60:40) to yield **149** (0.285 g, 96%).

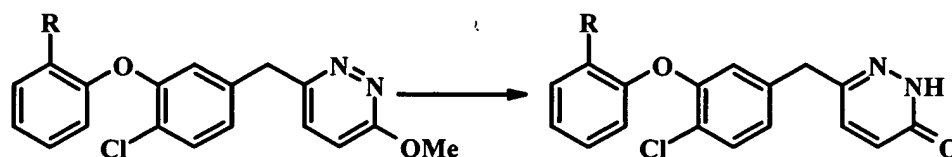
step 7

30 A mixture of HBr (1 mL) and HOAc (1 mL) was added to 1g of **149** (0.140 g; 0.376 mmol). The reaction mixture was heated for 3 h at 100 °C then cooled to rt and extracted with EtOAc. The organic layer was washed with H₂O and brine, dried (Na₂SO₄), filtered and evaporated to yield **150** (0.125 g, 93 %).

EXAMPLE 42

N-{2-[2-Chloro-5-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-phenyl}-methanesulfonamide (**152a**) and N-{2-[2-Chloro-5-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-phenyl}-acetamide (**152b**)

5



step 1 \rightarrow **151a**: R = NO₂
step 2 \rightarrow **151b**: R = NH₂
step 3 \rightarrow **151c**: R = NHSO₂Me
 \rightarrow **151d**: R = NHAc

step 4 \rightarrow **152a**: R = NHSO₂Me
 \rightarrow **152b**: R = NHAc

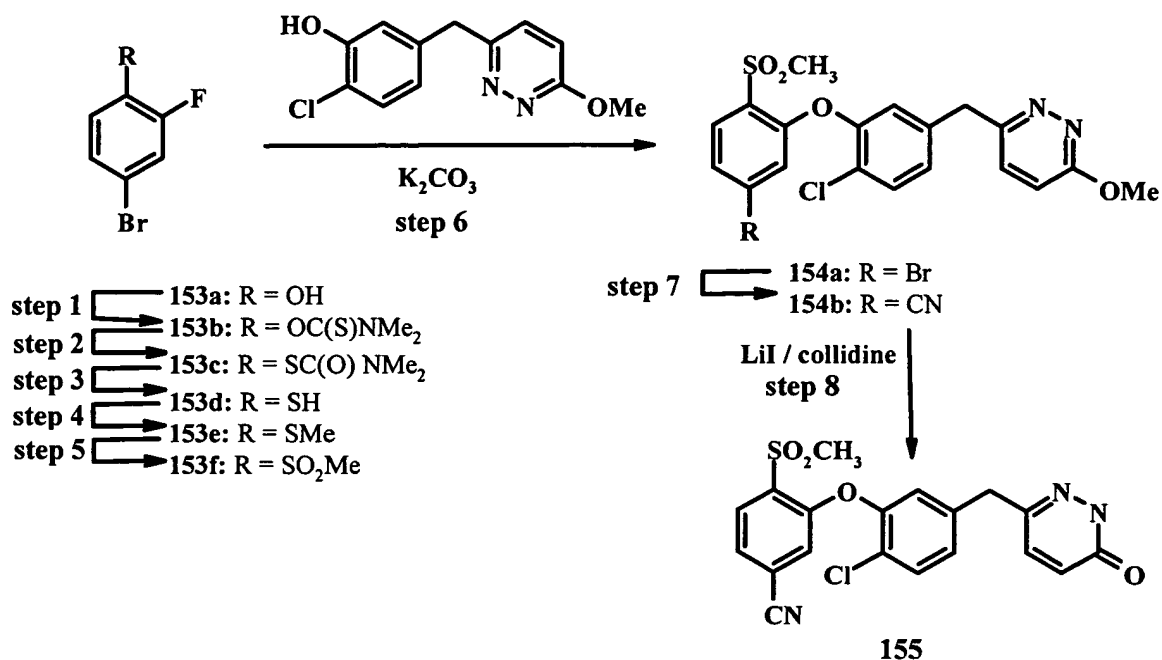
Reduction of the nitro group of **151a** was carried out as described in step 3 of example 43 to afford **151b**.
Sulfonylation and acetylation of the aryl amine to afford **151c** and **151d** respectively was carried out by
10 treating **151b** with methanesulfonyl chloride/TEA or acetyl/TEA using standard protocols.

To a solution of **151c** (0.115 g; 0.274 mmol) in collidine (2 mL) was added LiI (0.110 g; 0.822 mmol).
The mixture was heated at 180° C for 1h and then cooled to RT. The reaction mixture was diluted with
10% HCl and extracted with EtOAc. The organic extract was washed with H₂O and brine, dried
15 (Na₂SO₄), filtered and evaporated to afford **152a** (0.062g, 56 %).

The acetamide **151d** was demethylated under similar conditions to afford **152b**.

EXAMPLE 43

3-[2-Chloro-5-(6-oxo-1,6-dihydro-pyridazin-3-ylmethyl)-phenoxy]-4-methanesulfonyl-benzonitrile
(155)



5

step 1

To a solution of 4-bromo-2-fluorophenol (5.73 mL; 52.4 mmol) in DMF (100 mL) was added DABCO (11.75 g; 104.7 mmol) and N,N-dimethylthiocarbamoyl chloride (9.71 g; 78.5 mmol). The reaction mixture was heated at 75 °C for 1 h. The reaction was cooled to RT and H₂O was added. The suspension was filtered and the solid washed with H₂O, dried and used in step 2.

10

step 2

A solution of **153b** dissolved in sulfolane was heated at 220 °C for 14 h under an N₂ atmosphere. The solvent was removed under high vacuum and the product was purified by flash chromatography on silica gel to yield **153c** (4.90 g, 34% for steps 1 & 2).

15

step 3

To a solution of **153c** (4.88 g; 17.5 mmol) in MeOH (50 mL), was added NaOH (1.40 g, 35.1 mmol in 14 mL of H₂O). The reaction mixture was heated to reflux under N₂ atmosphere for 5 h. The reaction was cooled to RT and aqueous 10% NaHSO₄ was added. The aqueous solution was extracted with EtOAc and the combined organic extracts were washed with H₂O and brine, dried (Na₂SO₄), filtered and evaporated to yield **153d** (3.34 g, 92%).

20

step 4

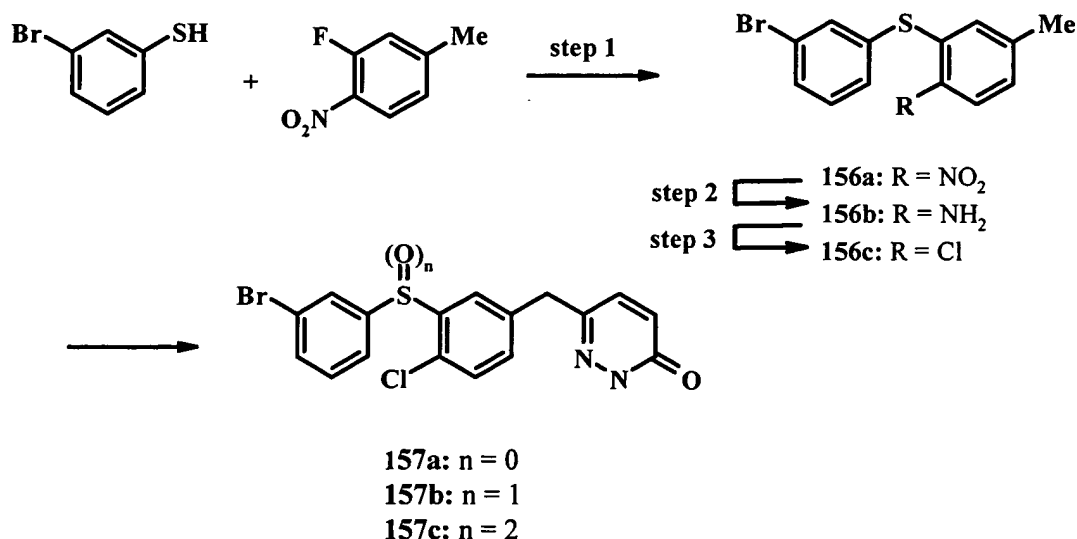
To a solution of **153d** (0.90 g; 4.3 mmol), K_2CO_3 (1.50 g; 10.9 mmol), in NMP (10 mL) was added MeI (0.54 ml; 8.7 mmol). The reaction was heated at 85° C in an oil bath for 1 h and then allowed to cool to RT. A saturated solution of $NaHCO_3$ was added and the mixture was extracted with EtOAc, washed
5 with saturated $NaHCO_3$ and brine, dried (Na_2SO_4) and concentrated to yield **153e** (0.95 g, 99%).

The thiomethyl ether was oxidized to the corresponding sulfone **152f** with MCPBA (procedure). Substitution of the bromo radical with a cyano radical was accomplished by palladium-mediated displacement by $Zn(CN)_2$ as described in step 5 of Example 20. Displacement of the bromide with
10 $Zn(CN)_2$ was achieved as described in example 44. demethylated with lithium chloride and collidine as describe in step 2 of example 30.

A similar series of reactions start from 2-fluorophenol was used to prepare 6-[4-chloro-3-(2-methanesulfonyl-phenoxy)-benzyl]-2H-pyridazin-3-one
15

EXAMPLE 44

6-[3-(3-Bromo-benzenesulfinyl)-4-chloro-benzyl]-2H-pyridazin-3-one (**157a**); 6-[3-(3-Bromo-benzenesulfinyl)-4-chloro-benzyl]-2H-pyridazin-3-one (**157b**); 6-[3-(3-Bromo-benzenesulfonyl)-4-chloro-benzyl]-2H-pyridazin-3-one (**157c**)
20



step 1

To a suspension of *m*-fluoro-*p*-nitrotoluene (4.1 g, 26.4 mmol) and K₂CO₃ (11 g, 79.5 mmol) in dry DMF (75 mL) under a nitrogen atmosphere was added *p*-bromothiophenol (5 g, 26.4 mmol) in single
5 portion. The suspension was heated to reflux at 160 °C for 4 h then allowed to cool to RT. The mixture was then filtered through silica gel. The filtrate was concentrated and purified by flash chromatography on silica gel (ethyl acetate/ hexanes 10% to 50%) to yield **156a** (6.8 g, 79%) an orange oil.

10 step 2

A round bottom flask was charged with **156a** (6.8 g, 20.97 mmol) and SnCl₂·2H₂O (18.93 g, 83.9 mmol). The mixture was suspended in EtOH (50 mL) and refluxed at 70° C for 4.5 h. The suspension was allowed to cool to RT and the pH was adjusted to pH 8 with 2 M NaOH. To this was added ethyl acetate (100 mL) and the mixture was filtered through CELITE®. The filtrate was washed with water
15 (100 mL), brine (50 mL), and dried (MgSO₄). The solvents were evaporated *in vacuo* to afford **156b** (5.6 g, 90%) as an orange oil and that used without any further purification.

step 3

A dry round bottom flask was charged with CuCl₂ (1.65 g, 12.2 mmol), purged with nitrogen and MeCN
20 (25 mL) was added. The suspension was heated to 65° C. To the suspension was added *t*-butyl nitrite (1.82 mL, 15.3 mmol) and the solution was stirred for 5 m and a solution of the **156b** (3 g, 10.2 mmol) in MeCN (15 mL) was added dropwise over 15 m. The reaction was stirred for another 75 m before cooling in an ice bath. The reaction mixture was diluted with 5% HCl (10 mL) and stirred for a few minutes before extracting with ethyl acetate (3 x 50 mL). The combined organic extracts were
25 combined, sequentially washed with water (100 mL) and brine (50 mL) and dried (MgSO₄). The crude was purified by flash chromatography on silica gel (hexanes) to afford

Preparation of 6-[3-(3-Bromo-benzenesulfinyl)-4-chloro-benzyl]-2H-pyridazin-3-one

A solution of **157a** (55 mg, 0.135 mmol), MCPBA (32 mg, 0.18 mmol) and dry dichloromethane was
30 stirred under nitrogen for 24 h at RT. The reaction was diluted with EtOAc and washed sequentially with saturated sodium bisulfite, 1M NaOH (aq, 10 mL), water (10 mL), brine (10 mL), and dried over anhydrous MgSO₄. The crude material was purified by prep TLC (SiO₂, ethyl acetate) to yield **157b** (29 mg, 57%).

Preparation of 6-[3-(3-Bromo-benzenesulfonyl)-4-chloro-benzyl]-2H-pyridazin-3-one (157c)

A solution of **157a** (55 mg, 0.135 mmol), MCPBA (64 mg, 0.37 mmol) and dry dichloromethane was stirred under nitrogen for 24 h at RT. The reaction was diluted with EtOAc and washed sequentially with saturated sodium bisulfite, 1M NaOH (10 mL), water (10 mL), brine (10 mL), and dried over
5 anhydrous MgSO₄. The crude material was purified by prep TLC (SiO₂, ethyl acetate) to yield **157c** (34 mg, 64%).

EXAMPLE 45

HIV Reverse Transcriptase Assay: Inhibitor IC₅₀ determination

10 HIV-1 RT assay was carried out in 96-well Millipore MultiScreen MADVNOB50 plates using purified recombinant enzyme and a poly(rA)/oligo(dT)₁₆ template-primer in a total volume of 50 µL. The assay constituents were 50 mM Tris/HCl, 50 mM NaCl, 1 mM EDTA, 6 mM MgCl₂, 5 µM dTTP, 0.15 µCi [³H] dTTP, 5 µg/ml poly (rA) pre annealed to 2.5 µg/ml oligo (dT)₁₆ and a range of inhibitor
15 concentrations in a final concentration of 10% DMSO. Reactions were initiated by adding 4 nM HIV-1 RT and after incubation at 37⁰C for 30 min, they were stopped by the addition of 50 µl ice cold 20%TCA and allowed to precipitate at 4⁰C for 30 min. The precipitates were collected by applying vacuum to the plate and sequentially washing with 3 x 200 µl of 10% TCA and 2 x 200 µl 70% ethanol. Finally, the plates were dried and radioactivity counted in a Packard TopCounter after the addition of
20 25 µl scintillation fluid per well. IC_{50's} were calculated by plotting % inhibition versus log₁₀ inhibitor concentrations.

Table 3	
Compound #	RT inhibition IC ₅₀ (μM)
I-136	0.008
I-103	0.00935
I-108	0.01115
I-135	0.0125
I-142	0.015
I-98	0.0194
I-102	0.02063
I-109	0.0216
I-123	0.0231
I-100	0.02415
I-77	0.02425
I-76	0.059
I-224	0.018
I-237	0.0128

EXAMPLE 46

PHARMACEUTICAL COMPOSITIONS

5

Composition for Oral Administration

Ingredient	% wt./wt.
Active ingredient	20.0%
Lactose	79.5%
Magnesium stearate	0.5%

The ingredients are mixed and dispensed into capsules containing about 100 mg each; one capsule would approximate a total daily dosage.

10

Composition for Oral Administration

Ingredient	% wt./wt.
Active ingredient	20.0%
Magnesium stearate	0.5%
Crosscarmellose sodium	2.0%
Lactose	76.5%
PVP (polyvinylpyrrolidone)	1.0%

The ingredients are combined and granulated using a solvent such as methanol. The formulation is then dried and formed into tablets (containing about 20 mg of active compound) with an appropriate tablet machine.

Composition for Oral Administration

Ingredient	Amount
Active compound	1.0 g
Fumaric acid	0.5 g
Sodium chloride	2.0 g
Methyl paraben	0.15 g
Propyl paraben	0.05 g
Granulated sugar	25.5 g
Sorbitol (70% solution)	12.85 g
Veegum K (Vanderbilt Co.)	1.0 g
Flavoring	0.035 ml
Colorings	0.5 mg
Distilled water	q.s. to 100 ml

5

The ingredients are mixed to form a suspension for oral administration.

Parenteral Formulation (IV)

Ingredient	% wt./wt.
Active ingredient	0.25 g
Sodium Chloride	qs to make isotonic
Water for injection to	100 ml

- 10 The active ingredient is dissolved in a portion of the water for injection. A sufficient quantity of sodium chloride is then added with stirring to make the solution isotonic. The solution is made up to weight with the remainder of the water for injection, filtered through a 0.2 micron membrane filter and packaged under sterile conditions.

15

Suppository Formulation

Ingredient	% wt./wt.
Active ingredient	1.0%
Polyethylene glycol 1000	74.5%
Polyethylene glycol 4000	24.5%

The ingredients are melted together and mixed on a steam bath, and poured into molds containing 2.5 g total weight.

Topical Formulation

Ingredients	grams
Active compound	0.2-2
Span 60	2
Tween 60	2
Mineral oil	5
Petrolatum	10
Methyl paraben	0.15
Propyl paraben	0.05
BHA (butylated hydroxy anisole)	0.01
Water	q.s. 100

All of the ingredients, except water, are combined and heated to about 60°C with stirring. A sufficient quantity of water at about 60°C is then added with vigorous stirring to emulsify the ingredients, and
5 water then added q.s. about 100 g.

Nasal Spray Formulations

Several aqueous suspensions containing from about 0.025-0.5 percent active compound are prepared as nasal spray formulations. The formulations optionally contain inactive ingredients such as, for example,
10 microcrystalline cellulose, sodium carboxymethylcellulose, dextrose, and the like. Hydrochloric acid may be added to adjust pH. The nasal spray formulations may be delivered via a nasal spray metered pump typically delivering about 50-100 microliters of formulation per actuation. A typical dosing schedule is 2-4 sprays every 4-12 hours.

15 The features disclosed in the foregoing description, or the following claims, or the accompanying drawings, expressed in their specific forms or in terms of a means for performing the disclosed function, or a method or process for attaining the disclosed result, as appropriate, may, separately, or in any combination of such features, be utilized for realizing the invention in diverse forms thereof.

20 The foregoing invention has been described in some detail by way of illustration and example, for purposes of clarity and understanding. It will be obvious to one of skill in the art that changes and modifications may be practiced within the scope of the appended claims. Therefore, it is to be understood that the above description is intended to be illustrative and not restrictive. The scope of the
25 invention should, therefore, be determined not with reference to the above description, but should instead be determined with reference to the following appended claims, along with the full scope of equivalents to which such claims are entitled.

All patents, patent applications and publications cited in this application are hereby incorporated by reference in their entirety for all purposes to the same extent as if each individual patent, patent application or publication were so individually denoted.